

# Peter Stange

## List of Publications by Year in descending order

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28  
papers

816  
citations

516710

16  
h-index

477307

29  
g-index

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31  
docs citations

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times ranked

696  
citing authors

| #  | ARTICLE  | IF   | CITATIONS |
|----|--|------|-----------|
| 1  | Structure, hydrogen bond dynamics and phase transition in a model ionic liquid electrolyte. <i>Physical Chemistry Chemical Physics</i> , 2022, 24, 6064-6071.  | 2.8  | 8         |
| 2  | Balance Between Contact and Solvent-Separated Ion Pairs in Mixtures of the Protic Ionic Liquid [Et <sub>3</sub> NH][MeSO <sub>3</sub> ] with Water Controlled by Water Content and Temperature. <i>Journal of Physical Chemistry B</i> , 2021, 125, 4476-4488.   | 2.6  | 9         |
| 3  | Three in One: The Versatility of Hydrogen Bonding Interaction in Halide Salts with Hydroxy-Functionalized Pyridinium Cations. <i>ChemPhysChem</i> , 2021, 22, 1850-1856.   | 2.1  | 5         |
| 4  | Dissecting intermolecular interactions in the condensed phase of ibuprofen and related compounds: the specific role and quantification of hydrogen bonding and dispersion forces. <i>Physical Chemistry Chemical Physics</i> , 2020, 22, 4896-4904.              | 2.8  | 13        |
| 5  | Freezing the Motion in Hydroxy-Functionalized Ionic Liquids—Temperature Dependent NMR Deuteron Quadrupole Coupling Constants for Two Types of Hydrogen Bonds Far below the Glass Transition. <i>Journal of Physical Chemistry Letters</i> , 2020, 11, 6000-6006. | 4.6  | 10        |
| 6  | Counting cations involved in cationic clusters of hydroxy-functionalized ionic liquids by means of infrared and solid-state NMR spectroscopy. <i>Physical Chemistry Chemical Physics</i> , 2020, 22, 6861-6867.  | 2.8  | 17        |
| 7  | Controlling “like-likes”-like-charge attraction in hydroxy-functionalized ionic liquids by polarizability of the cations, interaction strength of the anions and varying alkyl chain length. <i>Physical Chemistry Chemical Physics</i> , 2020, 22, 2763-2774.   | 2.8  | 29        |
| 8  | Hydrogen Bonding Between Ions of Like Charge in Ionic Liquids Characterized by NMR Deuteron Quadrupole Coupling Constants—Comparison with Salt Bridges and Molecular Systems. <i>Angewandte Chemie</i> , 2019, 131, 18027-18035.                                 | 2.0  | 7         |
| 9  | Hydrogen Bonding Between Ions of Like Charge in Ionic Liquids Characterized by NMR Deuteron Quadrupole Coupling Constants—Comparison with Salt Bridges and Molecular Systems. <i>Angewandte Chemie - International Edition</i> , 2019, 58, 17863-17871.          | 13.8 | 41        |
| 10 | When hydrogen bonding overcomes Coulomb repulsion: from kinetic to thermodynamic stability of cationic dimers. <i>Physical Chemistry Chemical Physics</i> , 2019, 21, 8215-8220.   | 2.8  | 29        |
| 11 | Zerlegung der Verdampfungsenthalpien ionischer Flüssigkeiten durch rein experimentelle Methoden: Coulomb-Wechselwirkung, Wasserstoffbrücken und Dispersionskräfte. <i>Angewandte Chemie</i> , 2019, 131, 8685.   | 2.0  | 0         |
| 12 | The Double-Faced Nature of Hydrogen Bonding in Hydroxy-Functionalized Ionic Liquids Shown by Neutron Diffraction and Molecular Dynamics Simulations. <i>Angewandte Chemie - International Edition</i> , 2019, 58, 12887-12892.                                   | 13.8 | 40        |
| 13 | Die zweigesichtige Natur der Wasserstoffbrückenbindung in hydroxyfunktionalisierten ionischen Flüssigkeiten, offenbart durch Neutronendiffraktometrie und Molekulardynamik-Simulation. <i>Angewandte Chemie</i> , 2019, 131, 13019-13024.                        | 2.0  | 5         |
| 14 | Dissecting the Vaporization Enthalpies of Ionic Liquids by Exclusively Experimental Methods: Coulomb Interaction, Hydrogen Bonding, and Dispersion Forces. <i>Angewandte Chemie - International Edition</i> , 2019, 58, 8589-8592.                               | 13.8 | 19        |
| 15 | Zerlegung der Verdampfungsenthalpien ionischer Flüssigkeiten durch rein experimentelle Methoden: Coulomb-Wechselwirkung, Wasserstoffbrücken und Dispersionskräfte. <i>Angewandte Chemie</i> , 2019, 131, 8679-8683.  | 2.0  | 1         |
| 16 | Simultaneous determination of deuteron quadrupole coupling constants and rotational correlation times: the model case of hydrogen bonded ionic liquids. <i>Physical Chemistry Chemical Physics</i> , 2019, 21, 25597-25605.                                      | 2.8  | 8         |
| 17 | Like-Likes-Like: Cooperative Hydrogen Bonding Overcomes Coulomb Repulsion in Cationic Clusters with Net Charges up to Q = +6 e. <i>ChemPhysChem</i> , 2018, 19, 1691-1695.   | 2.1  | 29        |
| 18 | Dynamical heterogeneities in ionic liquids as revealed from deuteron NMR. <i>Chemical Communications</i> , 2018, 54, 3098-3101.  | 4.1  | 21        |

| #  | ARTICLE  | IF   | CITATIONS |
|----|--|------|-----------|
| 19 | Charakterisierung von Wasserstoffbrücken zwischen Ionen in protischen ionischen Flüssigkeiten mittels NMR-Deuteron-Quadrupol-Kopplungskonstanten – Unterschiede zu H-Brücken in Amiden, Peptiden und Proteinen. <i>Angewandte Chemie</i> , 2017, 129, 14500-14505. | 2.0  | 5         |
| 20 | Characterization of Doubly Ionic Hydrogen Bonds in Protic Ionic Liquids by NMR Deuteron Quadrupole Coupling Constants: Differences to H-bonds in Amides, Peptides, and Proteins. <i>Angewandte Chemie - International Edition</i> , 2017, 56, 14310-14314.         | 13.8 | 35        |
| 21 | Spectroscopic Evidence for Clusters of Like-Charged Ions in Ionic Liquids Stabilized by Cooperative Hydrogen Bonding. <i>ChemPhysChem</i> , 2016, 17, 458-462.   | 2.1  | 115       |
| 22 | Dispersion and Hydrogen Bonding Rule: Why the Vaporization Enthalpies of Aprotic Ionic Liquids Are Significantly Larger than those of Protic Ionic liquids. <i>Angewandte Chemie - International Edition</i> , 2016, 55, 11682-11686.                              | 13.8 | 50        |
| 23 | Dispersion und Wasserstoffbrücken bestimmend – Warum die Verdampfungsenthalpien von aprotischen Ionenflüssigkeiten größer sind als die von protischen ionischen Flüssigkeiten sind. <i>Angewandte Chemie</i> , 2016, 128, 11856-11860.                             | 2.0  | 10        |
| 24 | Steuerung der subtilen Energiebalance in protischen ionischen Flüssigkeiten: Dispersionskräfte im Wettstreit mit Wasserstoffbrücken. <i>Angewandte Chemie</i> , 2015, 127, 2834-2837.  | 2.0  | 13        |
| 25 | Controlling the Subtle Energy Balance in Protic Ionic Liquids: Dispersion Forces Compete with Hydrogen Bonds. <i>Angewandte Chemie - International Edition</i> , 2015, 54, 2792-2795.  | 13.8 | 78        |
| 26 | Ion Pairing in Protic Ionic Liquids Probed by Far-Infrared Spectroscopy: Effects of Solvent Polarity and Temperature. <i>ChemPhysChem</i> , 2014, 15, 2604-2609.   | 2.1  | 40        |
| 27 | Equilibrium of Contact and Solvent-Separated Ion Pairs in Mixtures of Protic Ionic Liquids and Molecular Solvents Controlled by Polarity. <i>Angewandte Chemie - International Edition</i> , 2013, 52, 12439-12442.  | 13.8 | 59        |
| 28 | Ion Speciation of Protic Ionic Liquids in Water: Transition from Contact to Solvent-Separated Ion Pairs. <i>Angewandte Chemie - International Edition</i> , 2013, 52, 2990-2994.   | 13.8 | 89        |