

Julian Morales

List of Publications by Year in descending order

Source: <https://exaly.com/author-pdf/217623/publications.pdf>

Version: 2024-02-01

181
papers

6,808
citations

57758

44
h-index

82547

72
g-index

181
all docs

181
docs citations

181
times ranked

8026
citing authors

#	ARTICLE	IF	CITATIONS
1	Recent advances in lithium-sulfur batteries using biomass-derived carbons as sulfur host. <i>Renewable and Sustainable Energy Reviews</i> , 2022, 154, 111783.	16.4	83
2	Synergistic effect between PPy:PSS copolymers and biomass-derived activated carbons: a simple strategy for designing sustainable high-performance Li-S batteries. <i>Sustainable Energy and Fuels</i> , 2022, 6, 1568-1586.	4.9	14
3	A Stable High-Capacity Lithium-Ion Battery Using a Biomass-Derived Sulfur-Carbon Cathode and Lithiated Silicon Anode. <i>ChemSusChem</i> , 2021, 14, 3333-3343.	6.8	16
4	Contribution to the understanding of the performance differences between commercial current collectors in Li-S batteries. <i>Journal of Energy Chemistry</i> , 2021, 62, 295-306.	12.9	16
5	Revisiting the HKUST-1/S Composite as an Electrode for Li-S Batteries: Inherent Problems That Hinder Its Performance. <i>European Journal of Inorganic Chemistry</i> , 2021, 2021, 177-185.	2.0	6
6	Biomass Porous Carbons Derived from Banana Peel Waste as Sustainable Anodes for Lithium-Ion Batteries. <i>Materials</i> , 2021, 14, 5995.	2.9	13
7	Insights into the formation of N doped 3D-graphene quantum dots. Spectroscopic and computational approach. <i>Journal of Colloid and Interface Science</i> , 2020, 561, 678-686.	9.4	35
8	Highly graphitized carbon nanosheets with embedded Ni nanocrystals as anode for Li-ion batteries. <i>Nano Research</i> , 2020, 13, 86-94.	10.4	14
9	Lithium-Oxygen Battery Exploiting Highly Concentrated Glyme-Based Electrolytes. <i>ACS Applied Energy Materials</i> , 2020, 3, 12263-12275.	5.1	22
10	Porous Cr ₂ O ₃ @C composite derived from metal organic framework in efficient semi-liquid lithium-sulfur battery. <i>Materials Chemistry and Physics</i> , 2020, 255, 123484.	4.0	19
11	Pistachio Shell-Derived Carbon Activated with Phosphoric Acid: A More Efficient Procedure to Improve the Performance of Li-S Batteries. <i>Nanomaterials</i> , 2020, 10, 840.	4.1	33
12	Simple and Sustainable Preparation of Nonactivated Porous Carbon from Brewing Waste for High-Performance Lithium-Sulfur Batteries. <i>ChemSusChem</i> , 2020, 13, 3439-3446.	6.8	25
13	MIL-88A Metal-Organic Framework as a Stable Sulfur-Host Cathode for Long-Cycle Li-S Batteries. <i>Nanomaterials</i> , 2020, 10, 424.	4.1	44
14	Alternative lithium-ion battery using biomass-derived carbons as environmentally sustainable anode. <i>Journal of Colloid and Interface Science</i> , 2020, 573, 396-408.	9.4	67
15	A Comparative Study of Particle Size Distribution of Graphene Nanosheets Synthesized by an Ultrasound-Assisted Method. <i>Nanomaterials</i> , 2019, 9, 152.	4.1	89
16	Physical activation of graphene: An effective, simple and clean procedure for obtaining microporous graphene for high-performance Li/S batteries. <i>Nano Research</i> , 2019, 12, 759-766.	10.4	38
17	High capacity semi-liquid lithium sulfur cells with enhanced reversibility for application in new-generation energy storage systems. <i>Journal of Power Sources</i> , 2019, 412, 575-585.	7.8	23
18	A Lithium-Ion Battery using a 3D-Array Nanostructured Graphene-Sulfur Cathode and a Silicon Oxide-Based Anode. <i>ChemSusChem</i> , 2018, 11, 1512-1520.	6.8	46

#	ARTICLE	IF	CITATIONS
19	Low-cost disordered carbons for Li/S batteries: A high-performance carbon with dual porosity derived from cherry pits. <i>Nano Research</i> , 2018, 11, 89-100.	10.4	88
20	Biomass-derived carbon/ γ -MnO ₂ nanorods/S composites prepared by facile procedures with improved performance for Li/S batteries. <i>Electrochimica Acta</i> , 2018, 292, 522-531.	5.2	28
21	The Role of Current Collector in Enabling the High Performance of Li/S Battery. <i>ChemistrySelect</i> , 2018, 3, 10371-10377.	1.5	22
22	Versatility of a Nitrogen-Containing Monolithic Porous Carbon for Lithium-Based Energy Storage.. <i>ChemistrySelect</i> , 2018, 3, 8560-8567.	1.5	3
23	Lithium sulfur battery exploiting material design and electrolyte chemistry: 3D graphene framework and diglyme solution. <i>Journal of Power Sources</i> , 2018, 397, 102-112.	7.8	37
24	Almond Shell as a Microporous Carbon Source for Sustainable Cathodes in Lithium-Sulfur Batteries. <i>Materials</i> , 2018, 11, 1428.	2.9	42
25	Simultaneous recovery of Zn and Mn from used batteries in acidic and alkaline mediums: A comparative study. <i>Waste Management</i> , 2017, 68, 518-526.	7.4	43
26	Lithium battery using sulfur infiltrated in three-dimensional flower-like hierarchical porous carbon electrode. <i>Materials Chemistry and Physics</i> , 2016, 180, 82-88.	4.0	23
27	Improved performance of electrodes based on carbonized olive stones/S composites by impregnating with mesoporous TiO ₂ for advanced Li-S batteries. <i>Journal of Power Sources</i> , 2016, 313, 21-29.	7.8	39
28	Solvothermal-induced 3D graphene networks: Role played by the structural and textural properties on lithium storage. <i>Electrochimica Acta</i> , 2016, 222, 914-920.	5.2	13
29	Use of Polyelectrolytes for the Fabrication of Porous NiO Films by Electrophoretic Deposition for Supercapacitor Electrodes. <i>Electrochimica Acta</i> , 2016, 211, 110-118.	5.2	35
30	A long-life lithium ion sulfur battery exploiting high performance electrodes. <i>Chemical Communications</i> , 2015, 51, 14540-14542.	4.1	37
31	Deficiencies of Chemically Reduced Graphene as Electrode in Full Li-Ion Cells. <i>Electrochimica Acta</i> , 2015, 165, 365-371.	5.2	19
32	Nickel Oxide/Nickel Foam Composite as Supercapacitor Electrode via Electrophoretic Deposition. <i>Key Engineering Materials</i> , 2015, 654, 58-64.	0.4	3
33	Relevance of the Semiconductor Microstructure in the Pseudocapacitance of the Electrodes Fabricated by EPD of Binder-Free γ -Ni(OH) ₂ Nanoplatelets. <i>Journal of the Electrochemical Society</i> , 2015, 162, D3001-D3012.	2.9	21
34	Cyclability of binder-free γ -Ni(OH) ₂ anodes shaped by EPD for Li-ion batteries. <i>Journal of the European Ceramic Society</i> , 2015, 35, 573-584.	5.7	17
35	Efficient behaviour of hematite towards the photocatalytic degradation of NO gases. <i>Applied Catalysis B: Environmental</i> , 2015, 165, 529-536.	20.2	63
36	Enhanced Electrochemical Performance of Maghemite/Graphene Nanosheets Composite as Electrode in Half and Full Li-Ion Cells. <i>Electrochimica Acta</i> , 2014, 130, 551-558.	5.2	51

#	ARTICLE	IF	CITATIONS
37	Lithium–sulfur batteries with activated carbons derived from olive stones. <i>Carbon</i> , 2014, 70, 241-248.	10.3	112
38	Insights on the electrode/electrolyte interfaces in LiFePO ₄ based cells with LiAl(Al) and Li(Mg) anodes. <i>Journal of Electroanalytical Chemistry</i> , 2014, 732, 53-60.	3.8	6
39	Ordered mesoporous carbons obtained by a simple soft template method as sulfur immobilizers for lithium–sulfur cells. <i>Physical Chemistry Chemical Physics</i> , 2014, 16, 17332-17340.	2.8	35
40	On the potential use of carbon-free mesoporous precursors of LiFePO ₄ for lithium-ion batteries electrode. <i>Solid State Ionics</i> , 2014, 255, 30-38.	2.7	5
41	Contribution to the Understanding of Capacity Fading in Graphene Nanosheets Acting as an Anode in Full Li-Ion Batteries. <i>ACS Applied Materials & Interfaces</i> , 2014, 6, 3290-3298.	8.0	40
42	Improving the electrochemical properties of nanosized LiFePO ₄ -based electrode by boron doping. <i>Electrochimica Acta</i> , 2014, 135, 558-567.	5.2	29
43	A High-Capacity Anode for Lithium Batteries Consisting of Mesoporous NiO Nanoplatelets. <i>Energy & Fuels</i> , 2013, 27, 5545-5551.	5.1	49
44	Enhanced photocatalytic degradation of NO _x gases by regulating the microstructure of mortar cement modified with titanium dioxide. <i>Building and Environment</i> , 2013, 69, 55-63.	6.9	90
45	Controlling microstructure in cement based mortars by adjusting the particle size distribution of the raw materials. <i>Construction and Building Materials</i> , 2013, 41, 139-145.	7.2	15
46	Preparation of Sustainable Photocatalytic Materials through the Valorization of Industrial Wastes. <i>ChemSusChem</i> , 2013, 6, 2340-2347.	6.8	9
47	Electrochemical performance of a graphene nanosheets anode in a high voltage lithium-ion cell. <i>Physical Chemistry Chemical Physics</i> , 2013, 15, 20444.	2.8	27
48	Electrochemical properties of ultrasonically prepared Ni(OH) ₂ nanosheets in lithium cells. <i>Journal of Power Sources</i> , 2013, 238, 366-371.	7.8	21
49	On the limited electroactivity of Li ₂ NiTiO ₄ nanoparticles in lithium batteries. <i>Electrochimica Acta</i> , 2013, 100, 93-100.	5.2	19
50	Aqueous Rechargeable Lithium Battery Based on LiNi _{0.5} Mn _{1.5} O ₄ Spinel with Promising Performance. <i>Energy & Fuels</i> , 2013, 27, 7854-7857.	5.1	18
51	Vapor-Phase Fabrication of Iron Oxide Nanopyramids for Lithium-Ion Battery Anodes. <i>ChemPhysChem</i> , 2012, 13, 3798-3801.	2.1	21
52	On the Performances of Cu _x O-TiO ₂ (x = 1, 2) Nanomaterials As Innovative Anodes for Thin Film Lithium Batteries. <i>ACS Applied Materials & Interfaces</i> , 2012, 4, 3610-3619.	8.0	64
53	Can the performance of graphene nanosheets for lithium storage in Li-ion batteries be predicted?. <i>Nanoscale</i> , 2012, 4, 2083.	5.6	129
54	Use of Industrial Waste for the Manufacturing of Sustainable Building Materials. <i>ChemSusChem</i> , 2012, 5, 694-699.	6.8	9

#	ARTICLE	IF	CITATIONS
55	Influence of the lithium salt electrolyte on the electrochemical performance of copper/LiFePO ₄ composites. <i>Electrochimica Acta</i> , 2012, 61, 57-63.	5.2	9
56	Anchoring Si nanoparticles to carbon nanofibers: an efficient procedure for improving Si performance in Li batteries. <i>Journal of Materials Chemistry</i> , 2011, 21, 811-818.	6.7	37
57	Use of Olive Biomass Fly Ash in the Preparation of Environmentally Friendly Mortars. <i>Environmental Science & Technology</i> , 2011, 45, 6991-6996.	10.0	44
58	XAS study of the reversible reactivity mechanism of micro- and nanostructured electrodeposited Cu ₂ O thin films towards lithium. <i>Journal of Materials Chemistry</i> , 2011, 21, 5368.	6.7	24
59	Limitations of Disordered Carbons Obtained from Biomass as Anodes for Real Lithium-Ion Batteries. <i>ChemSusChem</i> , 2011, 4, 658-663.	6.8	87
60	Improved capacitive properties of layered manganese dioxide grown as nanowires. <i>Journal of Power Sources</i> , 2011, 196, 3350-3354.	7.8	54
61	Cobalt Oxide Nanomaterials by Vapor-Phase Synthesis for Fast and Reversible Lithium Storage. <i>Journal of Physical Chemistry C</i> , 2010, 114, 10054-10060.	3.1	61
62	Electrochemical instability of LiV ₃ O ₈ as an electrode material for aqueous rechargeable lithium batteries. <i>Journal of Power Sources</i> , 2010, 195, 4318-4321.	7.8	57
63	Re-examining the effect of ZnO on nanosized 5V LiNi _{0.5} Mn _{1.5} O ₄ spinel: An effective procedure for enhancing its rate capability at room and high temperatures. <i>Journal of Power Sources</i> , 2010, 195, 4278-4284.	7.8	97
64	Use of granite sludge wastes for the production of coloured cement-based mortars. <i>Cement and Concrete Composites</i> , 2010, 32, 617-622.	10.7	119
65	Cycling-induced stress in lithium ion negative electrodes: LiAl/LiFePO ₄ and Li ₄ Ti ₅ O ₁₂ /LiFePO ₄ cells. <i>Electrochimica Acta</i> , 2010, 55, 3075-3082.	5.2	40
66	3D Gold Nanocrystal Arrays: A Framework for Reversible Lithium Storage. <i>Journal of Physical Chemistry C</i> , 2010, 114, 2360-2364.	3.1	5
67	Improving the Performance of Biomass-Derived Carbons in Li-Ion Batteries by Controlling the Lithium Insertion Process. <i>Journal of the Electrochemical Society</i> , 2010, 157, A791.	2.9	84
68	High-energy, efficient and transparent electrode for lithium batteries. <i>Journal of Materials Chemistry</i> , 2010, 20, 2847.	6.7	23
69	Graphitized Carbons of Variable Morphology and Crystallinity: A Comparative Study of Their Performance in Lithium Cells. <i>Journal of the Electrochemical Society</i> , 2009, 156, A986.	2.9	43
70	A LiFePO ₄ -Based Cell with Li _x (Mg) as Lithium Storage Negative Electrode. <i>Electrochemical and Solid-State Letters</i> , 2009, 12, A145.	2.2	7
71	Suppressing Irreversible Capacity in Low Cost Disordered Carbons for Li-Ion Batteries. <i>Electrochemical and Solid-State Letters</i> , 2009, 12, A167.	2.2	11
72	Nanosized Si/cellulose fiber/carbon composites as high capacity anodes for lithium-ion batteries: A galvanostatic and dilatometric study. <i>Electrochimica Acta</i> , 2009, 54, 6713-6717.	5.2	41

#	ARTICLE	IF	CITATIONS
73	Effect of C and Au additives produced by simple coaters on the surface and the electrochemical properties of nanosized LiFePO ₄ . <i>Journal of Electroanalytical Chemistry</i> , 2009, 631, 29-35.	3.8	33
74	Effect of brief heat-curing on microstructure and mechanical properties in fresh cement based mortars. <i>Cement and Concrete Research</i> , 2009, 39, 573-579.	11.0	19
75	Combining 5V LiNi _{0.5} Mn _{1.5} O ₄ spinel and Si nanoparticles for advanced Li-ion batteries. <i>Electrochemistry Communications</i> , 2009, 11, 1061-1064.	4.7	40
76	A high energy Li-ion battery based on nanosized LiNi _{0.5} Mn _{1.5} O ₄ cathode material. <i>Journal of Power Sources</i> , 2008, 183, 310-315.	7.8	46
77	Electrochemical activity of rock-salt-structured LiFeO ₂ /Li ₄ /3Ti ₂ /3O ₂ nanocomposites in lithium cells. <i>Journal of Nanoparticle Research</i> , 2008, 10, 217-226.	1.9	13
78	Elucidating the Beneficial Effect of Vinylene Carbonate on the Electrochemistry of Antimony Electrodes in Lithium Batteries. <i>ChemPhysChem</i> , 2008, 9, 2610-2617.	2.1	11
79	Polymer-mediated Growth of Highly Crystalline Nano- and Micro-sized LiNi _{0.5} Mn _{1.5} O ₄ Spinel. <i>European Journal of Inorganic Chemistry</i> , 2008, 2008, 3295-3302.	2.0	15
80	A simple route to high performance nanometric metallic materials for Li-ion batteries involving the use of cellulose: The case of Sb. <i>Journal of Power Sources</i> , 2008, 175, 553-557.	7.8	37
81	PMMA-assisted synthesis of Li _{1-x} Ni _{0.5} Mn _{1.5} O ₄ for high-voltage lithium batteries with expanded rate capability at high cycling temperatures. <i>Journal of Power Sources</i> , 2008, 180, 852-858.	7.8	41
82	Insights into the electrochemical activity of nanosized \pm -LiFeO ₂ . <i>Electrochimica Acta</i> , 2008, 53, 6366-6371.	5.2	39
83	Precipitation of CoS vs Ceramic Synthesis for Improved Performance in Lithium Cells. <i>Journal of the Electrochemical Society</i> , 2008, 155, A189.	2.9	38
84	Nano-Si/Cellulose Composites as Anode Materials for Lithium-Ion Batteries. <i>Electrochemical and Solid-State Letters</i> , 2008, 11, A101.	2.2	31
85	Effects of Coating with Gold on the Performance of Nanosized LiNi _{0.5} Mn _{1.5} O ₄ for Lithium Batteries. <i>Journal of the Electrochemical Society</i> , 2007, 154, A178.	2.9	62
86	Nanostructured Cu ₂ O thin film electrodes prepared by electrodeposition for rechargeable lithium batteries. <i>Thin Solid Films</i> , 2007, 515, 5505-5511.	1.8	54
87	Highly electroactive nanosized \pm -LiFeO ₂ . <i>Electrochemistry Communications</i> , 2007, 9, 2116-2120.	4.7	36
88	Use of limestone obtained from waste of the mussel cannery industry for the production of mortars. <i>Cement and Concrete Research</i> , 2007, 37, 559-564.	11.0	80
89	Electrochemical properties of electrodeposited nicked phosphide thin films in lithium cells. <i>Journal of Power Sources</i> , 2007, 171, 870-878.	7.8	41
90	EPD of thick films for their application in lithium batteries. <i>Journal of the European Ceramic Society</i> , 2007, 27, 3823-3827.	5.7	20

#	ARTICLE	IF	CITATIONS
91	Antagonistic effects of copper on the electrochemical performance of LiFePO ₄ . <i>Electrochimica Acta</i> , 2007, 53, 920-926.	5.2	17
92	Relation between the magnetic properties and the crystal and electronic structures of manganese spinels LiNi _{0.5} Mn _{1.5} O ₄ and LiCu _{0.5} Mn _{1.5} O ₄ (0 < x < 0.125). <i>Journal of Applied Physics</i> , 2006, 100, 093908.	2.5	26
93	Beneficial effects of Mo on the electrochemical properties of tin as an anode material for lithium batteries. <i>Electrochimica Acta</i> , 2006, 51, 3391-3398.	5.2	11
94	Positive thin electrodes obtained from hydrothermally synthesized 4BS for lead-acid batteries. <i>Journal of Power Sources</i> , 2006, 157, 579-583.	7.8	7
95	LiNi _{0.5} Mn _{1.5} O ₄ thick-film electrodes prepared by electrophoretic deposition for use in high voltage lithium-ion batteries. <i>Journal of Power Sources</i> , 2006, 158, 583-590.	7.8	42
96	Synthesis and characterization of lead dioxide active material for lead-acid batteries. <i>Journal of Power Sources</i> , 2006, 158, 831-836.	7.8	62
97	Electrochemical properties of LiNi _{0.5} Mn _{1.5} O ₄ films prepared by spin-coating deposition. <i>Journal of Power Sources</i> , 2006, 162, 606-613.	7.8	28
98	Electrochemical reaction of lithium with nanosized vanadium antimonate. <i>Journal of Solid State Chemistry</i> , 2006, 179, 2554-2561.	2.9	21
99	A New and Fast Synthesis of Nanosized LiFePO ₄ Electrode Materials. <i>European Journal of Inorganic Chemistry</i> , 2006, 2006, 1758-1764.	2.0	33
100	Crystallinity Control of a Nanostructured LiNi _{0.5} Mn _{1.5} O ₄ Spinel via Polymer-Assisted Synthesis: A Method for Improving Its Rate Capability and Performance in 5 V Lithium Batteries. <i>Advanced Functional Materials</i> , 2006, 16, 1904-1912.	14.9	217
101	A First-Principles Investigation of the Role Played by Oxygen Deficiency in the Electrochemical Properties of LiCu _{0.5} Mn _{1.5} O ₄ Spinel. <i>Journal of the Electrochemical Society</i> , 2006, 153, A2098.	2.9	15
102	Nanocrystalline materials obtained by using a simple, rapid method for rechargeable lithium batteries. <i>Journal of Power Sources</i> , 2005, 150, 192-201.	7.8	39
103	Use of low-temperature nanostructured CuO thin films deposited by spray-pyrolysis in lithium cells. <i>Thin Solid Films</i> , 2005, 474, 133-140.	1.8	212
104	Expanding the Rate Capabilities of the LiNi _{0.5} Mn _{1.5} O ₄ Spinel by Exploiting the Synergistic Effect Between Nano and Microparticles. <i>Electrochemical and Solid-State Letters</i> , 2005, 8, A641.	2.2	40
105	Oxygen Lattice Instability as a Capacity Fading Mechanism for 5 V Cathode Materials. <i>Journal of the Electrochemical Society</i> , 2005, 152, A6.	2.9	27
106	Electrodeposition of Cu ₂ O: An Excellent Method for Obtaining Films of Controlled Morphology and Good Performance in Li-Ion Batteries. <i>Electrochemical and Solid-State Letters</i> , 2005, 8, A159.	2.2	102
107	Oxygen Deficiency as the Origin of the Disparate Behavior of LiM _{0.5} Mn _{1.5} O ₄ (M = Ni, Cu) Nanospinel in Lithium Cells. <i>Journal of the Electrochemical Society</i> , 2005, 152, A552.	2.9	50
108	Adverse Effect of Ag Treatment on the Electrochemical Performance of the 5 V Nanometric Spinel LiNi _{0.5} Mn _{1.5} O ₄ in Lithium Cells. <i>Electrochemical and Solid-State Letters</i> , 2005, 8, A303.	2.2	24

#	ARTICLE	IF	CITATIONS
109	Synthesis, Characterization, and Electrochemical Properties of Nanocrystalline Silver Thin Films Obtained by Spray Pyrolysis. <i>Journal of the Electrochemical Society</i> , 2004, 151, A151.	2.9	51
110	Influence of the mechanical treatment on the structure and the thermal stability of alkaline-earth carbonates. <i>Journal of Materials Science</i> , 2004, 39, 5189-5193.	3.7	15
111	Reaction of SbPO ₄ with lithium in non-aqueous electrochemical cells: preliminary study and evaluation of its electrochemical performance in anodes for lithium ion batteries. <i>Journal of Solid State Chemistry</i> , 2004, 177, 2920-2927.	2.9	24
112	Nanostructured CuO thin film electrodes prepared by spray pyrolysis: a simple method for enhancing the electrochemical performance of CuO in lithium cells. <i>Electrochimica Acta</i> , 2004, 49, 4589-4597.	5.2	189
113	Thin electrodes based on rolled Pb-Sn-Ca grids for VRLA batteries. <i>Journal of Power Sources</i> , 2004, 125, 246-255.	7.8	3
114	Enhancing the electrochemical properties of LT-LiCoO ₂ in lithium cells by doping with Mn. <i>Journal of Power Sources</i> , 2004, 128, 286-291.	7.8	26
115	Ion-exchange properties of P2-NaxMnO ₂ : evidence of the retention of the layer structure based on chemical reactivity data and electrochemical measurements of lithium cells. <i>Journal of Solid State Chemistry</i> , 2003, 174, 365-371.	2.9	12
116	XRD, XPS and Sn NMR study of tin sulfides obtained by using chemical vapor transport methods. <i>Journal of Solid State Chemistry</i> , 2003, 175, 359-365.	2.9	84
117	Preparation and characterization of thin electrodes for lead-acid batteries. <i>Journal of Power Sources</i> , 2003, 113, 376-381.	7.8	6
118	Development of high power VRLA batteries using novel materials and processes. <i>Journal of Power Sources</i> , 2003, 116, 61-72.	7.8	9
119	Mechanochemical synthesis of Sn _{1-x} MoxO ₂ anode materials for Li-ion batteries. <i>Journal of Materials Chemistry</i> , 2002, 12, 2979-2984.	6.7	48
120	Synthesis, characterization and comparative study of the electrochemical properties of doped lithium manganese spinels as cathodes for high voltage lithium batteries. <i>Journal of Materials Chemistry</i> , 2002, 12, 734-741.	6.7	35
121	Synthesis of Li _x MnO _y ·nH ₂ O birnessite oxide by the hydrothermal method. <i>Materials Letters</i> , 2002, 56, 653-659.	2.6	12
122	Synthesis and characterization of high-temperature hexagonal P2-Na _{0.6} MnO ₂ and its electrochemical behaviour as cathode in sodium cells. <i>Journal of Materials Chemistry</i> , 2002, 12, 1142-1147.	6.7	330
123	Spray pyrolysis as a method for preparing PbO coatings amenable to use in lead-acid batteries. <i>Journal of Power Sources</i> , 2002, 108, 35-40.	7.8	30
124	Study of in situ adsorption and intercalation of cobaltocene into SnS ₂ single crystals by photoelectron spectroscopy. <i>Surface Science</i> , 2001, 477, L295-L300.	1.9	5
125	Antimony doping effect on the electrochemical behavior of SnO ₂ thin film electrodes. <i>Journal of Power Sources</i> , 2001, 97-98, 232-234.	7.8	72
126	Electrochemical properties of lead oxide films obtained by spray pyrolysis as negative electrodes for lithium secondary batteries. <i>Electrochimica Acta</i> , 2001, 46, 2939-2948.	5.2	109

#	ARTICLE	IF	CITATIONS
127	Synthesis and Characterization of Diamine Intercalation Compounds of SnS ₂ Single Crystals. Journal of Solid State Chemistry, 2000, 150, 391-398.	2.9	15
128	Cation-deficient Mo _{1-x} Sn _x O ₂ oxides as anodes for lithium ion batteries. Electrochimica Acta, 2000, 46, 83-89.	5.2	18
129	Influence of Al, In, Cu, Fe and Sn dopants in the microstructure of zinc oxide thin films obtained by spray pyrolysis. Thin Solid Films, 2000, 366, 16-27.	1.8	131
130	Influence of Al, In, Cu, Fe and Sn dopants on the response of thin film ZnO gas sensor to ethanol vapour. Thin Solid Films, 2000, 373, 137-140.	1.8	288
131	Use of amorphous tin-oxide films obtained by spray pyrolysis as electrodes in lithium batteries. Journal of Power Sources, 2000, 87, 106-111.	7.8	44
132	X-ray Diffraction, XPS, and Magnetic Properties of Lanthanide-Based Misfit-Layered Sulfides Intercalated with Cobaltocene. Chemistry of Materials, 2000, 12, 3792-3797.	6.7	13
133	Electrochemical Cointercalation of Propylene Carbonate with Alkali Metals in SnS ₂ Single Crystals. Journal of the Electrochemical Society, 1999, 146, 657-662.	2.9	11
134	Electrochemical behaviour of SnO ₂ doped with boron and indium in anodes for lithium secondary batteries. Solid State Ionics, 1999, 126, 219-226.	2.7	29
135	Improving the Electrochemical Performance of SnO ₂ Cathodes in Lithium Secondary Batteries by Doping with Mo. Journal of the Electrochemical Society, 1999, 146, 1640-1642.	2.9	54
136	Ultra-high vacuum deposition of Na on SnS ₂ in single crystal and powder forms: evidence of a decomposition reaction. Surface Science, 1999, 426, 259-267.	1.9	8
137	Synthesis and Characterization of Poly(ethylene Oxide) Nanocomposites of Misfit Layer Chalcogenides. Journal of Solid State Chemistry, 1998, 141, 323-329.	2.9	15
138	Sodium Intercalation into (PbS) _{1.18} (TiS ₂) ₂ Misfit Layer Compound. Journal of Solid State Chemistry, 1996, 124, 238-243.	2.9	7
139	Microstructure and intercalation properties of petrol cokes obtained at 1400°C. Materials Science and Engineering B: Solid-State Materials for Advanced Technology, 1996, 39, 216-223.	3.5	13
140	New tin-containing spinel sulfide electrodes for ambient temperature rocking chair cells. Journal of Power Sources, 1996, 62, 101-105.	7.8	27
141	Raman study and lattice dynamics calculation of the misfit layer compound (PbS) _{1.12} VS ₂ . Journal of Raman Spectroscopy, 1995, 26, 675-681.	2.5	4
142	¹²⁵ Te Mössbauer spectroscopic study of layered transition metal ditellurides with interlayer communication. Solid State Communications, 1995, 96, 911-914.	1.9	1
143	Chemically deintercalated cathode materials for lithium cells. Ionics, 1995, 1, 246-250.	2.4	2
144	Structure and Electrochemical Properties of Li _{1-x} Co _{1-x} Ni _y Co _{1-x-y} O ₂ at 0°C. Journal of the Electrochemical Society, 1995, 142, 3997-4005.	2.9	51

#	ARTICLE	IF	CITATIONS
145	Diffraction and XPS Studies of Misfit Layer Chalcogenides Intercalated with Cobaltocene. Chemistry of Materials, 1995, 7, 1576-1582.	6.7	50
146	Electrochemical lithium insertion into $\text{In}_{16}\text{Sn}_4\text{S}_{32}$ and $\text{Cu}_4\text{In}_{20}\text{S}_{32}$ spinel sulphides. Journal of Alloys and Compounds, 1995, 217, 176-180.	5.5	19
147	Acid-Delithiated $\text{Li}_{1-x}(\text{Ni}_y\text{Co}_{1-y})_{1+x}\text{O}_2$ as Insertion Electrodes in Lithium Batteries. Journal of Solid State Chemistry, 1994, 113, 182-192.	2.9	25
148	Cobaltocene intercalation into misfit layer chalcogenides. Journal of the Chemical Society Chemical Communications, 1994, , 1081-1082.	2.0	5
149	Metalâ€”support interaction effects in the liquid-phase selective hydrogenation of 1,4-butyndiol with nickel catalysts supported on AlPO_4 and on other conventional non-reducible compounds. Journal of Molecular Catalysis, 1993, 85, 305-325.	1.2	18
150	Thermal behaviour of chemically deintercalated $\text{Li}_{1-x}\text{Ni}_x\text{O}_2$. Journal of Thermal Analysis, 1992, 38, 295-301.	0.6	17
151	Chromium substitution and crystallinity changes in $\gamma\text{-FeOOH}$. Journal of Materials Science, 1990, 25, 1813-1815.	3.7	5
152	Mn and Co substitution in $\gamma\text{-FeOOH}$ and its decomposition products. Journal of Materials Science, 1990, 25, 5207-5214.	3.7	10
153	Relationships between the surface properties of $\hat{\gamma}\text{-Fe}_2\text{O}_3$ and its cobalt-modified products. Journal of Colloid and Interface Science, 1990, 138, 565-579.	9.4	6
154	Preferential X-ray line Broadening and Thermal Behavior of $\gamma\text{-Fe}_2\text{O}_3$. Journal of the American Ceramic Society, 1989, 72, 1244-1246.	3.8	13
155	Effect of preliminary mechanical activation on the behaviour of orthorhombic lead dioxide. Journal of Thermal Analysis, 1988, 34, 1421-1425.	0.6	2
156	Textural evolution of $\hat{\gamma}\text{-Fe}_2\text{O}_3$ obtained by thermal and mechanochemical decomposition of $\hat{\gamma}\text{-FeOOH}$. Journal of Colloid and Interface Science, 1988, 122, 507-513.	9.4	7
157	Synthesis and alteration of $\gamma\text{-LiFeO}_2$ by mechanochemical processes. Journal of Materials Science, 1988, 23, 2971-2974.	3.7	9
158	Tk and DTA stkdies of lithikm insertion transition metal compoknds. Materials Chemistry and Physics, 1988, 20, 145-152.	4.0	7
159	Mechanochemical transformation of $\hat{\gamma}\text{-FeOOH}$ into $\hat{\gamma}\text{-Fe}_2\text{O}_3$ in the presence of Li_2CO_3 . Journal of Materials Science Letters, 1987, 6, 223-224.	0.5	2
160	Limitations in the use of X-ray crystallite size in the determination of surface area in Co_3O_4 . Journal of Colloid and Interface Science, 1987, 115, 274-276.	9.4	5
161	Texture, crystallinity, and catalytic properties of Co_3O_4 obtained by thermal and mechanical treatments. Journal of Colloid and Interface Science, 1986, 110, 172-180.	9.4	13
162	Changes in the kinetics of the vaterite-calcite transformation with temperature and sample crystallinity. Journal of Materials Science, 1986, 21, 947-952.	3.7	8

#	ARTICLE	IF	CITATIONS
163	X-ray line broadening in haematite derived from $\hat{\text{I}}\text{-FeOOH}$ by thermal and mechanical procedures. <i>Journal of Materials Science Letters</i> , 1986, 5, 1295-1297.	0.5	13
164	Changes in crystallinity and thermal effects in ground vaterite. <i>Journal of Materials Science</i> , 1985, 20, 941-946.	3.7	11
165	Effect of grinding on the kinetics of the transformation vaterite-calcite. <i>Thermochimica Acta</i> , 1985, 92, 211-214.	2.7	1
166	Thermal behaviour of synthetic akaganeite under different experimental conditions. <i>Thermochimica Acta</i> , 1985, 92, 525-528.	2.7	8
167	Textural evolution of synthetic $\hat{\text{I}}^3\text{-FeOOH}$ during thermal treatment by differential scanning calorimetry. <i>Journal of Colloid and Interface Science</i> , 1984, 101, 392-400.	9.4	39
168	Kinetic study of the thermal decomposition of cobalt(III) oxyhydroxide. <i>Journal of Thermal Analysis</i> , 1984, 29, 479-489.	0.6	11
169	Kinetic study of the thermal decomposition of cobalt(III) oxyhydroxide. <i>Journal of Thermal Analysis</i> , 1984, 29, 491-501.	0.6	7
170	Critical examination of the DTA techniques used to study the kinetics of solid-state transformations. <i>Journal of Materials Science</i> , 1983, 18, 2117-2125.	3.7	8
171	The applicability of DTA and DSC techniques to the study of the kinetics of phase transition reactions. <i>Journal of Thermal Analysis</i> , 1982, 24, 23-34.	0.6	10
172	Grinding-induced structural transformations in CaCO_3 . <i>Journal of Colloid and Interface Science</i> , 1981, 81, 500-510.	9.4	22
173	Alteration of kinetics and thermodynamics of thermal decomposition of alkaline-earth carbonates induced by grinding. <i>Thermochimica Acta</i> , 1979, 32, 99-110.	2.7	21
174	Some observations on the thermal behaviour of lead monoxide, PbO . <i>Thermochimica Acta</i> , 1979, 33, 395-400.	2.7	1
175	On the estimation of the reaction mechanisms of thermal decomposition of solids from the fraction reacted at the maximum reaction rate. <i>Thermochimica Acta</i> , 1978, 25, 257-260.	2.7	11
176	Determination of equilibrium constants of solid thermal decomposition reactions by thermogravimetry. <i>Thermochimica Acta</i> , 1978, 23, 257-261.	2.7	1
177	Computer kinetic analysis of simultaneously obtained TG and DTG curves. <i>Journal of Theoretical Biology</i> , 1978, 14, 221-228.	1.7	58
178	Thermal decomposition reactions of solids controlled by diffusion and phase-boundary processes: possible misinterpretation of the mechanism from thermogravimetric data. <i>Thermochimica Acta</i> , 1977, 19, 305-317.	2.7	95
179	On the thermal decomposition mechanism for dehydroxylation of alkaline-earth hydroxides. <i>Journal of Theoretical Biology</i> , 1976, 10, 103-110.	1.7	18
180	Defects of thermogravimetric analysis for discerning between first order reactions and those taking place through the Avrami-Erofeev's mechanism. <i>Thermochimica Acta</i> , 1976, 16, 382-387.	2.7	84

#	ARTICLE	IF	CITATIONS
181	Application of programmed temperature decomposition to the study of solid decomposition reactions taking place through the prout and tompkins mechanism. <i>Thermochimica Acta</i> , 1975, 12, 337-342.	2.7	20