Marwa Ammar

List of Publications by Year in descending order

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Version: 2024-02-01

1307594 1474206 91 10 7 9 citations g-index h-index papers 53 10 10 10 citing authors docs citations times ranked all docs

#	Article	lF	CITATIONS
1	Crystalline Swelling Process of Mg-Exchanged Montmorillonite: Effect of External Environmental Solicitation. Advances in Civil Engineering, 2018, 2018, 1-18.	0.7	2
2	Quantitative XRD Analysis of the Structural Changes of Ba-Exchanged Montmorillonite: Effect of an in Situ Hydrous Perturbation. Minerals (Basel, Switzerland), 2015, 5, 507-526.	2.0	19
3	The water retention mechanism of a Cs+ and Na+ exchanged montmorillonite: effect of relative humidity and ionic radius on the interlayer. Powder Diffraction, 2015, 30, S70-S75.	0.2	3
4	Interlamellar Space Configuration under Variable Environmental Conditions in the Case of Ni-Exchanged Montmorillonite: Quantitative XRD Analysis. Journal of Nanomaterials, 2014, 2014, 1-13.	2.7	9
5	Hydration performance of dioctahedral smectite saturated with Ba $<$ sup $>$ 2+ $<$ /sup $>$ and Cs $<$ sup $>$ + $<$ /sup $>$ cations: Quantitative XRD investigation. , 2014, , .		0
6	Quantitative XRD analysis of the dehydration–hydration performance of (Na+, Cs+) exchanged smectite. Desalination and Water Treatment, 2014, 52, 4314-4333.	1.0	12
7	Effect of the hydration sequence orientation on the structural properties of Hg exchanged montmorillonite: Quantitative XRD analysis. Journal of Environmental Chemical Engineering, 2014, 2, 1604-1611.	6.7	12
8	XRD profile modeling approach tools to investigate the effect of charge location on hydration behavior in the case of metal exchanged smectite. Powder Diffraction, 2013, 28, S284-S300.	0.2	8
9	Effect of temperature and pH value on cation exchange performance of a natural clay for selective (Cu2+, Co2+) removal: Equilibrium, sorption and kinetics. Progress in Natural Science: Materials International, 2013, 23, 23-35.	4.4	17
10	Effect of an "in situ―hydrous strain on the ionic exchange process of dioctahedral smectite: Case of solution containing (Cu2+, Co2+) cations. Applied Surface Science, 2012, 258, 9032-9040.	6.1	9