

# Alexander I Shames

## List of Publications by Year in descending order

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91  
papers

1,833  
citations

218381

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h-index

301761

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91  
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91  
docs citations

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times ranked

2225  
citing authors

#	ARTICLE	IF	CITATIONS
1	Anomalous Formation of Irradiation-Induced Nitrogen-Vacancy Centers in 5 nm-Sized Detonation Nanodiamonds. <i>Journal of Physical Chemistry C</i> , 2022, 126, 5206-5217.	1.5	6
2	All-inorganic ferric wheel based on hexaniobate-anion linkers. <i>Dalton Transactions</i> , 2022, 51, 8600-8604.	1.6	5
3	Influence of inversion level on the optical absorption spectra of Ti-doped transparent MgGa <sub>2</sub> O <sub>4</sub> ceramics. <i>Journal of the American Ceramic Society</i> , 2022, 105, 5944-5955.	1.9	1
4	Triphenyllead Hydroperoxide: A 1D Coordination Peroxo Polymer, Single-Crystal-to-Single-Crystal Disproportionation to a Superoxo/Hydroxo Complex, and Application in Catalysis. <i>Inorganic Chemistry</i> , 2022, 61, 8193-8205.	1.9	5
5	Mitochondria membrane transformations in colon and prostate cancer and their biological implications. <i>Biochimica Et Biophysica Acta - Biomembranes</i> , 2021, 1863, 183471.	1.4	8
6	Room-temperature hyperpolarization of polycrystalline samples with optically polarized triplet electrons: pentacene or nitrogen-vacancy center in diamond?. <i>Magnetic Resonance</i> , 2021, 2, 33-48.	0.8	8
7	PVP-coated Gd-grafted nanodiamonds as a novel and potentially safer contrast agent for in vivo MRI. <i>Magnetic Resonance in Medicine</i> , 2021, 86, 935-942.	1.9	32
8	Comment on "Sub-5 nm Nanodiamonds Fabricated by Plasma Immersion Ion Implantation as Fluorescent Probes". <i>ACS Applied Nano Materials</i> , 2021, 4, 5621-5623.	2.4	0
9	Soluble Complexes of Cobalt Oxide Fragments Bring the Unique CO <sub>2</sub> Photoreduction Activity of a Bulk Material into the Flexible Domain of Molecular Science. <i>Journal of the American Chemical Society</i> , 2021, 143, 20769-20778.	6.6	30
10	Toward production of diamond particles with improved fluorescence uniformity. <i>Physica B: Condensed Matter</i> , 2020, 579, 411868.	1.3	2
11	Near-Infrared Fluorescence from Silicon- and Nickel-Based Color Centers in High-Pressure High-Temperature Diamond Micro- and Nanoparticles. <i>Advanced Optical Materials</i> , 2020, 8, 2001047.	3.6	11
12	Enhanced Optical <sup>13</sup> C Hyperpolarization in Diamond Treated by High-Temperature Rapid Thermal Annealing. <i>Advanced Quantum Technologies</i> , 2020, 3, 2000050.	1.8	8
13	Thiourea-Mediated Halogenation of Alcohols. <i>Journal of Organic Chemistry</i> , 2020, 85, 12901-12911.	1.7	17
14	How to Identify, Attribute, and Quantify Triplet Defects in Ensembles of Small Nanoparticles. <i>Journal of Physical Chemistry Letters</i> , 2020, 11, 7438-7442.	2.1	6
15	Identification of Barium Hydroxo-Hydroperoxostannate Precursor for Low-Temperature Formation of Perovskite Barium Stannate. <i>Inorganic Chemistry</i> , 2020, 59, 18358-18365.	1.9	10
16	High Temperature Treatment of Diamond Particles Toward Enhancement of Their Quantum Properties. <i>Frontiers in Physics</i> , 2020, 8, .	1.0	11
17	Photoredox-Mediated Reaction of <i>gem</i> -Diborylalkenes: Reactivity Toward Diverse 1,1-Bisborylalkanes. <i>Chemistry - A European Journal</i> , 2020, 26, 5360-5364.	1.7	24
18	Stabilization of Ni(I)(1,4,8,11-tetraazacyclotetradecane) <sup>+</sup> in a Sol-Gel Matrix: It's Plausible Use in Catalytic Processes. <i>Israel Journal of Chemistry</i> , 2020, 60, 557-562.	1.0	4

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19	Sources of parasitic features in the visible range of oxide transparent ceramics absorption spectra. <i>Journal of the American Ceramic Society</i> , 2020, 103, 4803-4821.	1.9	8
20	Examining relaxivities in suspensions of nanodiamonds grafted by magnetic entities: comparison of two approaches. <i>Magnetic Resonance Materials in Physics, Biology, and Medicine</i> , 2020, 33, 885-888.	1.1	4
21	Unusual Stabilization of Zinc Peroxide by Manganese Oxide: Mechanistic Understanding by Temperature-Dependent EPR Studies. <i>Journal of Physical Chemistry C</i> , 2019, 123, 20884-20892.	1.5	10
22	Gd(III)-Grafted Detonation Nanodiamonds for MRI Contrast Enhancement. <i>Journal of Physical Chemistry C</i> , 2019, 123, 2627-2631.	1.5	46
23	Structural Characterization of Reconstituted Bioactive-Loaded Nanodomains after Embedding in Films Using Electron Paramagnetic Resonance and Self-Diffusion Nuclear Magnetic Resonance Techniques. <i>Langmuir</i> , 2019, 35, 7879-7886.	1.6	5
24	Monodisperse Five-Nanometer-Sized Detonation Nanodiamonds Enriched in Nitrogen-Vacancy Centers. <i>ACS Nano</i> , 2019, 13, 6461-6468.	7.3	38
25	Fluorescent Diamond Particles: From Fancy Blue to Red: Controlled Production of a Vibrant Color Spectrum of Fluorescent Diamond Particles ( <i>Adv. Funct. Mater.</i> 19/2019). <i>Advanced Functional Materials</i> , 2019, 29, 1970128.	7.8	2
26	Micro-characterization of modified microemulsions loaded with gossypol, pure and extracted from cottonseed. <i>Colloids and Surfaces B: Biointerfaces</i> , 2019, 180, 487-494.	2.5	4
27	Review Article: Synthesis, properties, and applications of fluorescent diamond particles. <i>Journal of Vacuum Science and Technology B: Nanotechnology and Microelectronics</i> , 2019, 37, 030802.	0.6	115
28	Magnetic resonance study of lightly boron-doped diamond. <i>Materials Research Express</i> , 2019, 6, 075612.	0.8	1
29	From Fancy Blue to Red: Controlled Production of a Vibrant Color Spectrum of Fluorescent Diamond Particles. <i>Advanced Functional Materials</i> , 2019, 29, 1808362.	7.8	29
30	A Surface Study of Ultrathin Ceria Nanoparticles Decorated with Transition Metal Ions. <i>Particle and Particle Systems Characterization</i> , 2019, 36, 1800452.	1.2	3
31	Photoxidation of Benzyl Alcohol with Heterogeneous Photocatalysts in the UV Range: The Complex Interplay with the Autoxidative Reaction. <i>ChemCatChem</i> , 2018, 10, 2541-2545.	1.8	16
32	Comment on "Ångstrom-scale probing of paramagnetic center location in nanodiamonds by <sup>3</sup> He NMR at low temperatures" by V. Kuzmin, K. Safiullin, G. Dolgorukov, A. Stanislavovas, E. Alakshin, T. Safin, B. Yavkin, S. Orlinskii, A. Kiiamov, M. Presnyakov, A. Klochkov and M. Tagirov, <i>Phys. Chem. Chem. Phys.</i> , 2018, 20, 1476. <i>Physical Chemistry Chemical Physics</i> , 2018, 20, 27694-27696.	1.3	1
33	Evolution of Triplet Paramagnetic Centers in Diamonds Obtained by Sintering of Detonation Nanodiamonds at High Pressure and Temperature. <i>Physics of the Solid State</i> , 2018, 60, 723-729.	0.2	3
34	Does Progressive Nitrogen Doping Intensify Negatively Charged Nitrogen Vacancy Emission from e-Beam-Irradiated Ib Type High-Pressure-High-Temperature Diamonds?. <i>Journal of Physical Chemistry C</i> , 2017, 121, 5232-5240.	1.5	18
35	Behavior of PPI-G2 Dendrimer in a Microemulsion. <i>Journal of Physical Chemistry B</i> , 2017, 121, 2339-2349.	1.2	5
36	Paramagnetic defects in nanodiamonds. , 2017, , 131-154.		9

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37	Light-induced generation of free radicals by fullerene derivatives: an important degradation pathway in organic photovoltaics?. <i>Journal of Materials Chemistry A</i> , 2017, 5, 8044-8050.	5.2	46
38	Fluence-Dependent Evolution of Paramagnetic Triplet Centers in e-Beam Irradiated Microcrystalline Ib Type HPHT Diamond. <i>Journal of Physical Chemistry C</i> , 2017, 121, 22335-22346.	1.5	22
39	Efficient solar cells are more stable: the impact of polymer molecular weight on performance of organic photovoltaics. <i>Journal of Materials Chemistry A</i> , 2016, 4, 7274-7280.	5.2	66
40	Assessing the outdoor photochemical stability of conjugated polymers by EPR spectroscopy. <i>Journal of Materials Chemistry A</i> , 2016, 4, 13166-13170.	5.2	13
41	Magnetic Resonance Study of Gadolinium-Grafted Nanodiamonds. <i>Journal of Physical Chemistry C</i> , 2016, 120, 19804-19811.	1.5	28
42	Direct Synthesis of Polyaryls by Consecutive Oxidative Cross-Coupling of Phenols with Arenes. <i>Organic Letters</i> , 2016, 18, 4324-4327.	2.4	31
43	XRD, NMR, and EPR study of polycrystalline micro- and nano-diamonds prepared by a shock wave compression method. <i>Physica Status Solidi (A) Applications and Materials Science</i> , 2015, 212, 2400-2409.	0.8	25
44	Comment on "Quantification of C• and O• Surface Carbons in Detonation Nanodiamond by NMR". <i>Journal of Physical Chemistry C</i> , 2015, 119, 21286-21287.	1.5	4
45	Size dependence of $^{13}\text{C}$ nuclear spin-lattice relaxation in micro- and nanodiamonds. <i>Journal of Physics Condensed Matter</i> , 2015, 27, 072203.	0.7	26
46	Magnetic resonance tracking of fluorescent nanodiamond fabrication. <i>Journal Physics D: Applied Physics</i> , 2015, 48, 155302.	1.3	38
47	Native and induced triplet nitrogen-vacancy centers in nano- and micro-diamonds: Half-field electron paramagnetic resonance fingerprint. <i>Applied Physics Letters</i> , 2014, 104, .	1.5	17
48	Influence of chemical treatment on the microstructure of nanographite. <i>Physica Status Solidi (A) Applications and Materials Science</i> , 2014, 211, 2765-2772.	0.8	6
49	On the correlation between the structure of lyotropic carriers and the delivery profiles of two common NSAIDs. <i>Colloids and Surfaces B: Biointerfaces</i> , 2014, 122, 231-240.	2.5	27
50	Parasitic Light Absorption Processes in Transparent Polycrystalline $\text{MgAl}_2\text{O}_4$ and $\text{YAG}$ . <i>Journal of the American Ceramic Society</i> , 2013, 96, 3523-3529.	1.9	20
51	Novel technology for the rapid total mineralization of carbon tetrachloride under ambient conditions. <i>RSC Advances</i> , 2013, 3, 24440.	1.7	6
52	Paramagnetic Impurities in Graphene Oxide. <i>Applied Magnetic Resonance</i> , 2013, 44, 107-116.	0.6	59
53	Examining the binding mechanism of 3,4-dihydro-3-(2-oxo-2-phenylethylidene)-quinoxalin-2(1H)-one and its fragments to $\text{Cu}^{2+}$ . <i>Journal of Coordination Chemistry</i> , 2013, 66, 2351-2366.	0.8	2
54	In situ Generation of Superoxide Anion Radical in Aqueous Medium under Ambient Conditions. <i>ChemPhysChem</i> , 2013, 14, 4158-4164.	1.0	28

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55	Nanometer size effects on magnetic order in $\text{La}_{1-x}\text{Ca}_x\text{MnO}_3$ ( $x=0.5$ and $0.6$ ) manganites, probed by ferromagnetic resonance. <i>Journal of Applied Physics</i> , 2012, 111, 07D701.	1.1	7
56	Light-induced electron paramagnetic resonance evidence of charge transfer in electrospun fibers containing conjugated polymer/fullerene and conjugated polymer/fullerene/carbon nanotube blends. <i>Applied Physics Letters</i> , 2012, 100, 113303.	1.5	6
57	W/O Microemulsions as Dendrimer Nanocarriers: An EPR Study. <i>Journal of Physical Chemistry B</i> , 2012, 116, 12633-12640.	1.2	16
58	The redox chemistry of copper tetraphenylporphyrin revisited. <i>Journal of Porphyrins and Phthalocyanines</i> , 2012, 16, 1124-1131.	0.4	7
59	Spin $S=1$ centers: a universal type of paramagnetic defects in nanodiamonds of dynamic synthesis. <i>Journal of Physics Condensed Matter</i> , 2012, 24, 225302.	0.7	14
60	Chemical disorder influence on magnetic state of optimally-doped $\text{La}_{0.7}\text{Ca}_{0.3}\text{MnO}_3$ . <i>Journal of Applied Physics</i> , 2011, 110, .	1.1	21
61	Factors Affecting DNP NMR in Polycrystalline Diamond Samples. <i>Journal of Physical Chemistry C</i> , 2011, 115, 19041-19048.	1.5	72
62	Diagnostics of plasmon resonance in optical absorption spectra of nanographite aqueous suspensions. <i>Optics and Spectroscopy (English Translation of Optika I Spektroskopiya)</i> , 2011, 111, 220-223.	0.2	3
63	Proton magnetic resonance study of diamond nanoparticles decorated by transition metal ions. <i>Journal Physics D: Applied Physics</i> , 2011, 44, 125303.	1.3	40
64	Thermodynamics of Paramagnetic-Ferromagnetic Phase Transition in $\text{La}_{0.7}\text{Ca}_{0.3}\text{MnO}_3$ Manganite: Griffiths singularity versus Chemical Disorder and Lattice Effects. <i>IEEE Transactions on Magnetics</i> , 2010, 46, 1299-1302.	1.2	14
65	Do food microemulsions and dietary mixed micelles interact?. <i>Colloids and Surfaces B: Biointerfaces</i> , 2010, 77, 22-30.	2.5	27
66	Structure and magnetic properties of detonation nanodiamond chemically modified by copper. <i>Journal of Applied Physics</i> , 2010, 107, .	1.1	45
67	Structure and Bonding in Fluorinated Nanodiamond. <i>Journal of Physical Chemistry C</i> , 2010, 114, 774-782.	1.5	51
68	Magnetic Resonance Study of Detonation Nanodiamonds with Surface Chemically Modified by Transition Metal Ions. <i>Applied Magnetic Resonance</i> , 2009, 36, 317-329.	0.6	37
69	Nanometer size effect on magnetic order in $\text{La}_{0.4}\text{Ca}_{0.6}\text{MnO}_3$ . Predominant influence of doped electron localization. <i>Physical Review B</i> , 2008, 78, .	1.1	41
70	EMR Probing of Magnetic Ordering in $\text{Pr}_{1-x}\text{Sr}_x\text{MnO}_3$ ( $x=0.22, 0.24, 0.26$ ) Manganite Single Crystals. <i>IEEE Transactions on Magnetics</i> , 2008, 44, 2918-2921.	1.2	5
71	Inherent inhomogeneity in the crystals of low-doped lanthanum manganites. <i>Applied Physics Letters</i> , 2008, 92, .	1.5	24
72	Comparative Study of Magnetic Ordering in Bulk and Nano-Grained $\text{La}_{0.4}\text{Ca}_{0.6}\text{MnO}_3$ Manganite. <i>IEEE Transactions on Magnetics</i> , 2008, 44, 2914-2917.	1.2	4

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73	Comparative electron magnetic resonance study of magnetic ordering in $\text{La}_{1-x}\text{Ca}_x\text{MnO}_3$ ( $x=0.1,0.3$ ) bulk and nanometer sized manganite crystals. <i>Journal of Applied Physics</i> , 2008, 103, 07F715.	1.1	8
74	Disorder-induced phase coexistence in bulk doped manganites and its suppression in nanometer-sized crystals: The case of $\text{La}_{0.9}\text{Ca}_{0.1}\text{MnO}_3$ . <i>Physical Review B</i> , 2007, 76, .	1.1	57
75	Magnetic Correlations and Spin Dynamics in Crystalline $\text{La}_{1-x}\text{Ca}_x\text{MnO}_3$ ( $x = 0, 0.1, 0.2$ .) <i>Tj ETQq1 1.0.784314 rgBT / 18</i>	1.2	18
76	Electron Magnetic Resonance, Neutron Diffraction and ac Susceptibility Study of $\text{CaMn}_{1-x}\text{Ru}_x\text{O}_3$ ( $x$ ) <i>Tj ETQq0 0.0 rgBT / 2</i>	0.8	2
77	Nuclear magnetic resonance study of ultrananocrystalline diamonds. <i>European Physical Journal B</i> , 2006, 52, 397-402.	0.6	86
78	Pyrophosphate and ATP as Stabilizing Ligands for High-Valent Nickel Complexes. <i>European Journal of Inorganic Chemistry</i> , 2006, 2006, 523-525.	1.0	2
79	Magnetic phase transition in $\text{YCo}_2\text{B}_2$ studied by magnetic resonance. <i>Journal of Applied Physics</i> , 2005, 98, 074105.	1.1	10
80	Carbon Encapsulated Magnetic Nanoparticles Produced by a Catalytic Disproportionation of Carbon Monoxide. <i>Materials Research Society Symposia Proceedings</i> , 2005, 877, 1.	0.1	0
81	Magnetic resonance study of multiwall boron nitride nanotubes. <i>Physical Review B</i> , 2005, 72, .	1.1	17
82	Silver(II) Complexes of Tetraazamacrocycles: Studies on e.p.r. and Electron Transfer Kinetics with Thiosulfate Ion. <i>Transition Metal Chemistry</i> , 2004, 29, 463-470.	0.7	20
83	Electron magnetic resonance in $\text{La}_{1-x}\text{Ca}_x\text{MnO}_3$ ( $x=0.18,0.20,0.22$ ): $\epsilon$ Crossing through the boundary between ferromagnetic insulating and metallic ground states. <i>Physical Review B</i> , 2003, 68, .	1.1	46
84	Magnetic, transport, and electron magnetic resonance properties of $\text{Pr}_{0.8}\text{Ca}_{0.2}\text{MnO}_3$ single crystals. <i>Physical Review B</i> , 2003, 68, .	1.1	29
85	Effect of chemical treatment on the structure of ultradisperse diamond and onion-like carbon. <i>AIP Conference Proceedings</i> , 2001, , .	0.3	1
86	Observation of magnetic inhomogeneities in crystalline-doped manganites by electron magnetic resonance. <i>Physical Review B</i> , 2001, 64, .	1.1	71
87	Polysilyl radicals: EPR study of the formation and decomposition of star polysilanes. <i>Applied Magnetic Resonance</i> , 2000, 18, 425-434.	0.6	28
88	EPR spin labeling study of conformational transitions of $\beta$ -glycosidase from the hyperthermophilic archaeon <i>Sulfolobus solfataricus</i> expressed in <i>Escherichia coli</i> . <i>Applied Magnetic Resonance</i> , 2000, 18, 515-526.	0.6	3
89	Spin Diffusion in Pure Multiple-Pulse NQR. <i>Zeitschrift Fur Naturforschung - Section A Journal of Physical Sciences</i> , 2000, 55, 54-60.	0.7	5
90	NMR Studies of electrostatic fields around charged monosaccharides and related molecules. <i>Israel Journal of Chemistry</i> , 2000, 40, 263-269.	1.0	4

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91	Changes in the photoelectrical properties and generation of photoinduced defects under light/air exposure of C60 thin films. Journal of Applied Physics, 1998, 84, 3333-3337.	1.1	23