

Mikhail V Ryzhkov

List of Publications by Year in descending order

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13
papers

163
citations

1478505

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1281871

11
g-index

13
all docs

13
docs citations

13
times ranked

190
citing authors

#	ARTICLE	IF	CITATIONS
1	Plutonium complexes in water: new approach to ab initio modeling. <i>Radiochimica Acta</i> , 2021, 109, 327-342.	1.2	2
2	First-principles study on the plutonium ions interaction with diamide molecules in acid solutions. <i>International Journal of Quantum Chemistry</i> , 2021, 121, e26681.	2.0	0
3	Transformation of electron density distribution induced by the cation point defects in uranium dioxide. <i>Journal of Radioanalytical and Nuclear Chemistry</i> , 2020, 325, 253-262.	1.5	0
4	The nature of the chemical bond in UO ₂ . <i>International Journal of Quantum Chemistry</i> , 2019, 119, e26040.	2.0	4
5	Electronic structure and effective charges on atoms near anion point defects in uranium dioxide. <i>Computational Condensed Matter</i> , 2019, 18, e00353.	2.1	1
6	The electronic structure and the nature of the chemical bond in CeO ₂ . <i>Physical Chemistry Chemical Physics</i> , 2018, 20, 16167-16175.	2.8	45
7	X-ray photoelectron spectra structure and chemical bonding in AmO ₂ . <i>Nuclear Technology and Radiation Protection</i> , 2015, 30, 83-98.	0.8	10
8	Electronic structure of predicted endohedral fullerenes An@C40 (An=Th-Md). <i>Computational and Theoretical Chemistry</i> , 2013, 1013, 70-77.	2.5	19
9	Electronic structure of endohedral fullerenes An@C28 (An=Th-Md). <i>Computational and Theoretical Chemistry</i> , 2012, 985, 46-52.	2.5	36
10	Valence electronic state density in thorium dioxide. <i>Nuclear Technology and Radiation Protection</i> , 2008, 23, 34-42.	0.8	10
11	Electronic structure and geometry optimization of nanoparticles Fe ₂ C, FeC ₂ , Fe ₃ C, FeC ₃ and Fe ₂ C ₂ . <i>Chemical Physics Letters</i> , 2005, 404, 400-408.	2.6	34
12	Electronic Structure and Chemical Bonding in Î ² -Sialons. <i>Journal of Structural Chemistry</i> , 2002, 43, 18-25.	1.0	1
13	Quantum-chemical simulation of the electronic structure and chemical bonding in the new superstoichiometric titanium carbonitride Ti ₂ CN ₄ . <i>Mendeleev Communications</i> , 2001, 11, 184-185.	1.6	1