Caroline M Jonsson

List of Publications by Year in descending order

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#	Article	IF	CITATIONS
1	Influence of humic acid and dihydroxy benzoic acid on the agglomeration, adsorption, sedimentation and dissolution of copper, manganese, aluminum and silica nanoparticles – A tentative exposure scenario. PLoS ONE, 2018, 13, e0192553.	2.5	26
2	Influence of organic molecules on the aggregation of TiO2 nanoparticles in acidic conditions. Journal of Nanoparticle Research, 2017, 19, 133.	1.9	18
3	Potentiometric Study of Dissociation Constants of Dihydroxybenzoic Acids at Reduced Ionic Strengths and Temperatures. American Journal of Analytical Chemistry, 2017, 08, 142-150.	0.9	9
4	Evaluating Glutamate and Aspartate Binding Mechanisms to Rutile (α-TiO ₂) via ATR-FTIR Spectroscopy and Quantum Chemical Calculations. Langmuir, 2011, 27, 1778-1787.	3.5	65
5	Adsorption and Surface Complexation Study of L-DOPA on Rutile (α-TiO ₂) in NaCl Solutions. Environmental Science & Technology, 2011, 45, 3959-3966.	10.0	49
6	The adsorption of short single-stranded DNA oligomers to mineral surfaces. Chemosphere, 2011, 83, 1560-1567.	8.2	60
7	Adsorption of Nucleic Acid Components on Rutile (TiO ₂) Surfaces. Astrobiology, 2010, 10, 311-323.	3.0	64
8	Adsorption of l-aspartate to rutile (α-TiO2): Experimental and theoretical surface complexation studies. Geochimica Et Cosmochimica Acta, 2010, 74, 2356-2367.	3.9	53
9	Glyphosate complexation to aluminium(III). An equilibrium and structural study in solution using potentiometry, multinuclear NMR, ATR–FTIR, ESI-MS and DFT calculations. Journal of Inorganic Biochemistry, 2009, 103, 1426-1438.	3.5	27
10	Attachment of <scp>l</scp> -Glutamate to Rutile (α-TiO ₂): A Potentiometric, Adsorption, and Surface Complexation Study. Langmuir, 2009, 25, 12127-12135.	3.5	72
11	Adsorption of Glyphosate on Goethite (α-FeOOH): Surface Complexation Modeling Combining Spectroscopic and Adsorption Data. Environmental Science & Technology, 2008, 42, 2464-2469.	10.0	50
12	Glutamate Surface Speciation on Amorphous Titanium Dioxide and Hydrous Ferric Oxide. Environmental Science & Technology, 2008, 42, 6034-6039.	10.0	39