

# Abolfazl Mohammadi

## List of Publications by Year in descending order

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Version: 2024-02-01

26  
papers

817  
citations

623734

14  
h-index

610901

24  
g-index

26  
all docs

26  
docs citations

26  
times ranked

437  
citing authors

#	ARTICLE	IF	CITATIONS
1	Effects of Graphene Oxide Nanosheets and Al <sub>2</sub> O <sub>3</sub> Nanoparticles on CO <sub>2</sub> Uptake in Semi-clathrate Hydrates. <i>Chemical Engineering and Technology</i> , 2021, 44, 48-57.	1.5	27
2	Influence of Tetra-n-Butylammonium Chloride and Polysorbate 80 on the Kinetics of Methane Hydrate Formation. <i>Journal of the Japan Petroleum Institute</i> , 2021, 64, 22-28.	0.6	2
3	Rate of Methane Hydrate Formation in Presence of Tetra-n-Butylammonium Chloride and Tween 80: Application in Air Conditioning Systems. <i>Theoretical Foundations of Chemical Engineering</i> , 2021, 55, 451-456.	0.7	3
4	Predicting semi-clathrate hydrates dissociation pressure using a rigorous machine learning approach. <i>Journal of Dispersion Science and Technology</i> , 2020, 41, 863-872.	2.4	4
5	Kinetics of CO <sub>2</sub> hydrate formation in coffee aqueous solution: Application in coffee concentration. <i>Journal of Dispersion Science and Technology</i> , 2020, 41, 895-901.	2.4	12
6	The effects of graphene oxide nanosheets and Al <sub>2</sub> O <sub>3</sub> nanoparticles on the kinetics of methane hydrate formation at moderate conditions. <i>Journal of Molecular Liquids</i> , 2020, 316, 113872.	4.9	34
7	Investigation of the Effect of NaCl on the Kinetics of R410a Hydrate Formation in the Presence and Absence of Cyclopentane with Potential Application in Hydrate-Based Desalination. <i>Industrial &amp; Engineering Chemistry Research</i> , 2020, 59, 14115-14125.	3.7	28
8	Experimental study and kinetic modeling of R410a hydrate formation in presence of SDS, tween 20, and graphene oxide nanosheets with application in cold storage. <i>Journal of Molecular Liquids</i> , 2020, 304, 112665.	4.9	27
9	The roles TBAF and SDS on the kinetics of methane hydrate formation as a cold storage material. <i>Journal of Molecular Liquids</i> , 2020, 309, 113175.	4.9	23
10	Optimization of determination of CO <sub>2</sub> gas hydrates surface tension in the presence of non-ionic surfactants and TBAC. <i>Eurasian Chemical Communications</i> , 2020, 2, 420-426.	0.9	4
11	Kinetic study of methane hydrate formation in the presence of carbon nanostructures. <i>Petroleum Science</i> , 2019, 16, 657-668.	4.9	32
12	Characterization and Catalytic Reactivity of LaNi <sub>1-x</sub> Mg <sub>x</sub> O <sub>3</sub> Perovskite Oxides in Reforming of Methane with CO <sub>2</sub> and O <sub>2</sub> . <i>International Journal of Chemical Reactor Engineering</i> , 2018, 16, .	1.1	3
13	Combined methane reforming over nano LaNiO <sub>3</sub> catalyst with modified active surface. <i>Research on Chemical Intermediates</i> , 2018, 44, 1755-1773.	2.7	8
14	Semicompletion time of carbon dioxide uptake in the process of gas hydrate formation in presence and absence of SDS and silver nanoparticles. <i>Petroleum Science and Technology</i> , 2017, 35, 37-44.	1.5	5
15	Effect of SDS, silver nanoparticles, and SDS + silver nanoparticles on methane hydrate semicompletion time. <i>Petroleum Science and Technology</i> , 2017, 35, 1542-1548.	1.5	12
16	The effect of a TEG additive on hydrate formation. <i>Petroleum Science and Technology</i> , 2017, 35, 1154-1159.	1.5	0
17	Measurement of the amount and rate of methane dissolution in pure water and aqueous solution of SDS + multi-wall carbon nanotubes + $\beta$ -cyclodextrin. <i>Petroleum Science and Technology</i> , 2017, 35, 1549-1555.	1.5	3
18	Induction time, storage capacity, and rate of methane hydrate formation in the presence of SDS and silver nanoparticles. <i>Chemical Engineering Communications</i> , 2017, 204, 1420-1427.	2.6	37

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19	Kinetic study of carbon dioxide hydrate formation in presence of silver nanoparticles and SDS. Chemical Engineering Journal, 2014, 237, 387-395.	12.7	249
20	Thermodynamic modeling of the dissociation conditions of hydrogen sulfide clathrate hydrate in the presence of aqueous solution of inhibitor (alcohol, salt or ethylene glycol). Chemical Engineering Research and Design, 2014, 92, 2283-2293.	5.6	23
21	Phase equilibria of semiclathrate hydrates for methane+tetra n-butylammonium chloride (TBAC), carbon dioxide+TBAC, and nitrogen+TBAC aqueous solution systems. Fluid Phase Equilibria, 2014, 381, 102-107.	2.5	50
22	The equilibrium conditions, hydrate formation and dissociation rate and storage capacity of ethylene hydrate in presence of 1,4-dioxane. Chemical Engineering Journal, 2013, 217, 379-384.	12.7	45
23	Dissociation Data of Semiclathrate Hydrates for the Systems of Tetra- <i>n</i> -butylammonium Fluoride (TBAF) + Methane + Water, TBAF + Carbon Dioxide + Water, and TBAF + Nitrogen + Water. Journal of Chemical & Engineering Data, 2013, 58, 3545-3550.	1.9	59
24	Effect of synthesized silver nanoparticles in promoting methane hydrate formation at 4.7MPa and 5.7MPa. Chemical Engineering Research and Design, 2013, 91, 1050-1054.	5.6	104
25	Effect of $\beta$ -cyclodextrin on dissolution of methane in water. Chemical Engineering Research and Design, 2011, 89, 421-427.	5.6	20
26	The effect tetra butyl ammonium hydroxide and tween on the kinetics of carbon dioxide hydrate formation. Petroleum Science and Technology, 0, , 1-19.	1.5	3