Zhenisgul I Sagintaeva

List of Publications by Year
in descending order

48	58	2257263 3	2053342 5
papers	citations	h-index	g-index

48	48	48	26
all docs	docs citations	times ranked	citing authors

#	Article	IF	Citations
1	Ferrites YbSrFe2O5.5 and YbBaFe2O5.5: Synthesis and X-ray diffraction, thermodynamic, and electrophysical properties. Russian Journal of Inorganic Chemistry, 2006, 51, 368-373.	0.3	4
2	Heat capacity and thermodynamic functions of manganite ferrites NdMIMnFeO5 (MI = Li, Na) in the range of 298â€“673 K. Russian Journal of Physical Chemistry A, 2013, 87, 719-723.	0.1	4
3	Heat capacities and thermodynamic functions of new cobalt manganites LaM 2 II CoMnO6 (MII-Mg, Ca,) Tj ETQq1 1 0.784314 rgBT /Ov	0.1	4
4	A thermodynamic investigation of NdMe3Sr3Mn4O12 (Meâ€”Li, Na, K) manganites in the range from 298.15 to 673 K. High Temperature, 2010, 48, 198-204.	0.1	3
5	Thermodynamic and electrophysical properties of LaSrMnFeO5.5 ferrite. High Temperature, 2012, 50, 736-738.	0.1	3
6	Synthesis and x-ray diffraction study of new nanostructured manganite ferrites NdM 1.5 II MnFeO6 (MII) Tj ETQq0 0 0 rgBT /Overlock 10	0.3	3
7	Heat capacity and electrophysical properties of GdMeFe2O5(Me â€” Li, Na, K, Cs)-type ferrites. High Temperature, 2013, 51, 54-59.	0.1	3
8	Thermochemistry of myricetin flavonoid. Russian Journal of Physical Chemistry A, 2014, 88, 1277-1280.	0.1	3
9	Heat capacity and thermodynamic functions of nanostructured manganese ferrites of composition NdMe1.5MnFeO6 (Me = Mg, Ca, Sr, and Ba) in the temperature range from 298.15 to 673 K. Russian Journal of Physical Chemistry A, 2015, 89, 586-591.	0.1	3
10	Heat Capacity and Thermodynamic Functions of NdMeFe2O5(Me is Li, Na, K, Cs) Ferrites. High Temperature, 2004, 42, 409-413.	0.1	2
11	Synthesis and X-ray diffraction and calorimetric studies of LaLiMnFeO5 and LaCsMnFeO5 ferrites. Russian Journal of Inorganic Chemistry, 2008, 53, 1455-1458.	0.3	2
12	Synthesis and X-ray diffraction study of nanostructured particles of cuprate manganites LaM 2 II CuMnO6 (MII = Mg, Ca, Sr, Ba). Russian Journal of Inorganic Chemistry, 2014, 59, 1010-1014.	0.3	2
13	Heat capacity and thermodynamic functions of new nanostructured cuprate-manganite NdCa2CuMnO6. Russian Journal of Physical Chemistry A, 2014, 88, 1802-1805.	0.1	2
14	Calorimetric investigation of heat capacity of the ErMFe2O5.5 (M = Mg, Ca, Sr, Ba) ferrites in the temperature range of 298.15â€“673 K and calculation of their thermodynamic functions. High Temperature, 2015, 53, 358-362.	0.1	2
15	Heat capacity and thermodynamic functions of new cobaltic manganites NdM 2 II CoMnO6 (MII is Mg,) Tj ETQq1 1 0.784314 rgBT /Ov	0.1	2
16	Heat capacities and thermodynamic functions of new nanosized ferro-chromo-manganites LaM0.5 IIFeCrMnO6.5 (MIIâ€“Mg, Ca, Sr, Ba). Russian Journal of Physical Chemistry A, 2017, 91, 430-436.	0.1	2
17	Heat capacity and thermodynamic functions of new cobalt manganites NdM2 I CoMnO5 (MI = Li, Na, and) Tj ETQq1 1 0.784314 rgBT /O	0.1	2
18	Thermochemistry of sesquiterpene lactone argolide. Russian Journal of Physical Chemistry A, 2017, 91, 6-9.	0.1	2

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19	Calorimetric studies of LaM2NiMnO5 (Mâ€"Li, Na, K) nickelite-manganite heat capacity within the temperature range of 298.15â€"673 K. High Temperature, 2017, 55, 465-468.	0.1	2
20	Synthesis and Properties of NdMCr2O5(M = Na, K, Cs) and NdMgCr2O5.5Chromites. Inorganic Materials, 2004, 40, 976-978.	0.2	1
21	The Heat Capacity and Thermodynamic Functions of Ternary Manganites DyMIMgMn2O6 (MI â^' Na, K, Cs) in the Temperature Range from 223 to 673 K. High Temperature, 2005, 43, 727-732.	0.1	1
22	Manganites NdMg 3 I Mg3Mn4O12 (MI = Li, Na, K): X-ray diffraction data. Russian Journal of Inorganic Chemistry, 2009, 54, 30-32.	0.3	1
23	New manganites NdM3Sr3Mn4O12 and NdM3Ba3Mn4O12 (M = Li, Na, K): Synthesis and X-ray diffraction characteristics. Russian Journal of Inorganic Chemistry, 2009, 54, 377-380.	0.3	1
24	X-ray powder diffraction features of manganites DyM 3 I M 3 II Mn4O12 (MI = Li, Na, K; MII = Mg, Ba). Russian Journal of Inorganic Chemistry, 2010, 55, 1454-1457.	0.3	1
25	Synthesis and X-ray diffraction study of ferrites ErMIFe2O5 (MI = Li, Na, K, Cs). Russian Journal of Inorganic Chemistry, 2010, 55, 1607-1610.	0.3	1
26	Characteristics of coal from the Kushmurun deposit. Solid Fuel Chemistry, 2014, 48, 147-148.	0.2	1
27	Thermodynamic properties of sesquiterpene lactone grossheimin. Russian Journal of Physical Chemistry A, 2016, 90, 1521-1524.	0.1	1
28	The Heat Capacity and Electrophysical Properties of Neodymium and Lithium Chromite NdLiCr2O5. High Temperature, 2005, 43, 796-799.	0.1	0
29	Thermodynamic properties of ferrites of composition GdMIIFe2O5.5 (MII = Mg, Ca, Sr). Russian Journal of Applied Chemistry, 2006, 79, 1225-1229.	0.1	0
30	Heat Capacity and thermodynamic functions of DyMeIICr2O5.5(MeII-Mg, Ca) in the range from 298.15 to 673 K. High Temperature, 2007, 45, 645-648.	0.1	0
31	Synthesis and X-ray diffraction study of manganites LaM 3 I M 3 II Mn4O12(MI= Li, Na, K; MII = Mg, Ca). Russian Journal of Inorganic Chemistry, 2007, 52, 1340-1342.	0.3	0
32	Synthesis and X-ray diffraction study of the LaMgIMg(CrO3)2 (MI = Li, Na, K) compounds. Russian Journal of Inorganic Chemistry, 2008, 53, 1691-1693.	0.3	0
33	The calorimetry and thermodynamic functions of Nd Mg 3 I Mn4O12 (MeI-Li, Na, K) manganites in the range from 298.15 to 673 K. High Temperature, 2009, 47, 27-32.	0.1	0
34	Chromites YbMCr2O5 (M = Li, Na, K, Cs): X-ray diffraction study. Russian Journal of Inorganic Chemistry, 2009, 54, 27-29.	0.3	0
35	Study of the heat capacity of the derivatives C21H16N2O and C21H19N2O2Br of the alkaloid harmine. Russian Journal of Applied Chemistry, 2011, 84, 1454-1455.	0.1	0
36	X-ray diffraction characteristics of new chromitomanganites LaM 3 I CrMnO6 and LaM 3 II CrMnO7.5 (MI = Li, Na; MII = Mg, Ca). Russian Journal of Inorganic Chemistry, 2013, 58, 206-208.	0.3	0

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37	Estimating the standard thermodynamic functions of rare-earth and alkali-earth manganitoferrites LnMIIMnFeO5.5 (Ln = La, Nd, Gd, Dy, Er; MII = Mg, Ca, Sr, Ba). Russian Journal of Physical Chemistry A, 2013, 87, 1057-1059.	0.1	0
38	X-ray powder diffraction study of nanostructured particles of manganite ferrites NdMIMnFeO5 (MI =) Tj ETQq0 0 0 rgBT /Overlock 10 Tf	0.3	0
39	Synthesis and X-ray diffraction study of LaM 1.5 II MnFeO6 manganitoferrites (MII = Mg, Ca, Sr, Ba). Russian Journal of Inorganic Chemistry, 2014, 59, 373-375.	0.3	0
40	Enthalpies of dissolution of flavonoids in 96% ethanol at 25Â°C. Russian Journal of Physical Chemistry A, 2015, 89, 1804-1807.	0.1	0
41	Thermodynamic Properties of Zincate-Manganites of LaM2 IIZnMnO6 (ÐœII = Mg, Ca, Sr, Ba) Composition. Russian Journal of Physical Chemistry A, 2016, 90, 739-743.	0.1	0
42	Chemical composition and heat capacity of shale from the Kendyrlyk and Shubarkol deposits. Solid Fuel Chemistry, 2016, 50, 149-151.	0.2	0
43	Thermochemistry of Sesquiterpene Lactone 3,4Î²-Epoxyarglabin. Russian Journal of Physical Chemistry A, 2018, 92, 232-234.	0.1	0
44	Thermodynamic and Electrophysical Properties of Nanosized LaMeFeCrMnO6.5 (Me = Li, Na, K) Ferro-Chromo-Manganites. Russian Journal of Physical Chemistry A, 2018, 92, 760-767.	0.1	0
45	Thermodynamic Properties of Nanosized Cobaltite (Nickelite) Cuprate Manganites LaMgCoCuMnO6 and LaMgNiCuMnO6. Russian Journal of Physical Chemistry A, 2020, 94, 18-22.	0.1	0
46	New nanostructured manganites of LaMeIICuZnMnO6 (MeII â€" Mg, Ca, Sr, Ba). Bulletin of the Karaganda University Chemistry Series, 2021, 103, 60-66.	0.2	0
47	SYNTHESIS AND STUDY OF THERMODYNAMIC PROPERTIES OF NEW ZINCATE-MANGANITES NdM2IIZnMnO6 (MII â^' Mg, Ca). ChemChemTech, 2018, 61, 16.	0.1	0
48	Calorimetric Research into the Heat Capacity of Novel Nano-Sized Cobalt(Nickelite)-Cuprate-Manganites of LaBaMeIICuMnO6 (MeII = Co, Ni) and their Thermodynamic Properties. Eurasian Chemico-Technological Journal, 2020, 22, 27.	0.3	0