

Margherita Silvetti

List of Publications by Year in descending order

Source: <https://exaly.com/author-pdf/213350/publications.pdf>

Version: 2024-02-01

21
papers

907
citations

471509

17
h-index

713466

21
g-index

21
all docs

21
docs citations

21
times ranked

921
citing authors

#	ARTICLE	IF	CITATIONS
1	Municipal solid wastes as a resource for environmental recovery: Impact of water treatment residuals and compost on the microbial and biochemical features of As and trace metal-polluted soils. <i>Ecotoxicology and Environmental Safety</i> , 2019, 174, 445-454.	6.0	39
2	Municipal solid waste compost as a novel sorbent for antimony(V): adsorption and release trials at acidic pH. <i>Environmental Science and Pollution Research</i> , 2018, 25, 5603-5615.	5.3	33
3	Mutual effect of <i>Phragmites australis</i> , <i>Arundo donax</i> and immobilization agents on arsenic and trace metals phytostabilization in polluted soils. <i>Geoderma</i> , 2018, 314, 63-72.	5.1	37
4	Interaction of the water soluble fraction of MSW-composts with Pb(II) and Cu(II) ions. <i>Journal of Environmental Management</i> , 2017, 192, 39-47.	7.8	9
5	Use of municipal solid wastes for chemical and microbiological recovery of soils contaminated with metal(loid)s. <i>Soil Biology and Biochemistry</i> , 2017, 111, 25-35.	8.8	47
6	Sorption of Pb, Cu, Cd, and Zn by Municipal Solid Waste Composts: Metal Retention and Desorption Mechanisms. <i>Clean - Soil, Air, Water</i> , 2017, 45, .	1.1	13
7	Influence of lead in the sorption of arsenate by municipal solid waste composts: metal(loid) retention, desorption and phytotoxicity. <i>Bioresource Technology</i> , 2017, 225, 90-98.	9.6	25
8	Influence of iron-rich water treatment residues and compost on the mobility of metal(loid)s in mine soils. <i>Geoderma</i> , 2016, 283, 1-9.	5.1	31
9	Sorption of Cadmium(II) and Zinc(II) from Aqueous Solution by Water Treatment Residuals at Different pH Values. <i>Water, Air, and Soil Pollution</i> , 2015, 226, 1.	2.4	18
10	Copper(II) and lead(II) removal from aqueous solution by water treatment residues. <i>Journal of Hazardous Materials</i> , 2015, 283, 140-147.	12.4	83
11	Water treatment residues as accumulators of oxoanions in soil. Sorption of arsenate and phosphate anions from an aqueous solution. <i>Journal of Hazardous Materials</i> , 2014, 264, 144-152.	12.4	44
12	Stabilising metal(loid)s in soil with iron and aluminium-based products: Microbial, biochemical and plant growth impact. <i>Journal of Environmental Management</i> , 2014, 139, 146-153.	7.8	60
13	Leachability, bioaccessibility and plant availability of trace elements in contaminated soils treated with industrial by-products and subjected to oxidative/reductive conditions. <i>Geoderma</i> , 2014, 214-215, 204-212.	5.1	41
14	Arsenic Mobilization by Citrate and Malate from a Red Mud-Treated Contaminated Soil. <i>Journal of Environmental Quality</i> , 2013, 42, 774-781.	2.0	19
15	Study of the Interaction Mechanism in the Biosorption of Copper(II) Ions onto <i>Posidonia oceanica</i> and Peat. <i>Clean - Soil, Air, Water</i> , 2012, 40, 428-437.	1.1	26
16	X-ray Diffraction and Thermal Analysis of Bauxite Ore-Processing Waste (Red Mud) Exchanged with Arsenate and Phosphate. <i>Clays and Clay Minerals</i> , 2011, 59, 189-199.	1.3	12
17	Long-term influence of red mud on As mobility and soil physico-chemical and microbial parameters in a polluted sub-acidic soil. <i>Journal of Hazardous Materials</i> , 2011, 185, 1241-1248.	12.4	77
18	Study of sorption processes and FT-IR analysis of arsenate sorbed onto red muds (a bauxite ore) Tj ETQqO 0 0 rgBT (Overlock, 10 Tf 50 62	12.4	74

#	ARTICLE	IF	CITATIONS
19	Influence of the pH on the accumulation of phosphate by red mud (a bauxite ore processing waste). <i>Journal of Hazardous Materials</i> , 2010, 182, 266-272.	12.4	42
20	Influence of pea and wheat growth on Pb, Cd, and Zn mobility and soil biological status in a polluted amended soil. <i>Geoderma</i> , 2009, 151, 241-248.	5.1	81
21	XRD, FTIR, and thermal analysis of bauxite ore-processing waste (red mud) exchanged with heavy metals. <i>Clays and Clay Minerals</i> , 2008, 56, 461-469.	1.3	96