

Yasuo Matubara

List of Publications by Year in descending order

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Version: 2024-02-01

20
papers

752
citations

759233

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docs citations

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times ranked

1179
citing authors

| # | ARTICLE | IF | CITATIONS |
|----|---|------|-----------|
| 1 | A Small yet Complete Framework for a Potentiostat, Galvanostat, and Electrochemical Impedance Spectrometer. <i>Journal of Chemical Education</i> , 2021, 98, 3362-3370. | 2.3 | 11 |
| 2 | A Small All-in-One Photon-Counting Device for Measuring Luminescence Decays to Determine the Lifetimes of Photoexcited Materials. <i>Journal of Chemical Education</i> , 2020, 97, 300-304. | 2.3 | 3 |
| 3 | A Bi-functional Second Coordination Sphere for Electrocatalytic CO ₂ Reduction: The Concerted Improvement by a Local Proton Source and Local Coulombic Interactions. <i>Chemistry Letters</i> , 2020, 49, 315-317. | 1.3 | 11 |
| 4 | Standard Electrode Potentials for Electrochemical Hydrogen Production, Carbon Dioxide Reduction, and Oxygen Reduction Reactions in <i>N,N</i> -Dimethylacetamide. <i>Chemistry Letters</i> , 2020, 49, 915-917. | 1.3 | 5 |
| 5 | Unified Benchmarking of Electrocatalysts in Noninnocent Second Coordination Spheres for CO ₂ Reduction. <i>ACS Energy Letters</i> , 2019, 4, 1999-2004. | 17.4 | 29 |
| 6 | Thermodynamic Cycles Relevant to Hydrogenation of CO ₂ to Formic Acid in Water and Acetonitrile. <i>Chemistry Letters</i> , 2019, 48, 627-629. | 1.3 | 9 |
| 7 | Boundary Temperatures at Which Ionic Liquid Solutions Dissolving an Electroactive Ion Start to Exhibit a Colligative Behavior. <i>Chemistry Letters</i> , 2019, 48, 925-927. | 1.3 | 0 |
| 8 | Standard Electrode Potentials for the Reduction of CO ₂ to CO in Acetonitrile/Water Mixtures Determined Using a Generalized Method for Proton-Coupled Electron-Transfer Reactions. <i>ACS Energy Letters</i> , 2017, 2, 1886-1891. | 17.4 | 53 |
| 9 | Experimental Insight into the Thermodynamics of the Dissolution of Electrolytes in Room-Temperature Ionic Liquids: From the Mass Action Law to the Absolute Standard Chemical Potential of a Proton. <i>ACS Omega</i> , 2016, 1, 1393-1411. | 3.5 | 16 |
| 10 | Striking Differences in Properties of Geometric Isomers of [Ir(tpy)(ppy)H] ⁺ : Experimental and Computational Studies of their Hydricities, Interaction with CO ₂ , and Photochemistry. <i>Angewandte Chemie - International Edition</i> , 2015, 54, 14128-14132. | 13.8 | 51 |
| 11 | Thermodynamic Aspects of Electrocatalytic CO ₂ Reduction in Acetonitrile and with an Ionic Liquid as Solvent or Electrolyte. <i>ACS Catalysis</i> , 2015, 5, 6440-6452. | 11.2 | 162 |
| 12 | Hydride Reduction of NAD(P) ⁺ Model Compounds with a Ru(II) Hydrido Complex. <i>Organometallics</i> , 2015, 34, 5530-5539. | 2.3 | 13 |
| 13 | Reactivity of a fac-ReCl(±-diimine)(CO) ₃ complex with an NAD ⁺ model ligand toward CO ₂ reduction. <i>Chemical Communications</i> , 2014, 50, 728-730. | 4.1 | 22 |
| 14 | Electrocatalytic CO ₂ Reduction with a Homogeneous Catalyst in Ionic Liquid: High Catalytic Activity at Low Overpotential. <i>Journal of Physical Chemistry Letters</i> , 2014, 5, 2033-2038. | 4.6 | 108 |
| 15 | Formation of π -Coordinated Dihydropyridine Ruthenium(II) Complexes by Hydride Transfer from Ruthenium(II) to Pyridinium Cations. <i>Organometallics</i> , 2013, 32, 6162-6165. | 2.3 | 11 |
| 16 | Photochemistry of fac-[Re(bpy)(CO) ₃ Cl]. <i>Chemistry - A European Journal</i> , 2012, 18, 15722-15734. | 3.3 | 74 |
| 17 | Thermodynamic and Kinetic Hydricity of Ruthenium(II) Hydride Complexes. <i>Journal of the American Chemical Society</i> , 2012, 134, 15743-15757. | 13.7 | 117 |
| 18 | Development of an Efficient and Durable Photocatalytic System for Hydride Reduction of an NAD(P) ⁺ Model Compound Using a Ruthenium(II) Complex Based on Mechanistic Studies. <i>Journal of the American Chemical Society</i> , 2010, 132, 10547-10552. | 13.7 | 35 |

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|----|---|-----|-----------|
| 19 | Quantitative Photochemical Formation of [Ru(tpy)(bpy)H] ⁺ . Inorganic Chemistry, 2009, 48, 10138-10145. | 4.0 | 12 |
| 20 | Colloidal platinum nanoparticles dispersed by polyvinylpyrrolidone and poly(diallyldimethylammonium chloride) with high catalytic activity for hydrogen production based on formate decomposition. Sustainable Energy and Fuels, 0, , . | 4.9 | 0 |