

Mikhail A Ryumin

List of Publications by Year in descending order

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papers

591
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686830

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#	ARTICLE	IF	CITATIONS
1	Heat capacity and thermodynamic functions of LuPO ₄ in the range 0–320K. <i>Thermochimica Acta</i> , 2006, 448, 63-65.	1.2	51
2	Refined heat capacity of LaPO ₄ in the temperature range 0–1600K. <i>Thermochimica Acta</i> , 2008, 474, 47-51.	1.2	34
3	Novel heterometallic polymeric lanthanide acetylacetonates with bridging cymantrenecarboxylate groups – synthesis, magnetism and thermolysis. <i>Polyhedron</i> , 2015, 102, 48-59.	1.0	31
4	Synthesis, structure and thermolysis of Ba(II)–M(II) (M = Co, Zn) bimetallic 3D-polymers as precursors of complex oxides. <i>Polyhedron</i> , 2015, 87, 28-37.	1.0	29
5	Thermodynamic functions of erbium orthophosphate ErPO ₄ in the temperature range of 0–1600K. <i>Thermochimica Acta</i> , 2012, 535, 1-7.	1.2	26
6	Heat capacity and thermodynamic functions of xenotime YPO ₄ (c) at 0–1600 K. <i>Geochemistry International</i> , 2010, 48, 932-939.	0.2	24
7	Heat Capacity and Thermodynamic Functions of La ₂ Sn ₂ O ₇ . <i>Inorganic Materials</i> , 2020, 56, 97-104.	0.2	19
8	The heat capacity and thermodynamic functions of EuPO ₄ over the temperature range 0–1600 K. <i>Russian Journal of Physical Chemistry A</i> , 2009, 83, 901-906.	0.1	18
9	Heat capacity and thermodynamic functions of YbPO ₄ from 0 to 1800 K. <i>Inorganic Materials</i> , 2013, 49, 701-708.	0.2	18
10	Thermodynamic properties of GdTaO ₄ . <i>Inorganic Materials</i> , 2017, 53, 728-733.	0.2	18
11	Low-temperature heat capacity of yttrium orthotantalate. <i>Inorganic Materials</i> , 2016, 52, 1149-1154.	0.2	16
12	Low-temperature heat capacity and thermodynamic functions of DyVO ₄ . <i>Inorganic Materials</i> , 2014, 50, 917-923.	0.2	14
13	High-temperature thermodynamic properties of LuPO ₄ . <i>Inorganic Materials</i> , 2012, 48, 841-844.	0.2	13
14	Synthesis of cerium orthophosphates with monazite and rhabdophane structure from phosphoric acid solutions in the presence of hydrogen peroxide. <i>Russian Journal of Inorganic Chemistry</i> , 2016, 61, 1219-1224.	0.3	13
15	Potassium europium molybdate phosphate. <i>Russian Journal of Inorganic Chemistry</i> , 2010, 55, 1010-1013.	0.3	12
16	Heat capacity and thermodynamic functions of LaVO ₄ and LuVO ₄ from 7 to 345 K. <i>Inorganic Materials</i> , 2010, 46, 776-783.	0.2	12
17	Heat capacity and thermodynamic properties of GdPO ₄ in the temperature range 0–1600 K. <i>Geochemistry International</i> , 2012, 50, 702-710.	0.2	12
18	Thermal expansion and thermodynamic properties of M ²⁺ -YbTaO ₄ ceramics. <i>Ceramics International</i> , 2020, 46, 5402-5406.	2.3	12

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19	Thermal behavior of LaPO ₄ ·nH ₂ O and NdPO ₄ ·nH ₂ O nanopowders. <i>Journal of Thermal Analysis and Calorimetry</i> , 2010, 102, 809-811.	2.0	11
20	Revised heat capacity and thermodynamic functions of GdVO ₄ . <i>Inorganic Materials</i> , 2011, 47, 1120-1125.	0.2	11
21	Heat capacity and thermodynamic functions of pretulite ScPO ₄ (c) at 0–1600 K. <i>Geochemistry International</i> , 2010, 48, 390-397.	0.2	10
22	Thermodynamic Properties of Monoclinic Neodymium Orthotantalate M-NdTaO ₄ . <i>Russian Journal of Inorganic Chemistry</i> , 2019, 64, 1041-1046.	0.3	10
23	Low-temperature heat capacity and thermodynamic properties of PrPO ₄ . <i>Geochemistry International</i> , 2016, 54, 362-368.	0.2	9
24	Thermophysical properties of M ²⁺ -LuTaO ₄ : Structural and calorimetric studies. <i>Journal of Alloys and Compounds</i> , 2019, 803, 1016-1022.	2.8	9
25	Synthesis and crystal structure of new complex sodium lanthanide phosphate molybdates Na ₂ M ^{III} (MoO ₄)(PO ₄)(M ^{III} = Tb, Dy, Ho, Er, Tm, Lu). <i>Russian Journal of Inorganic Chemistry</i> , 2007, 52, 653-660.	0.3	8
26	Complex potassium yttrium molybdate phosphates K ₂ Y _{1-x} Eu _x (MoO ₄)(PO ₄) _{0.9} (VO ₄) _{0.1} : Synthesis and study of luminescent properties. <i>Russian Journal of Inorganic Chemistry</i> , 2011, 56, 1343-1350.	0.3	7
27	New complex ytterbium molybdophosphates M ₂ Yb(PO ₄)(MoO ₄) (M = K, Na): Synthesis and structure solution. <i>Russian Journal of Inorganic Chemistry</i> , 2006, 51, 350-356.	0.3	6
28	Heat capacity and thermodynamic functions of Na ₂ MoO ₄ in the temperature range 0–300K. <i>Thermochimica Acta</i> , 2007, 463, 41-43.	1.2	6
29	Low-temperature heat capacity and thermal behavior of Zn _{0.98} Co _{0.02} O in the high-temperature region. <i>Russian Journal of Inorganic Chemistry</i> , 2009, 54, 1-5.	0.3	6
30	Heat capacity and thermodynamic functions of YVO ₄ in the 13–347 K region. <i>Russian Journal of Inorganic Chemistry</i> , 2010, 55, 1935-1939.	0.3	6
31	Heat capacity and thermodynamic functions of petrovskaita (AgAuS) at 0–583 K and mineral equilibria in the Ag-Au-S system. <i>Geochemistry International</i> , 2011, 49, 422-428.	0.2	6
32	Phase transformations in sodium molybdate studied by differential scanning calorimetry. <i>Russian Journal of Inorganic Chemistry</i> , 2012, 57, 1123-1127.	0.3	6
33	Thermodynamic properties and phase transition of monoclinic terbium orthophosphate. <i>Thermochimica Acta</i> , 2016, 641, 63-70.	1.2	6
34	Phase transitions and thermodynamic properties of lanthanide compounds LnAO ₄ (A = P, V, Nb). <i>Russian Journal of General Chemistry</i> , 2017, 87, 583-590.	0.3	6
35	Calorimetric study of ytterbium orthovanadate YbVO ₄ polycrystalline ceramics. <i>Ceramics International</i> , 2018, 44, 18103-18107.	2.3	6
36	Heat Capacity and Thermal Expansion of Neodymium Orthotantalate. <i>Inorganic Materials</i> , 2019, 55, 959-963.	0.2	6

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37	Thermodynamic functions of ScVO ₄ at temperatures from 0 to 350 K. <i>Inorganic Materials</i> , 2012, 48, 845-850.	0.2	5
38	Heat capacity and thermodynamic functions of SmPO ₄ at 10 ² –1600 K. <i>Geochemistry International</i> , 2015, 53, 607-616.	0.2	5
39	Heat capacity and thermodynamic functions of Gd(VO ₄) _{0.5} (PO ₄) _{0.5} solid solution in the low-temperature region. <i>Russian Journal of Inorganic Chemistry</i> , 2015, 60, 702-708.	0.3	5
40	The heat capacity of LaPO ₄ and PrPO ₄ nanowhiskers. <i>Journal of Thermal Analysis and Calorimetry</i> , 2018, 132, 337-342.	2.0	5
41	Thermodynamic functions of holmium orthophosphate HoPO ₄ in the range 9 ² –1370 K. <i>Thermochimica Acta</i> , 2020, 683, 178459.	1.2	5
42	Synthesis, single crystal growth and thermodynamic properties of SrNdAlO ₄ -CaNdAlO ₄ solid solutions. <i>Crystal Research and Technology</i> , 2005, 40, 405-409.	0.6	4
43	A calorimetric study of the thermodynamic properties of potassium molybdate. <i>Russian Journal of Physical Chemistry A</i> , 2009, 83, 327-333.	0.1	4
44	Synthesis and study of the properties of K ₂ Y _{1-x} Eu _x Tb _y (MoO ₄)(PO ₄) and K ₂ Y _{1-x} Eu _x Tb _y (MoO ₄)(PO ₄) _{1-y} (VO ₄) _y solid solutions. <i>Russian Journal of Inorganic Chemistry</i> , 2011, 56, 1943-1950.	0.3	4
45	Heat capacity and thermodynamic properties of Ba _{0.7} La _{0.3} F _{2.3} solid solution from 5 to 1000K. <i>Thermochimica Acta</i> , 2013, 558, 1-5.	1.2	4
46	Low-Temperature Heat Capacity of M-Type Terbium Orthotantalate and Schottky Anomaly. <i>Russian Journal of Inorganic Chemistry</i> , 2020, 65, 655-662.	0.3	4
47	Thermodynamic and Magnetic Properties of Praseodymium Stannate. <i>Russian Journal of Inorganic Chemistry</i> , 2020, 65, 1891-1898.	0.3	4
48	Low-temperature heat capacity and high-temperature thermal behavior of sodium lutetium molybdate phosphate Na ₂ Lu(MoO ₄)(PO ₄). <i>Russian Journal of Inorganic Chemistry</i> , 2007, 52, 727-732.	0.3	3
49	Heat capacity and thermodynamic functions of cadmium fluoride. <i>Russian Journal of Inorganic Chemistry</i> , 2009, 54, 1445-1450.	0.3	3
50	Thermodynamic properties of tetrabridged binuclear copper complexes with apical substituted pyridine ligands. <i>Thermochimica Acta</i> , 2010, 509, 67-72.	1.2	3
51	Synthesis and study of the heat capacity of orthovanadate TbVO ₄ in the range 5 ² –859 K. <i>Russian Journal of Inorganic Chemistry</i> , 2016, 61, 1-6.	0.3	3
52	Heat capacity and thermodynamic functions of thulium orthophosphate TmPO ₄ in the range of 10 ² –1350 K. <i>Russian Journal of Physical Chemistry A</i> , 2017, 91, 2310-2316.	0.1	3
53	Heat Capacity and Thermodynamic Functions of Neodymium Orthoniobate. <i>Russian Journal of Inorganic Chemistry</i> , 2021, 66, 237-244.	0.3	3
54	Thermodynamic Properties of CaNdAlO ₄ -SrNdAlO ₄ Solid Solutions. <i>Inorganic Materials</i> , 2005, 41, 850-853.	0.2	2

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55	Low-Temperature Heat Capacities of Terbium Molybdate Phosphates $M_2Tb(MoO_4)(PO_4)$ ($M = Na$ or K). Russian Journal of Inorganic Chemistry, 2008, 53, 268-274.	0.3	2
56	Synthesis and study of the properties of the solid solutions $K_2Y_{1-x}Tb_x(MoO_4)(PO_4)$ and $K_2Y_{1-x}Tm_x(MoO_4)(PO_4)_{0.95}(VO_4)_{0.05}$. Russian Chemical Bulletin, 2012, 61, 659-661.	0.4	2
57	Thermodynamic properties of mixed-ligand rare earth pivalates. Thermochemica Acta, 2013, 556, 68-74.	1.2	2
58	Structural and thermodynamic properties of anionically substituted solid solution ceramics $Y(VO_4)_{1-x}(PO_4)_x$ with zircon structure. Ceramics International, 2020, 46, 2576-2579.	2.3	2
59	Heat Capacity and Thermodynamic Functions of Dysprosium Orthoniobate in the Range 2×10^2 –1300 K. Russian Journal of Inorganic Chemistry, 2020, 65, 688-694.	0.3	2
60	Low-temperature heat capacity of sodium erbium molybdophosphate $Na_2Er(MoO_4)(PO_4)$. Russian Journal of Inorganic Chemistry, 2007, 52, 1607-1611.	0.3	1
61	Low-temperature heat capacity of $Li_xNi_{2-x}O_2$ solid solutions. Inorganic Materials, 2010, 46, 1031-1037.	0.2	1
62	Heat capacity and thermodynamic functions of $La(VO_4)_{1-x}(PO_4)_x$ solid solutions at low temperatures. Russian Journal of Inorganic Chemistry, 2017, 62, 77-83.	0.3	1
63	Thermodynamic Properties of $M-EuTaO_4$. Russian Journal of Inorganic Chemistry, 2020, 65, 1873-1878.	0.3	1
64	Phase transitions in the solvate of the heterospin complex $Cu(hfac)_2$ with the nitronyl nitroxide radical. Russian Chemical Bulletin, 2013, 62, 403-407.	0.4	0
65	Low-temperature heat capacity and Schottky anomaly of $ErVO_4$. Russian Journal of Physical Chemistry A, 2017, 91, 727-732.	0.1	0