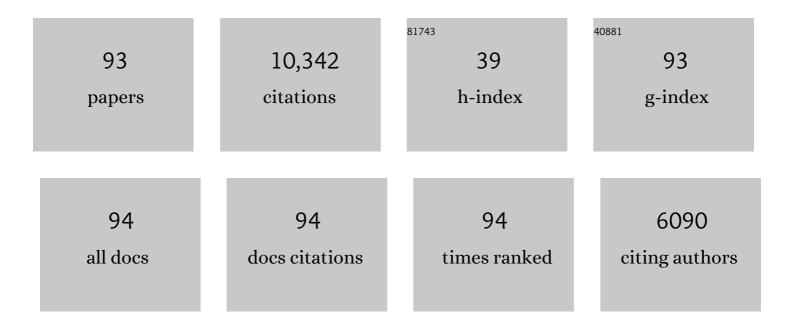
## Jose L F Abascal

List of Publications by Year in descending order

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#	Article	IF	CITATIONS
1	A general purpose model for the condensed phases of water: TIP4P/2005. Journal of Chemical Physics, 2005, 123, 234505.	1.2	2,907
2	A potential model for the study of ices and amorphous water: TIP4P/Ice. Journal of Chemical Physics, 2005, 122, 234511.	1.2	1,041
3	Simulating water with rigid non-polarizable models: a general perspective. Physical Chemistry Chemical Physics, 2011, 13, 19663.	1.3	749
4	The melting point of ice Ih for common water models calculated from direct coexistence of the solid-liquid interface. Journal of Chemical Physics, 2006, 124, 144506.	1.2	386
5	What ice can teach us about water interactions: a critical comparison of the performance of different water models. Faraday Discussions, 2009, 141, 251-276.	1.6	375
6	The melting temperature of the most common models of water. Journal of Chemical Physics, 2005, 122, 114507.	1.2	338
7	The shear viscosity of rigid water models. Journal of Chemical Physics, 2010, 132, 096101.	1.2	276
8	Widom line and the liquid–liquid critical point for the TIP4P/2005 water model. Journal of Chemical Physics, 2010, 133, 234502.	1.2	267
9	Phase Diagram of Water from Computer Simulation. Physical Review Letters, 2004, 92, 255701.	2.9	264
10	Homogeneous Ice Nucleation at Moderate Supercooling from Molecular Simulation. Journal of the American Chemical Society, 2013, 135, 15008-15017.	6.6	256
11	Vapor-liquid equilibria from the triple point up to the critical point for the new generation of TIP4P-like models: TIP4P/Ew, TIP4P/2005, and TIP4P/ice. Journal of Chemical Physics, 2006, 125, 034503.	1.2	210
12	Determination of phase diagrams via computer simulation: methodology and applications to water, electrolytes and proteins. Journal of Physics Condensed Matter, 2008, 20, 153101.	0.7	209
13	A force field of Li+, Na+, K+, Mg2+, Ca2+, Clâ^', and SO42â^' in aqueous solution based on the TIP4P/2005 water model and scaled charges for the ions. Journal of Chemical Physics, 2019, 151, 134504.	1.2	166
14	Anomalies in water as obtained from computer simulations of the TIP4P/2005 model: density maxima, and density, isothermal compressibility and heat capacity minima. Molecular Physics, 2009, 107, 365-374.	0.8	153
15	A flexible model for water based on TIP4P/2005. Journal of Chemical Physics, 2011, 135, 224516.	1.2	153
16	Relation between the melting temperature and the temperature of maximum density for the most common models of water. Journal of Chemical Physics, 2005, 123, 144504.	1.2	117
17	Radial distribution functions and densities for the SPC/E, TIP4P and TIP5P models for liquid water and ices Ih, Ic, II, III, IV, V, VI, VII, VIII, IX, XI and XII. Physical Chemistry Chemical Physics, 2005, 7, 1450.	1.3	111
18	Molecular mechanism for cavitation in water under tension. Proceedings of the National Academy of Sciences of the United States of America. 2016, 113, 13582-13587.	3.3	110

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19	Two-structure thermodynamics for the TIP4P/2005 model of water covering supercooled and deeply stretched regions. Journal of Chemical Physics, 2017, 146, 034502.	1.2	107
20	Anomalies in bulk supercooled water at negative pressure. Proceedings of the National Academy of Sciences of the United States of America, 2014, 111, 7936-7941.	3.3	103
21	The Voronoi polyhedra as tools for structure determination in simple disordered systems. The Journal of Physical Chemistry, 1993, 97, 4211-4215.	2.9	97
22	Heat capacity of water: A signature of nuclear quantum effects. Journal of Chemical Physics, 2010, 132, 046101.	1.2	90
23	A potential model for sodium chloride solutions based on the TIP4P/2005 water model. Journal of Chemical Physics, 2017, 147, 104501.	1.2	82
24	Ionic distribution around simple DNA models. I. Cylindrically averaged properties. Journal of Chemical Physics, 1995, 103, 8273-8284.	1.2	80
25	The melting temperature of the six site potential model of water. Journal of Chemical Physics, 2006, 125, 166101.	1.2	74
26	Tracing the phase diagram of the four-site water potential (TIP4P). Journal of Chemical Physics, 2004, 121, 1165-1166.	1.2	73
27	Can simple models describe the phase diagram of water?. Journal of Physics Condensed Matter, 2005, 17, S3283-S3288.	0.7	72
28	Dipole-Quadrupole Force Ratios Determine the Ability of Potential Models to Describe the Phase Diagram of Water. Physical Review Letters, 2007, 98, 237801.	2.9	69
29	Determining the phase diagram of water from direct coexistence simulations: The phase diagram of the TIP4P/2005 model revisited. Journal of Chemical Physics, 2013, 139, 154505.	1.2	61
30	Monte Carlo simulation and integral-equation studies of a fluid of charged hard spheres near the critical region. Physical Review E, 1995, 51, 289-296.	0.8	59
31	The fluid–solid equilibrium for a charged hard sphere model revisited. Journal of Chemical Physics, 2003, 119, 964-971.	1.2	59
32	Combinatorial entropy and phase diagram of partially ordered ice phases. Journal of Chemical Physics, 2004, 121, 10145-10158.	1.2	54
33	The range of meta stability of ice-water melting for two simple models of water. Molecular Physics, 2005, 103, 1-5.	0.8	54
34	The Water Forcefield:  Importance of Dipolar and Quadrupolar Interactions. Journal of Physical Chemistry C, 2007, 111, 15811-15822.	1.5	54
35	Note: Equation of state and compressibility of supercooled water: Simulations and experiment. Journal of Chemical Physics, 2011, 134, 186101.	1.2	52
36	Order-Disorder Transition in the Solid Phase of a Charged Hard Sphere Model. Physical Review Letters, 2000, 85, 3217-3220.	2.9	49

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37	Fluid-solid equilibrium of a charged hard-sphere model. Physical Review E, 1996, 54, 2746-2760.	0.8	48
38	A comprehensive scenario of the thermodynamic anomalies of water using the TIP4P/2005 model. Journal of Chemical Physics, 2016, 145, 054505.	1.2	48
39	Equation of state for water and its line of density maxima down to â~'120 MPa. Physical Chemistry Chemical Physics, 2016, 18, 5896-5900.	1.3	45
40	Perturbation theory for polyatomic fluids. Molecular Physics, 1981, 42, 999-1008.	0.8	36
41	The Madrid-2019 force field for electrolytes in water using TIP4P/2005 and scaled charges: Extension to the ions Fâ^', Brâ^', Iâ^', Rb+, and Cs+. Journal of Chemical Physics, 2022, 156, 044505.	1.2	36
42	Extension of the optimized RHNC equation to multicomponent liquids. Journal of Chemical Physics, 1987, 87, 2249-2256.	1.2	35
43	Homogeneous bubble nucleation in water at negative pressure: A Voronoi polyhedra analysis. Journal of Chemical Physics, 2013, 138, 084508.	1.2	35
44	Structure and thermodynamics of the dipolar hard sphere fluid from the referenceâ€hypernetted chain equation with minimized free energy. Journal of Chemical Physics, 1986, 85, 2916-2921.	1.2	33
45	Ionic distribution around simple B-DNA models II. Deviations from cylindrical symmetry. Journal of Chemical Physics, 1998, 109, 6200-6210.	1.2	33
46	Scaled charges at work: Salting out and interfacial tension of methane with electrolyte solutions from computer simulations. Fluid Phase Equilibria, 2020, 513, 112548.	1.4	33
47	Properties of ices at 0 K: A test of water models. Journal of Chemical Physics, 2007, 127, 154518.	1.2	32
48	The melting point of hexagonal ice (Ih) is strongly dependent on the quadrupole of the water models. Physical Chemistry Chemical Physics, 2007, 9, 2775.	1.3	30
49	Ionic distribution around simple B-DNA models. III. The effect of ionic charge. Journal of Chemical Physics, 2001, 114, 4277-4284.	1.2	29
50	lonic association in electrolyte solutions: A Voronoi polyhedra analysis. Journal of Chemical Physics, 1994, 101, 10892-10898.	1.2	27
51	Triple points and coexistence properties of the dense phases of water calculated using computer simulation. Physical Chemistry Chemical Physics, 2009, 11, 556-562.	1.3	26
52	The influence of concentration and ionic strength on the cluster structure of highly charged electrolyte solutions. Molecular Physics, 1994, 81, 143-156.	0.8	24
53	Formation of high density amorphous ice by decompression of ice VII and ice VIII at 135 K. Journal of Chemical Physics, 2004, 121, 11907-11911.	1.2	24
54	Estimating the solubility of 1:1 electrolyte aqueous solutions: the chemical potential difference rule. Molecular Physics, 2017, 115, 1301-1308.	0.8	21

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55	A unified treatment of the equation of state of hard linear bodies. Journal of Molecular Liquids, 1985, 30, 133-137.	2.3	20
56	"In Silico―Seawater. Journal of Chemical Theory and Computation, 2021, 17, 1715-1725.	2.3	20
57	The solution of the reference hypernetted chain equation for the dipolar hard diatomic fluid. Journal of Chemical Physics, 1989, 91, 2581-2586.	1.2	19
58	Theory and simulation of the triplet structure factor and triplet direct correlation functions in binary mixtures. Journal of Chemical Physics, 2002, 116, 730-736.	1.2	19
59	Detecting vapour bubbles in simulations of metastable water. Journal of Chemical Physics, 2014, 141, 18C511.	1.2	19
60	MC test of perturbation theories based on site–site potentials. Journal of Chemical Physics, 1985, 82, 2445-2452.	1.2	16
61	Cluster structure in model electrolyte solutions. Chemical Physics, 1991, 153, 79-89.	0.9	16
62	Pair connectedness functions and percolation in highly charged electrolyte solutions. Journal of Chemical Physics, 1993, 99, 9037-9046.	1.2	15
63	lce: A fruitful source of information about liquid water. Journal of Molecular Liquids, 2007, 136, 214-220.	2.3	15
64	Bubble nucleation in simple and molecular liquids via the largest spherical cavity method. Journal of Chemical Physics, 2015, 142, 154903.	1.2	15
65	The free energy difference between simple models of B- and Z-DNA: Computer simulation and theoretical predictions. Journal of Chemical Physics, 1997, 106, 8239-8253.	1.2	13
66	Nucleation free-energy barriers with Hybrid Monte-Carlo/Umbrella Sampling. Physical Chemistry Chemical Physics, 2014, 16, 24913-24919.	1.3	13
67	New aspects in the simulation and behaviour of polar molecular fluids. Molecular Physics, 1989, 68, 1067-1078.	0.8	12
68	Influence of association on the liquid–vapor phase coexistence of simple systems. Journal of Chemical Physics, 1997, 106, 1569-1575.	1.2	12
69	Physics and chemistry of water and ice. Physical Chemistry Chemical Physics, 2011, 13, 19660.	1.3	12
70	Background and bridge functions for the homonuclear hard diatomic fluid. Journal of Chemical Physics, 1989, 90, 7330-7337.	1.2	10
71	A Modulated Bulk as a Fuzzy Boundary for the Simulation of Long-Ranged Inhomogeneous Systems. Molecular Simulation, 1995, 14, 313-329.	0.9	10
72	On the triplet structure of binary liquids. Journal of Chemical Physics, 2000, 113, 3302-3309.	1.2	9

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73	The structure of liquid neon: an anomaly resolved. Journal of Physics C: Solid State Physics, 1983, 16, L441-L444.	1.5	8
74	Thermodynamics of a quadrupolar hard diatomic fluid using a perturbation theory with nonspherical reference system. The Journal of Physical Chemistry, 1989, 93, 4636-4642.	2.9	8
75	An inhomogeneous integral equation for the triplet structure of binary liquids. Journal of Chemical Physics, 2001, 114, 3562-3569.	1.2	8
76	Characterization of the order-disorder transition of a charged hard-sphere model. Physical Review E, 2003, 68, 052501.	0.8	8
77	Monte Carlo simulation of the equation of state of hard tetrahedral molecules. Molecular Physics, 1992, 76, 1411-1421.	0.8	7
78	Theory and simulation of central force model potentials: Application to homonuclear diatomic molecules. Journal of Chemical Physics, 1996, 105, 10008-10021.	1.2	7
79	A method for the computer simulation of the free-energy difference in conformational changes of polyelectrolytes. Application to the B- to Z-DNA transition. Europhysics Letters, 1996, 34, 471-476.	0.7	7
80	The role of the molecular shape on the conformational transition from B-to Z-DNA. Journal of Chemical Physics, 1999, 110, 11094-11095.	1.2	7
81	Study of the triplet and pair structure of strong electrolytes modeled via truncated Coulomb interactions. Journal of Chemical Physics, 2002, 117, 3763-3771.	1.2	6
82	Thermodynamic properties of halogens from a perturbation theory for interaction site model fluids. Journal of the Chemical Society, Faraday Transactions 2, 1982, 78, 965.	1.1	5
83	Generalized boublick equation: an accurate expression for the equation of state of hard fused spheres. Physics and Chemistry of Liquids, 1983, 12, 183-189.	0.4	5
84	Computer simulation results for the free-energy difference between B-DNA and Z-DNA. Journal of Physics Condensed Matter, 2000, 12, A327-A332.	0.7	5
85	An improved integral equation for linear polyatomic fluids. Chemical Physics Letters, 1987, 135, 133-136.	1.2	4
86	Computer simulation of the thermodynamics of the B → Z-DNA transition: effect of the ionic size and charge. Molecular Physics, 2004, 102, 2141-2148.	0.8	4
87	Computer Simulation of the Ionic Atmosphere around Z-DNA. Journal of Physical Chemistry B, 2006, 110, 25080-25090.	1.2	4
88	Effect of dissolved salt on the anomalies of water at negative pressure. Journal of Chemical Physics, 2020, 152, 194501.	1.2	4
89	Thermodynamic properties of nitrogen from a perturbation theory for ISM fluids. Chemical Physics Letters, 1982, 85, 117-119.	1.2	2
90	Monte Carlo simulations of a hydrophobic weak polyelectrolyte. Charge distribution as a function of conformation. Journal of Chemical Physics, 1993, 99, 4231-4232.	1.2	2

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91	The Radial Boundary of Polyelectrolyte Solutions Containing Asymmetrical Salts. Molecular Simulation, 1999, 21, 249-255.	0.9	2
92	Fractal analysis of ionic trajectories in electrolyte solutions. Chemical Physics Letters, 1991, 183, 125-128.	1.2	1
93	Connectivity in a binary mixture of randomly centered spheres with selective particle clustering. Journal of Chemical Physics, 1994, 100, 1769-1770.	1.2	0