Peter J Swedlund

List of Publications by Year in descending order

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471509 361022 36 1,836 17 35 citations h-index g-index papers 36 36 36 2173 docs citations times ranked citing authors all docs

#	Article	IF	CITATIONS
1	An Experiment to Visualize and Probe the Relationship between Polysaccharide Structure and a Glycemic Index Proxy. Journal of Chemical Education, 2021, 98, 553-558.	2.3	1
2	Depositing alginate on the surface of bilayer emulsions. Food Hydrocolloids, 2020, 100, 105385.	10.7	18
3	Fabrication of Spray-Dried Microcapsules Containing Noni Juice Using Blends of Maltodextrin and Gum Acacia: Physicochemical Properties of Powders and Bioaccessibility of Bioactives during In Vitro Digestion. Foods, 2020, 9, 1316.	4.3	20
4	Effect of O2, NiO coatings, and iron oxide phases on pentachlorophenol dechlorination by zero-valent iron. Environmental Science and Pollution Research, 2019, 26, 27687-27698.	5. 3	4
5	Mid-infrared Spectroscopy as a Probe for Caramelization in Dispersed Systems. Food Analytical Methods, 2019, 12, 2121-2128.	2.6	O
6	H4SiO4 sorption and polymerization at the magnetite - aqueous interface: The influence of interfacial redox state. Applied Geochemistry, 2019, 104, 146-157.	3.0	4
7	Pentachlorophenol dechlorination with zeroÂvalent iron: a Raman and GCMS study of the complex role of surficial iron oxides. Environmental Science and Pollution Research, 2018, 25, 17797-17806.	5.3	10
8	From swollen gels to dried films: Relating the IR spectra of ferrihydrite dried as a film on an ATRIR crystal to aqueous suspensions. Applied Geochemistry, 2018, 91, 89-96.	3.0	1
9	Physical and rheological properties of fish gelatin gel as influenced by κ-carrageenan. Food Bioscience, 2017, 20, 88-95.	4.4	68
10	The Application of Raman Spectroscopy to Probe the Association of H4SiO4 with Iron Oxides. Aquatic Geochemistry, 2017, 23, 21-31.	1.3	8
11	Retrogradation of Maize Starch after High Hydrostatic Pressure Gelation: Effect of Amylose Content and Depressurization Rate. PLoS ONE, 2016, 11, e0156061.	2.5	12
12	Effect of high hydrostatic pressure on the supramolecular structure of corn starch with different amylose contents. International Journal of Biological Macromolecules, 2016, 85, 604-614.	7.5	52
13	Cu(II) removal by Anoxybacillus flavithermus–iron oxide composites during the addition of Fe(II)aq. Geochimica Et Cosmochimica Acta, 2016, 172, 139-158.	3.9	17
14	Bacterial exudate effects on Cu2+ sorption by cells: Quantifying significant ternary interactions. Geochimica Et Cosmochimica Acta, 2015, 149, 268-278.	3.9	6
15	Demystifying the interfacial aquatic geochemistry of thallium(I): New and old data reveal just a regular cation. Chemical Geology, 2015, 398, 97-103.	3.3	38
16	Quantifying arsenate partitioning in aquatic systems: Narrowing the laboratory-real world gap with kinetic sediment extractions and the Diffuse Layer Model. Applied Geochemistry, 2015, 58, 38-45.	3.0	1
17	A spectroscopic and Monte Carlo study of the unexpected promotion of interfacial H4SiO4 polymerization on an iron oxide in the presence of arsenate. Colloids and Surfaces A: Physicochemical and Engineering Aspects, 2015, 486, 69-77.	4.7	3
18	Calcium Phosphates in Ca ²⁺ -Fortified Milk: Phase Identification and Quantification by Raman Spectroscopy. Journal of Agricultural and Food Chemistry, 2014, 62, 12223-12228.	5.2	14

#	Article	IF	CITATIONS
19	Arsenate–Ferrihydrite Systems from Minutes to Months: A Macroscopic and IR Spectroscopic Study of an Elusive Equilibrium. Environmental Science & Equilibrium. Environmental Equilibrium. Envi	10.0	31
20	The influence of surface structure on H4SiO4 sorption and oligomerization on goethite surfaces: An XPS study using goethites differing in morphology. Chemical Geology, 2013, 347, 114-122.	3.3	16
21	The Influence of Surface Structure on H ₄ SiO ₄ Oligomerization on Rutile and Amorphous TiO ₂ Surfaces: An ATR-IR and Synchrotron XPS Study. Langmuir, 2012, 28, 16890-16899.	3.5	16
22	Short range order at the amorphous TiO2–water interface probed by silicic acid adsorption and interfacial oligomerization: An ATR-IR and 29Si MAS-NMR study. Journal of Colloid and Interface Science, 2012, 368, 447-455.	9.4	16
23	Experimental and theoretical investigations into the counter-intuitive shift in the antisymmetric ν(Si–O) vibrational modes upon deuteration of solvated silicic acid (H ₄ SiO ₄). Physical Chemistry Chemical Physics, 2011, 13, 2314-2322.	2.8	27
24	Ionic Strength Effects on Silicic Acid (H ₄ SiO ₄) Sorption and Oligomerization on an Iron Oxide Surface: An Interesting Interplay between Electrostatic and Chemical Forces. Langmuir, 2011, 27, 12930-12937.	3.5	33
25	Cadmium (II) distribution in complex aquatic systems containing ferrihydrite, bacteria and an organic ligand: The effect of bioactivity. Applied Geochemistry, 2011, 26, 898-906.	3.0	6
26	Assessing the role of silicate polymerization on metal oxyhydroxide surfaces using X-ray photoelectron spectroscopy. Chemical Geology, 2011, 285, 62-69.	3.3	46
27	Degradation of Chlorinated Phenols by Zero Valent Iron and Bimetals of Iron: A Review. Environmental Engineering Research, 2011, 16, 187-203.	2.5	57
28	Insights into H4SiO4 surface chemistry on ferrihydrite suspensions from ATR-IR, Diffuse Layer Modeling and the adsorption enhancing effects of carbonate. Journal of Colloid and Interface Science, 2010, 352, 149-157.	9.4	33
29	Silicic Acid Adsorption and Oligomerization at the Ferrihydriteâ 'Water Interface: Interpretation of ATR-IR Spectra Based on a Model Surface Structure. Langmuir, 2010, 26, 3394-3401.	3.5	71
30	Cadmium(II) Speciation in Complex Aquatic Systems: A Study with Ferrihydrite, Bacteria, and an Organic Ligand. Environmental Science & Eamp; Technology, 2009, 43, 7430-7436.	10.0	37
31	Goethite adsorption of Cu(II), Pb(II), Cd(II), and Zn(II) in the presence of sulfate: Properties of the ternary complex. Geochimica Et Cosmochimica Acta, 2009, 73, 1548-1562.	3.9	126
32	An attenuated total reflectance IR study of silicic acid adsorbed onto a ferric oxyhydroxide surface. Geochimica Et Cosmochimica Acta, 2009, 73, 4199-4214.	3.9	74
33	Copper(II) and Cadmium(II) Sorption onto Ferrihydrite in the Presence of Phthalic Acid: Some Properties of the Ternary Complex. Environmental Science & Environmental Science	10.0	40
34	The physicochemical properties of spray-dried watermelon powders. Chemical Engineering and Processing: Process Intensification, 2007, 46, 386-392.	3.6	560
35	The effect of SO4 on the ferrihydrite adsorption of Co, Pb and Cd: ternary complexes and site heterogeneity. Applied Geochemistry, 2003, 18, 1671-1689.	3.0	74
36	Trace Metal Adsorption onto an Acid Mine Drainage Iron(III) Oxy Hydroxy Sulfate. Environmental Science & Environmental Science	10.0	296