

G P Kopitsa

List of Publications by Year in descending order

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all docs

89
docs citations

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times ranked

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citing authors

#	ARTICLE	IF	CITATIONS
1	Carbon cryogel preparation and characterization. <i>Diamond and Related Materials</i> , 2022, 121, 108727.	1.8	5
2	Morphology and Structure of a Charge of Detonation Nanodiamond Doped with Boron. <i>Glass Physics and Chemistry</i> , 2022, 48, 43-49.	0.2	1
3	Novel biocompatible Cu ²⁺ -containing composite hydrogels based on bacterial cellulose and poly-1-vinyl-1,2,4-triazole. <i>Smart Materials in Medicine</i> , 2022, 3, 382-389.	3.7	7
4	Engineering SiO ₂ –TiO ₂ binary aerogels for sun protection and cosmetic applications. <i>Journal of Supercritical Fluids</i> , 2021, 169, 105099.	1.6	12
5	The Structure and Properties of TiO ₂ Nanopowders for Use in Agricultural Technologies. <i>Biointerface Research in Applied Chemistry</i> , 2021, 11, 12285-12300.	1.0	4
6	Mesostructure of Composite Materials Based on Segmented Poly(Urethane Imide) Containing Ferrite Nanoparticles. <i>Russian Journal of Inorganic Chemistry</i> , 2021, 66, 225-236.	0.3	2
7	Magnetic Neutron Scattering in Reduced Graphene Oxide. <i>JETP Letters</i> , 2021, 113, 384-388.	0.4	2
8	Hydrophobization of organic resorcinol-formaldehyde aerogels by fluoroacylation. <i>Journal of Fluorine Chemistry</i> , 2021, 244, 109742.	0.9	8
9	SiO ₂ –TiO ₂ Binary Aerogels: A Small-Angle Scattering Study. <i>Russian Journal of Inorganic Chemistry</i> , 2021, 66, 874-882.	0.3	7
10	Application of Rock Weathering and Colonization by Biota for the Relative Dating of Moraines from the Arid Part of the Russian Altai Mountains. <i>Geosciences (Switzerland)</i> , 2021, 11, 342.	1.0	2
11	Chemoresistive gas-sensitive ZnO/Pt nanocomposites films applied by microplotter printing with increased sensitivity to benzene and hydrogen. <i>Materials Science and Engineering B: Solid-State Materials for Advanced Technology</i> , 2021, 271, 115233.	1.7	22
12	Aqueous chemical synthesis of iron oxides magnetic nanoparticles of different morphology and mesostructure. <i>Ceramics International</i> , 2021, 47, 28866-28873.	2.3	9
13	The first amorphous and crystalline yttrium lactate: synthesis and structural features. <i>RSC Advances</i> , 2021, 11, 30195-30205.	1.7	3
14	Nanodiamond Batch Enriched with Boron: Properties and Prospects for Use in Agriculture. <i>Biointerface Research in Applied Chemistry</i> , 2021, 12, 6134-6147.	1.0	2
15	Sol-Gel Synthesis and Structure of Nanocomposites Based on Tetraethoxysilane and Boron Compounds. <i>Glass Physics and Chemistry</i> , 2021, 47, S48-S62.	0.2	2
16	Synthesis of Iron Oxide Magnetic Nanoparticles and Their Effect on Growth, Productivity, and Quality of Tomato. <i>Glass Physics and Chemistry</i> , 2021, 47, S67-S74.	0.2	2
17	Photochromic and Photocatalytic Properties of Ultra-Small PVP-Stabilized WO ₃ Nanoparticles. <i>Molecules</i> , 2020, 25, 154.	1.7	12
18	Influence of Stabilizing Ion Content on the Structure, Photoluminescence and Biological Properties of Zr _{1-x} EuxO ₂ –0.5x Nanoparticles. <i>Crystals</i> , 2020, 10, 1038.	1.0	4

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19	Calcifying Bacteria Flexibility in Induction of CaCO ₃ Mineralization. <i>Life</i> , 2020, 10, 317.	1.1	15
20	Is Supercritical So Critical? The Choice of Temperature to Synthesize SiO ₂ Aerogels. <i>Russian Journal of Inorganic Chemistry</i> , 2020, 65, 255-262.	0.3	6
21	1D Cerium Hydrogen Phosphate Aerogels: Noncarbonaceous Ultraflyweight Monolithic Aerogels. <i>ACS Omega</i> , 2020, 5, 17592-17600.	1.6	8
22	Synthesis of Magnetic Nanopowders of Iron Oxide: Magnetite and Maghemite. <i>Russian Journal of Inorganic Chemistry</i> , 2020, 65, 426-430.	0.3	13
23	Aqueous Chemical Co-Precipitation of Iron Oxide Magnetic Nanoparticles for Use in Agricultural Technologies. <i>Letters in Applied NanoBioScience</i> , 2020, 10, 2215-2239.	0.5	4
24	Crystal and Supramolecular Structure of Bacterial Cellulose Hydrolyzed by Cellobiohydrolase from <i>Scytalidium Candidum</i> 3C: A Basis for Development of Biodegradable Wound Dressings. <i>Materials</i> , 2020, 13, 2087.	1.3	8
25	Hydrothermal synthesis of CeO ₂ nanostructures and their electrochemical properties. <i>Nanosystems: Physics, Chemistry, Mathematics</i> , 2020, 11, 355-364.	0.2	5
26	Synthesis and study of multiferroic and ferroelectric core-shell powders for application in electronic devices for medicine and ecology. , 2019, , 183-207.		1
27	Model of Fractal Particles of Hydrated Zirconium Dioxide, Based on Small-Angle Neutron Scattering Data. <i>Journal of Surface Investigation</i> , 2019, 13, 908-913.	0.1	1
28	Investigating the Relationship between the Conditions of Polythiophene Electrosynthesis and the Pseudocapacitive Properties of Polythiophene-Based Electrodes. <i>Glass Physics and Chemistry</i> , 2019, 45, 281-290.	0.2	1
29	Photoluminescent porous aerogel monoliths containing ZnEu-complex: the first example of aerogel modified with a heteronuclear metal complex. <i>Journal of Sol-Gel Science and Technology</i> , 2019, 92, 304-318.	1.1	13
30	Application of BaTiO ₃ /CoFe ₂ O ₄ @SiO ₂ Structure to Control the Electrical Properties of Composites. <i>Glass Physics and Chemistry</i> , 2019, 45, 513-517.	0.2	1
31	A sol-gel synthesis and gas-sensing properties of finely dispersed ZrTiO ₄ . <i>Materials Chemistry and Physics</i> , 2019, 225, 347-357.	2.0	12
32	Temperature-responsive star-shaped poly(2-ethyl-2-oxazoline) and poly(2-isopropyl-2-oxazoline) with central thiacalix[4]arene fragments: structure and properties in solutions. <i>Colloid and Polymer Science</i> , 2019, 297, 285-296.	1.0	5
33	The influence of chemical prehistory on the structure, photoluminescent properties, surface and biological characteristics of Zr _{0.98} Eu _{0.02} O _{1.99} nanophosphors. <i>Nanosystems: Physics, Chemistry, Mathematics</i> , 2019, 10, 164-175.	0.2	5
34	Hybrid mesoporous silica with controlled drug release. <i>Journal of the Serbian Chemical Society</i> , 2019, 84, 1027-1039.	0.4	10
35	Structure and photoluminescent properties of TiO ₂ :Eu ³⁺ nanoparticles synthesized under hydro and solvothermal conditions from different precursors. <i>Nanosystems: Physics, Chemistry, Mathematics</i> , 2019, , 361-373.	0.2	4
36	Luminescence of Eu ³⁺ ions in hybrid polymer-inorganic composites based on poly(methyl) Tj ETQq0,0,0 rgBT /QOverlock 1	1.5	14

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37	First rare-earth phosphate aerogel: sol-gel synthesis of monolithic ceric hydrogen phosphate aerogel. <i>Journal of Sol-Gel Science and Technology</i> , 2018, 85, 574-584.	1.1	13
38	Heat-Treatment-Induced Evolution of the Mesostructure of Finely Divided Y ₃ Al ₅ O ₁₂ Produced by the Sol-Gel Method. <i>Russian Journal of Inorganic Chemistry</i> , 2018, 63, 691-699.	0.3	12
39	Structural Analysis of Aluminum Oxyhydroxide Aerogel by Small Angle X-Ray Scattering. <i>Journal of Surface Investigation</i> , 2018, 12, 296-305.	0.1	9
40	Mesoporous silica obtained with methyltriethoxysilane as co-precursor in alkaline medium. <i>Applied Surface Science</i> , 2017, 424, 275-281.	3.1	40
41	Morphological structure of <i>Gluconacetobacter xylinus</i> cellulose and cellulose-based organic-inorganic composite materials. <i>Journal of Physics: Conference Series</i> , 2017, 848, 012017.	0.3	10
42	Study of the effect of methods for liquid-phase synthesis of nanopowders on the structure and physicochemical properties of ceramics in the CeO ₂ -Y ₂ O ₃ system. <i>Russian Journal of Inorganic Chemistry</i> , 2017, 62, 1275-1285.	0.3	18
43	Comparative analysis of the physicochemical characteristics of SiO ₂ aerogels prepared by drying under subcritical and supercritical conditions. <i>Inorganic Materials</i> , 2017, 53, 1270-1278.	0.2	9
44	Structure and proton conductivity of a hydrated Nafion-115 membrane. <i>Glass Physics and Chemistry</i> , 2016, 42, 637-639.	0.2	5
45	How xerogel carbonization conditions affect the reactivity of highly disperse SiO ₂ -C composites in the sol-gel synthesis of nanocrystalline silicon carbide. <i>Russian Journal of Inorganic Chemistry</i> , 2016, 61, 1347-1360.	0.3	8
46	Mesostructure of yttrium and aluminum basic salts coprecipitated from aqueous solutions under ultrasonic treatment. <i>Journal of Surface Investigation</i> , 2016, 10, 177-186.	0.1	2
47	Effect of biocidal additives on the mesostructure of epoxy-siloxane bioactive coatings. <i>Journal of Surface Investigation</i> , 2016, 10, 113-122.	0.1	3
48	Combined SANS and SAXS study of the action of ultrasound on the structure of amorphous zirconia gels. <i>Ultrasonics Sonochemistry</i> , 2015, 24, 230-237.	3.8	18
49	Microstructure of Zirconia-Based Sol-Gel Glasses Studied by SANS. <i>Acta Physica Polonica A</i> , 2015, 128, 582-585.	0.2	0
50	Structure of zirconium dioxide based porous glasses. <i>Journal of Surface Investigation</i> , 2014, 8, 967-975.	0.1	4
51	Small-angle neutron scattering study of the mesostructure of bioactive coatings for stone materials based on nanodiamond-modified epoxy siloxane sols. <i>Physics of the Solid State</i> , 2014, 56, 105-113.	0.2	9
52	Complete inheritance of fractal properties during first-order phase transition. <i>Journal of Physics and Chemistry of Solids</i> , 2014, 75, 296-299.	1.9	7
53	On the size effect in nanocrystalline cerium dioxide: Is the Tsunekawa model correct?. <i>Journal of Surface Investigation</i> , 2014, 8, 997-1001.	0.1	6
54	pH control of the structure, composition, and catalytic activity of sulfated zirconia. <i>Journal of Solid State Chemistry</i> , 2013, 198, 496-505.	1.4	24

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55	Effect of high intensity ultrasound on the mesostructure of hydrated zirconia. Journal of Physics: Conference Series, 2012, 340, 012057.	0.3	5
56	Effect of synthesis conditions of the micro- and mesostructure of monodisperse Y(OH)CO ₃ powders. Doklady Chemistry, 2012, 446, 207-211.	0.2	2
57	One-stage synthesis of ceria colloid solutions for biomedical use. Doklady Chemistry, 2011, 437, 103-106.	0.2	29
58	Effect of thermal treatment on characteristics nanodiamonds and diamond blend. Acta Crystallographica Section A: Foundations and Advances, 2011, 67, C557-C557.	0.3	0
59	Evolution of composition and fractal structure of hydrous zirconia xerogels during thermal annealing. Russian Journal of Inorganic Chemistry, 2010, 55, 155-161.	0.3	9
60	Oxygen nonstoichiometry of nanocrystalline ceria. Russian Journal of Inorganic Chemistry, 2010, 55, 325-327.	0.3	27
61	Investigation of the evolution of the hydrated zirconia mesostructure at different stages of heat treatment. Physics of the Solid State, 2010, 52, 957-963.	0.2	5
62	Specific features of the mesostructure of amorphous iron(III) hydroxide xerogels synthesized in an ultrasonic field. Physics of the Solid State, 2010, 52, 979-984.	0.2	0
63	Ultrasound-induced changes in mesostructure of amorphous iron (III) hydroxide xerogels: A small-angle neutron scattering study. Physical Review B, 2010, 81, .	1.1	9
64	Hydrothermal microwave synthesis of nanocrystalline cerium dioxide. Doklady Chemistry, 2009, 426, 131-133.	0.2	14
65	Mesostructure of hydrated hafnia xerogels. Doklady Chemistry, 2009, 427, 160-163.	0.2	3
66	Specifics of high-temperature coarsening of ceria nanoparticles. Russian Journal of Inorganic Chemistry, 2009, 54, 1689-1696.	0.3	16
67	Hydrothermal growth of ceria nanoparticles. Russian Journal of Inorganic Chemistry, 2009, 54, 1857-1861.	0.3	18
68	Mesostructure, fractal properties and thermal decomposition of hydrous zirconia and hafnia. Russian Journal of Inorganic Chemistry, 2009, 54, 2091-2106.	0.3	22
69	Fractal structure of ceria nanopowders. Inorganic Materials, 2008, 44, 272-277.	0.2	14
70	Mesostructure of xerogels of hydrated zirconium dioxide. JETP Letters, 2007, 85, 122-126.	0.4	12
71	Study of the heavy-fermion compound CeRu ₂ Si ₂ by the small-angle neutron scattering method. JETP Letters, 2005, 81, 556-560.	0.4	5
72	Determining the structural parameters of fractal and nonfractal objects in multiple small-angle neutron scattering experiments. Journal of Experimental and Theoretical Physics, 2005, 101, 427-436.	0.2	1

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73	Fluctuations of chemical composition of austenite and their consequence on shape memory effect in Fe-Mn-(Si, Cr, Ni, C, N) alloys. <i>Acta Materialia</i> , 2004, 52, 4791-4799.	3.8	25
74	The investigation of Fe-Mn-based alloys with shape memory effect by small-angle scattering of polarized neutrons. <i>Physica B: Condensed Matter</i> , 2003, 335, 134-139.	1.3	8
75	Small-angle polarized neutron scattering in YBa ₂ (Cu _{1-x} Fe _x) ₃ O _{7-y} ceramics at T =290-550 K. <i>Applied Physics A: Materials Science and Processing</i> , 2002, 74, s628-s630.	1.1	2
76	The spin correlations in YBa ₂ (Cu _{1-x} Fe _x) ₃ O _y ceramics at T=15K investigated by the small-angle scattering of polarized neutrons. <i>Physica B: Condensed Matter</i> , 2001, 297, 245-249.	1.3	1
77	The investigation of the spin correlations in YBa ₂ (Cu _{1-x} Fe _x) ₃ O _y ceramics by the small-angle scattering of polarized neutrons. <i>Physica B: Condensed Matter</i> , 2000, 276-278, 788-789.	1.3	2
78	Small-angle polarized neutron scattering in Sm _{1-x} Sr _x MnO ₃ (x<0.5) perovskite. <i>Physica B: Condensed Matter</i> , 2000, 276-278, 795-796.	1.3	16
79	Spatial spin-resonance of polarized neutrons in period-modulated static magnetic fields. <i>Journal of Neutron Research</i> , 1999, 8, 1-15.	0.4	0
80	Spin correlations and magnetonuclear cross-correlation in Sm(Sr)-Mn-O perovskites in the low-temperature phase. <i>JETP Letters</i> , 1999, 69, 353-360.	0.4	16
81	Spin correlations in YBa ₂ (Cu _{1-x} Fe _x) ₃ O _{7+y} ceramic. <i>Physics of the Solid State</i> , 1998, 40, 19-22.	0.2	7
82	Effect of carbon and nitrogen on chemical homogeneity of fcc iron-based alloys. <i>Metallurgical and Materials Transactions A: Physical Metallurgy and Materials Science</i> , 1997, 28, 2195-2199.	1.1	24
83	The observation of the magnetic correlations in YBa ₂ (Cu _{1-x} Fe _x) ₃ O _{7+y} ceramics by small-angle polarized neutron scattering. <i>Physica B: Condensed Matter</i> , 1997, 234-236, 839-840.	1.3	4
84	Spatial spin resonance of polarized neutrons in amplitude-modulated magnetic fields. <i>Nuclear Instruments and Methods in Physics Research, Section A: Accelerators, Spectrometers, Detectors and Associated Equipment</i> , 1997, 389, 441-446.	0.7	6
85	Ferroelectric core/magnetic shell approach to control electric properties of composites. , 0, , .		0