

Charles A Geiger

List of Publications by Year in descending order

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#	ARTICLE	IF	CITATIONS
1	Are the thermodynamic properties of natural and synthetic $Mg_2SiO_4Fe_2SiO_4$ olivines the same?. American Mineralogist, 2021, 106, 317-321.	1.9	2
2	Micro- and nano-size hydrogarnet clusters in calcium silicate garnet: Part II. Mineralogical, petrological, and geochemical aspects. American Mineralogist, 2020, 105, 468-478.	1.9	9
3	Micro- and nano-size hydrogrossular-like clusters in pyrope crystals from ultra-high-pressure rocks of the Dora-Maira Massif, western Alps. Contributions To Mineralogy and Petrology, 2020, 175, 1.	3.1	3
4	Micro- and nano-size hydrogarnet clusters and proton ordering in calcium silicate garnet: Part I. The quest to understand the nature of "water" in garnet continues. American Mineralogist, 2020, 105, 455-467.	1.9	15
5	An experimental calorimetric and a DFT+G+U study of the thermodynamic properties of $\text{Cs}_8[(\text{UO}_2)_4(\text{WO}_4)_4(\text{WO}_5)_2]$. Journal of Chemical Thermodynamics, 2019, 139, 105873.	2.0	29
6	An analysis of the magnetic behavior of olivine and garnet substitutional solid solutions. American Mineralogist, 2019, 104, 1246-1255.	1.9	5
7	A calorimetric and thermodynamic investigation of cesium uranyl tungstate $\text{Cs}_8[(\text{UO}_2)_4(\text{WO}_4)_4(\text{WO}_5)_2]$. Journal of Chemical Thermodynamics, 2019, 137, 48-55.	2.0	4
8	Thermodynamic behaviour of grossular-andradite, $\text{Ca}_3(\text{Al}_x\text{Fe}_{3+1-x})_2\text{Si}_3\text{O}_{12}$, garnets: a calorimetric study. European Journal of Mineralogy, 2019, 31, 443-451.	1.3	7
9	Recent developments and the future of low-T calorimetric investigations in the Earth sciences: Consequences for thermodynamic calculations and databases. Journal of Metamorphic Geology, 2018, 36, 283-295.	3.4	12
10	IR spectroscopy and OH in silicate garnet: The long quest to document the hydrogarnet substitution. American Mineralogist, 2018, 103, 384-393.	1.9	33
11	Heat capacity and entropy behavior of andradite: a multi-sample and "methodological investigation. European Journal of Mineralogy, 2018, 30, 681-694.	1.3	8
12	Nitrogen and carbon concentrations and isotopic compositions of the silica clathrate melanophlogite. American Mineralogist, 2017, 102, 686-689.	1.9	6
13	A calorimetric investigation of $\text{A}_2[(\text{UO}_2)_2(\text{WO}_5)\text{O}]$ compounds with $\text{A} = \text{K}, \text{Rb}$ and Cs and calculated phase relations in the $\text{K}_2\text{WO}_4\text{-}\text{UO}_3\text{-H}_2\text{O}$ and $\text{K}_2\text{MoO}_4\text{-K}_2\text{WO}_4\text{-H}_2\text{O}$ systems. Journal of Chemical Thermodynamics, 2017, 112, 23-30.	2.0	3
14	Trivalent transition-metal cations and local structure in pyrope- and grossular-rich solid solutions investigated by ^{27}Al and ^{29}Si MAS NMR spectroscopy. European Journal of Mineralogy, 2016, 28, 179-187.	1.3	3
15	A tale of two garnets: The role of solid solution in the development toward a modern mineralogy. American Mineralogist, 2016, 101, 1735-1749.	1.9	19
16	Pathways for nitrogen cycling in Earth's crust and upper mantle: A review and new results for microporous beryl and cordierite. American Mineralogist, 2016, 101, 7-24.	1.9	35
17	Cation order-disorder in Fe-bearing pyrope and grossular garnets: A ^{27}Al and ^{29}Si MAS NMR and ^{57}Fe Mossbauer spectroscopy study. American Mineralogist, 2015, 100, 536-547.	1.9	25
18	A calorimetric and thermodynamic investigation of $\text{A}_2[(\text{UO}_2)_2(\text{MoO}_4)\text{O}_2]$ compounds with $\text{A}=\text{K}$ and Rb and calculated phase relations in the system $(\text{K}_2\text{MoO}_4+\text{UO}_3+\text{H}_2\text{O})$. Journal of Chemical Thermodynamics, 2015, 90, 270-276.	2.0	4

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19	Thermodynamic mixing properties and behavior of almandine–spessartine solid solutions. <i>Geochimica Et Cosmochimica Acta</i> , 2014, 125, 210-224.		3.9	10
20	DFT Study of the Role of Al ³⁺ in the Fast Ion-Conductor Li ₇ Al ₃ La ₃ Zr ₂ O ₁₂ Garnet. <i>Chemistry of Materials</i> , 2014, 26, 2617-2623.			108
21	A Synthesis and Crystal Chemical Study of the Fast Ion Conductor Li ₇ Ca ₃ La ₃ Zr ₂ O ₁₂ with $x = 0.08$ to 0.84 . <i>Inorganic Chemistry</i> , 2014, 53, 6264-6269.		4.0	93
22	Thermodynamic mixing properties and behavior of grossular–spessartine, $(\text{Ca Mn}^{1+})_3\text{Al}_2\text{Si}_3\text{O}_{12}$, solid solutions. <i>Geochimica Et Cosmochimica Acta</i> , 2014, 141, 294-302.		3.9	7
23	Thermodynamic properties and behaviour of $\text{A}_2[(\text{UO}_2)(\text{MoO}_4)_2]$ compounds with $\text{A}=\text{Li}, \text{Na}, \text{K}, \text{Rb}$, and Cs . <i>Journal of Chemical Thermodynamics</i> , 2014, 79, 205-214.		2.0	7
24	Synthesis and Crystal Chemistry of the Fast Li-Ion Conductor Li ₇ La ₃ Zr ₂ O ₁₂ Doped with Fe. <i>Inorganic Chemistry</i> , 2013, 52, 8005-8009.		4.0	71
25	A calorimetric and thermodynamic investigation of potassium uranyl tungstate $\text{K}_2[(\text{UO}_2)(\text{W}_2\text{O}_8)]$. <i>Journal of Chemical Thermodynamics</i> , 2013, 57, 430-435.		2.0	7
26	Static disorders of atoms and experimental determination of Debye temperature in pyrope: Low- and high-temperature single-crystal X-ray diffraction study--Discussion. <i>American Mineralogist</i> , 2013, 98, 780-782.		1.9	7
27	Almandine: Lattice and non-lattice heat capacity behavior and standard thermodynamic properties. <i>American Mineralogist</i> , 2012, 97, 1771-1782.		1.9	25
28	Thermodynamic behavior and properties of katoite (hydrogrossular): A calorimetric study. <i>American Mineralogist</i> , 2012, 97, 1252-1255.		1.9	17
29	Grossular: A crystal-chemical, calorimetric, and thermodynamic study. <i>American Mineralogist</i> , 2012, 97, 1299-1313.		1.9	22
30	A low-temperature IR spectroscopic investigation of the H ₂ O molecules in the zeolite mesolite. <i>European Journal of Mineralogy</i> , 2012, 24, 439-445.		1.3	5
31	A 57Fe Mössbauer spectroscopic study of sogdianite: an example of a symmetric electric field gradient around Fe ³⁺ . <i>Physics and Chemistry of Minerals</i> , 2012, 39, 73-78.		0.8	2
32	Crystal Chemistry and Stability of Li ₇ La ₃ Zr ₂ O ₁₂ Garnet: A Fast Lithium-Ion Conductor. <i>Inorganic Chemistry</i> , 2011, 50, 1089-1097.		4.0	600
33	A calorimetric and thermodynamic investigation of uranyl molybdate UO ₂ MoO ₄ . <i>Journal of Chemical Thermodynamics</i> , 2010, 42, 873-878.		2.0	19
34	Molecular H ₂ O in armenite, BaCa ₂ Al ₆ Si ₉ O ₃₀ ·2H ₂ O, and epididymite, Na ₂ Be ₂ Si ₆ O ₁₅ ·H ₂ O: Heat capacity, entropy and local-bonding behavior of confined H ₂ O in microporous silicates. <i>Geochimica Et Cosmochimica Acta</i> , 2010, 74, 5202-5215.		3.9	10
35	Heat-capacity behaviour of hemimorphite, Zn ₄ Si ₂ O ₇ (OH)2H ₂ O, and its dehydrated analogue Zn ₄ Si ₂ O ₇ (OH)2: a calorimetric and thermodynamic investigation of their phase transitions. <i>European Journal of Mineralogy</i> , 2009, 21, 971-983.		1.3	11
36	Quasi-ice-like CP behavior of molecular H ₂ O in hemimorphite Zn ₄ Si ₂ O ₇ (OH)2·H ₂ O: CP and entropy of confined H ₂ O in microporous silicates. <i>American Mineralogist</i> , 2009, 94, 634-637.		1.9	12

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37	A calorimetric investigation of spessartine: Vibrational and magnetic heat capacity. <i>Geochimica Et Cosmochimica Acta</i> , 2009, 73, 3393-3409.	3.9	22
38	A crystal-chemical investigation of clinzozoisite synthesized along the join Ca ₂ Al ₃ Si ₃ O ₁₂ (OH)-Ca ₂ Al ₂ CrSi ₃ O ₁₂ (OH). <i>American Mineralogist</i> , 2009, 94, 1351-1360.	1.9	12
39	Cation order/disorder behavior and crystal chemistry of pyrope-grossular garnets: An ¹⁷ O 3QMAS and ²⁷ Al MAS NMR spectroscopic study. <i>American Mineralogist</i> , 2008, 93, 134-143.	1.9	22
40	Silicate garnet: A micro to macroscopic (re)view. <i>American Mineralogist</i> , 2008, 93, 360-372.	1.9	45
41	Low-temperature heat capacity of synthetic Fe- and Mg-cordierite: thermodynamic properties and phase relations in the system FeO-Al ₂ O ₃ -SiO ₂ -(H ₂ O). <i>European Journal of Mineralogy</i> , 2008, 20, 47-62.	1.3	10
42	Crystal chemistry of macfallite: Relationships to sursassite and pumpellyite. <i>American Mineralogist</i> , 2008, 93, 1851-1857.	1.9	10
43	Heat capacity and entropy of melanophlogite: Molecule-containing porosils in nature. <i>American Mineralogist</i> , 2008, 93, 1179-1182.	1.9	13
44	Heat capacity of synthetic hydrous Mg-cordierite at low temperatures: Thermodynamic properties and the behavior of the H ₂ O molecule in selected hydrous micro and nanoporous silicates. <i>American Mineralogist</i> , 2007, 92, 388-396.	1.9	19
45	Entropies of mixing and subsolidus phase relations of forsterite-fayalite (Mg ₂ SiO ₄ -Fe ₂ SiO ₄) solid solution. <i>American Mineralogist</i> , 2007, 92, 699-702.	1.9	8
46	A low-temperature calorimetric study of synthetic (forsterite+fayalite) {(Mg ₂ SiO ₄ +Fe ₂ SiO ₄)} solid solutions: An analysis of vibrational, magnetic, and electronic contributions to the molar heat capacity and entropy of mixing. <i>Journal of Chemical Thermodynamics</i> , 2007, 39, 906-933.	2.0	57
47	Heat capacities and entropies of mixing of pyrope-grossular (Mg ₃ Al ₂ Si ₃ O ₁₂ -Ca ₃ Al ₂ Si ₃ O ₁₂) garnet solid solutions: A low-temperature calorimetric and a thermodynamic investigation. <i>American Mineralogist</i> , 2006, 91, 894-906.	1.9	77
48	A low-temperature heat-capacity study of synthetic anhydrous Mg-cordierite (Mg ₂ Al ₄ Si ₅ O ₁₈). <i>American Mineralogist</i> , 2006, 91, 35-38.	1.9	25
49	The vibrational spectrum of synthetic hydrogrossular (katoite) Ca ₃ Al ₂ (O ₄ H ₄) ₃ : A low-temperature IR and Raman spectroscopic study. <i>American Mineralogist</i> , 2005, 90, 1335-1341.	1.9	46
50	Microscopic strain in synthetic pyrope-grossular solid solutions determined by synchrotron X-ray powder diffraction at 5 K: The relationship to enthalpy of mixing behavior. <i>American Mineralogist</i> , 2005, 90, 506-509.	1.9	28
51	Molecules in the SiO ₂ -clathrate melanophlogite: A single-crystal Raman study. <i>American Mineralogist</i> , 2004, 88, 1364-1368.	1.9	33
52	Fe ²⁺ -O and Mn ²⁺ -O bonding and Fe ²⁺ - and Mn ²⁺ -vibrational properties in synthetic almandine-spessartine solid solutions: an X-ray absorption fine structure study. <i>European Journal of Mineralogy</i> , 2004, 16, 801-808.	1.3	8
53	Cordierite IV: structural heterogeneity and energetics of Mg?Fe solid solutions. <i>Contributions To Mineralogy and Petrology</i> , 2003, 145, 752-764.	3.1	14
54	Ti(III) in synthetic pyrope: A single-crystal electron paramagnetic resonance study. <i>European Journal of Mineralogy</i> , 2003, 15, 697-699.	1.3	9

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55	The crystal structures of grossular and spessartine between 100 and 600 K and the crystal chemistry of grossular-spessartine solid solutions. <i>American Mineralogist</i> , 2002, 87, 542-549.	1.9	35
56	The dynamic properties of zircon studied by single-crystal X-ray diffraction and Raman spectroscopy. <i>European Journal of Mineralogy</i> , 2001, 13, 939-948.	1.3	79
57	Cordierite I: The coordination of Fe ²⁺ . <i>American Mineralogist</i> , 2000, 85, 1255-1264.	1.9	43
58	Volumes of mixing in aluminosilicate garnets: Solid solution and strain behavior. <i>American Mineralogist</i> , 2000, 85, 893-897.	1.9	22
59	Cordierite III: the site occupation and concentration of Fe ³⁺ . <i>Contributions To Mineralogy and Petrology</i> , 2000, 140, 344-352.	3.1	30
60	Single-crystal IR- and UV/VIS-spectroscopic measurements on transition-metal-bearing pyrope: the incorporation of hydroxide in garnet. <i>European Journal of Mineralogy</i> , 2000, 12, 259-271.	1.3	36
61	Local Ca-Mg distribution of Mg-rich pyrope-grossular garnets synthesized at different temperatures revealed by ²⁹ Si MAS NMR spectroscopy. <i>American Mineralogist</i> , 1999, 84, 1422-1432.	1.9	35
62	Raspberry-red grossular from Sierra de Cruces Range, Coahuila, Mexico. <i>European Journal of Mineralogy</i> , 1999, 11, 1109-1114.	1.3	15
63	Mn ₃ Al ₂ Si ₃ O ₁₂ spessartine and Ca ₃ Al ₂ Si ₃ O ₁₂ grossular garnet; structural dynamic and thermodynamic properties. <i>American Mineralogist</i> , 1997, 82, 740-747.	1.9	88
64	Molar volumes of mixing of almandine-pyrope and almandine-spessartine garnets and the crystal chemistry and thermodynamic-mixing properties of the aluminosilicate garnets. <i>American Mineralogist</i> , 1997, 82, 571-581.	1.9	50
65	Heat capacity measurements of synthetic pyrope-grossular garnets between 320 and 1000 K by differential scanning calorimetry. <i>Geochimica Et Cosmochimica Acta</i> , 1996, 60, 3215-3227.	3.9	65
66	A ²⁹ Si MAS NMR and IR spectroscopic investigation of synthetic pyrope-grossular garnet solid solutions. <i>American Mineralogist</i> , 1995, 80, 691-704.	1.9	56
67	Crystal field stabilization energies of almandine-pyrope and almandine-spessartine garnets determined by FTIR near infrared measurements. <i>Physics and Chemistry of Minerals</i> , 1994, 21, 516.	0.8	30
68	Andradite crystal chemistry, dynamic X-site disorder and structural strain in silicate garnets. <i>European Journal of Mineralogy</i> , 1993, 5, 59-72.	1.3	101
69	29Si and 27Al MAS-NMR spectroscopy of glasses in the system CaSiO ₃ -MgSiO ₃ -Al ₂ O ₃ . <i>Chemical Geology</i> , 1992, 96, 387-397.	3.3	18
70	The Hardwood Gneiss: Evidence for High P-T Archean Metamorphism in the Southern Province of the Lake Superior Region. <i>Journal of Geology</i> , 1990, 98, 273-281.	1.4	2