Sayantan Paria

List of Publications by Year in descending order

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#	Article	IF	CITATIONS
1	Highly Luminescent Mn-Doped ZnS Nanocrystals: Gram-Scale Synthesis. Journal of Physical Chemistry Letters, 2010, 1, 1454-1458.	4.6	192
2	Oxidative Decarboxylation of Benzilic Acid by a Biomimetic Iron(II) Complex: Evidence for an Iron(IV)–Oxo–Hydroxo Oxidant from O ₂ . Angewandte Chemie - International Edition, 2011, 50, 11129-11132.	13.8	71
3	Oxidative Carbonï£;Carbon Bond Cleavage of a αâ€Hydroxy Ketone by a Functional Model of 2,4′â€Dihydroxyacetophenone Dioxygenase. Angewandte Chemie - International Edition, 2012, 51, 6195-6199	.13.8	51
4	Reactivity of an Iron–Oxygen Oxidant Generated upon Oxidative Decarboxylation of Biomimetic Iron(II) α-Hydroxy Acid Complexes. Inorganic Chemistry, 2014, 53, 2810-2821.	4.0	41
5	Dioxygen Reactivity of Biomimetic Iron–Catecholate and Iron– <i>o</i> â€Aminophenolate Complexes of a Tris(2â€pyridylthio)methanido Ligand: Aromatic CC Bond Cleavage of Catecholate versus <i>o</i> â€Iminobenzosemiquinonate Radical Formation. Chemistry - A European Journal, 2012, 18, 11778-11787.	3.3	36
6	Oxidative decarboxylation of $\hat{l}\pm$ -hydroxy acids by a functional model of the nonheme iron oxygenase, CloR. Chemical Communications, 2010, 46, 1830-1832.	4.1	34
7	Tetrahedral Copper(II) Complexes with a Labile Coordination Site Supported by a Tris-tetramethylguanidinato Ligand. Inorganic Chemistry, 2017, 56, 9634-9645.	4.0	34
8	A Functional Model of Extradiol-Cleaving Catechol Dioxygenases: Mimicking the 2-His-1-Carboxylate Facial Triad. Inorganic Chemistry, 2010, 49, 4518-4523.	4.0	31
9	Generation, Characterization, and Reactivity of a Cu ^{II} –Alkylperoxide/Anilino Radical Complex: Insight into the O–O Bond Cleavage Mechanism. Journal of the American Chemical Society, 2015, 137, 10870-10873.	13.7	29
10	Reductive Activation of O ₂ by Non-Heme Iron(II) Benzilate Complexes of N ₄ Ligands: Effect of Ligand Topology on the Reactivity of O ₂ -Derived Oxidant. Inorganic Chemistry, 2017, 56, 359-371.	4.0	25
11	Copper(II)-mediated oxidation of 1,2-dioxime to furoxan. Tetrahedron Letters, 2008, 49, 5924-5927.	1.4	22
12	Copper(II)-Mediated Oxidative Transformation of <i>vic</i> -Dioxime to Furoxan: Evidence for a Copper(II)-Dinitrosoalkene Intermediate. Inorganic Chemistry, 2011, 50, 11375-11383.	4.0	18
13	Electrocatalytic Water Oxidation Activity of Molecular Copper Complexes: Effect of Redox-Active Ligands. Inorganic Chemistry, 2022, 61, 3152-3165.	4.0	14
14	Synthesis, characterization and emission study of zinc(II) and cobalt(II) complexes: Bis(bidentate) iminophenols as zinc(II) selective fluorescence probes. Inorganica Chimica Acta, 2012, 387, 332-337.	2.4	13
15	Aliphatic C–C Bond Cleavage of α-Hydroxy Ketones by Non-Heme Iron(II) Complexes: Mechanistic Insight into the Reaction Catalyzed by 2,4′-Dihydroxyacetophenone Dioxygenase. Inorganic Chemistry, 2015, 54, 10576-10586.	4.0	13
16	Characterization and reactivity study of non-heme high-valent iron–hydroxo complexes. Chemical Science, 2021, 12, 4418-4424.	7.4	12
17	Characterization and Reactivity of a Tetrahedral Copper(II) Alkylperoxido Complex. Chemistry - A European Journal, 2019, 25, 11157-11165.	3.3	11
18	Copper(I)–Dioxygen Reactivity in the Isolated Cavity of a Nanoscale Molecular Architecture. European Journal of Inorganic Chemistry, 2018, 2018, 1976-1983.	2.0	10

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19	Electrochemical Properties and Reactivity Study of [Mn ^V (O)(μ-OR–Lewis Acid)] Cores. Inorganic Chemistry, 2021, 60, 18006-18016.	4.0	10
20	Generation and characterisation of a stable nickel(ii)-aminoxyl radical complex. Dalton Transactions, 2017, 46, 8013-8016.	3.3	6
21	Structure and Reactivity of Copper Complexes Supported by a Bulky Tripodal N 4 Ligand: Copper(I)/Dioxygen Reactivity and Formation of a Hydroperoxide Copper(II) Complex. Zeitschrift Fur Anorganische Und Allgemeine Chemie, 2018, 644, 780-789.	1.2	5
22	Dioxygen Activation and Mandelate Decarboxylation by Iron(II) Complexes of N4 Ligands: Evidence for Dioxygen-Derived Intermediates from Cobalt Analogues. Inorganic Chemistry, 2022, 61, 10461-10476.	4.0	2