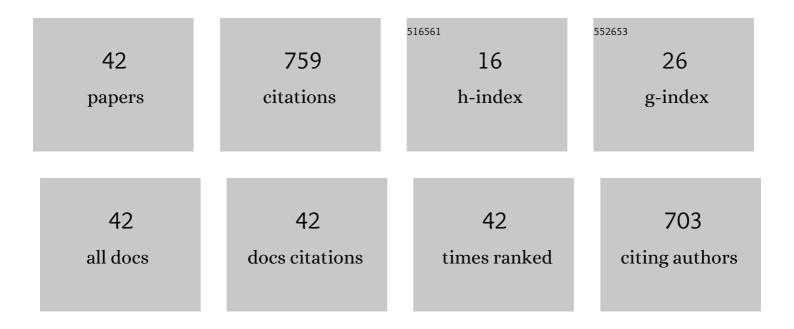
Namal Piyantha

List of Publications by Year in descending order

Source: https://exaly.com/author-pdf/1737300/publications.pdf Version: 2024-02-01



| # | Article | IF | CITATIONS |
|----|---|-----|-----------|
| 1 | Environmentally friendly adsorbent derived from rock melon skin for effective removal of toxic brilliant green dye: linear versus non-linear analyses. International Journal of Environmental Analytical Chemistry, 2023, 103, 4904-4923. | 1.8 | 12 |
| 2 | Inhibitive Action of Selected Model Compounds of Eugenol on Mild Steel Corrosion in Salty Medium. Protection of Metals and Physical Chemistry of Surfaces, 2021, 57, 412-421. | 0.3 | 0 |
| 3 | Effective and Simple NaOH-Modification Method to Remove Methyl Violet Dye via Ipomoea aquatica Roots. Adsorption Science and Technology, 2021, 2021, 1-12. | 1.5 | 25 |
| 4 | Chemical characteristics of wet precipitation at Peradeniya in Sri Lanka. Environmental Monitoring and Assessment, 2021, 193, 14. | 1.3 | 5 |
| 5 | Removal of textile dyes from industrial effluents using burnt brick pieces: adsorption isotherms, kinetics and desorption. SN Applied Sciences, 2020, 2, 1. | 1.5 | 12 |
| 6 | Source apportionment of rainwater chemical composition in wet precipitation at Kelaniya in Sri Lanka. Air Quality, Atmosphere and Health, 2020, 13, 1497-1504. | 1.5 | 13 |
| 7 | Risk assessment and source apportionment of wet bulk deposition in three typical sites of Gampaha District, Sri Lanka. SN Applied Sciences, 2020, 2, 1. | 1.5 | 9 |
| 8 | Adsorption of phosphates from water by two polymer-silicate composites. Bioremediation Journal, 2020, 24, 231-250. | 1.0 | 0 |
| 9 | Trace Metal Composition of Bulk Precipitation in Selected Locations of Kandy District, Sri Lanka. Water, Air, and Soil Pollution, 2020, 231, 1. | 1.1 | 5 |
| 10 | Atmospheric chemical composition of bulk deposition at two geographically distinct locations in Sri Lanka. Environmental Monitoring and Assessment, 2020, 192, 452. | 1.3 | 7 |
| 11 | Biosorption of Cr(III) and Cr(VI) species on NaOH-modified peel of Artocarpus nobilis fruit. 1. Investigation of kinetics. Applied Water Science, 2020, 10, 1. | 2.8 | 12 |
| 12 | Irreversible sorption of Pb(II) from aqueous solution on breadfruit peel to mitigate environmental pollution problems. Water Science and Technology, 2019, 80, 2241-2249. | 1.2 | 5 |
| 13 | Artocarpus odoratissimus Leaves as an Eco-friendly Adsorbent for the Removal of Toxic Rhodamine B Dye in Aqueous Solution: Equilibrium Isotherm, Kinetics, Thermodynamics and Regeneration Studies. Arabian Journal for Science and Engineering, 2018, 43, 6011-6020. | 1.7 | 27 |
| 14 | Biosorption of cationic dyes on breadfruit (Artocarpus altilis) peel and core. Applied Water Science, 2018, 8, 1. | 2.8 | 23 |
| 15 | Sodium hydroxide modified rice husk for enhanced removal of copper ions. Water Science and Technology, 2018, 78, 1615-1623. | 1.2 | 3 |
| 16 | A superb modified new adsorbent, Artocarpus odoratissimus leaves, for removal of cationic methyl violet 2B dye. Environmental Earth Sciences, 2016, 75, 1. | 1.3 | 17 |
| 17 | Utilizing <i>Artocarpus altilis</i> (breadfruit) skin for the removal of malachite green: isotherm, kinetics, regeneration, and column studies. Desalination and Water Treatment, 2016, 57, 16601-16610. | 1.0 | 14 |
| 18 | Removal of crystal violet dye from aqueous solution using yeast-treated peat as adsorbent: thermodynamics, kinetics, and equilibrium studies. Environmental Earth Sciences, 2016, 75, 1. | 1.3 | 41 |

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|----|---|-----|-----------|
| 19 | Adsorption behaviour of Cr(VI) by Muthurajawela peat. Desalination and Water Treatment, 2016, 57, 16592-16600. | 1.0 | 8 |
| 20 | <i>Artocarpus camansi</i> Blanco (Breadnut) core as low-cost adsorbent for the removal of methylene blue: equilibrium, thermodynamics, and kinetics studies. Desalination and Water Treatment, 2016, 57, 5673-5685. | 1.0 | 23 |
| 21 | Investigation of the sorption characteristics of water lettuce (WL) as a potential low-cost biosorbent for the removal of methyl violet 2B. Desalination and Water Treatment, 2016, 57, 8319-8329. | 1.0 | 14 |
| 22 | Adsorption of crystal violet dye from aqueous solution onto chemically treated <i>Artocarpus odoratissimus</i> skin: equilibrium, thermodynamics, and kinetics studies. Desalination and Water Treatment, 2016, 57, 10246-10260. | 1.0 | 30 |
| 23 | Artocarpus altilis (breadfruit) skin as a potential low-cost biosorbent for the removal of crystal violet dye: equilibrium, thermodynamics and kinetics studies. Environmental Earth Sciences, 2015, 73, 3239-3247. | 1.3 | 58 |
| 24 | Removal behavior of peat collected from Brunei Darussalam for Pb(II) ions from aqueous solution: equilibrium isotherm, thermodynamics, kinetics and regeneration studies. Environmental Earth Sciences, 2015, 74, 2541-2551. | 1.3 | 14 |
| 25 | Effective adsorption of toxic brilliant green from aqueous solution using peat of Brunei Darussalam: isotherms, thermodynamics, kinetics and regeneration studies. RSC Advances, 2015, 5, 34603-34615. | 1.7 | 51 |
| 26 | Sorption characteristics of peat from Brunei Darussalam for the removal of rhodamine B dye from aqueous solution: adsorption isotherms, thermodynamics, kinetics and regeneration studies. Desalination and Water Treatment, 2015, 55, 664-677. | 1.0 | 52 |
| 27 | Sorption characteristics of peat of Brunei Darussalam V: removal of Congo red dye from aqueous solution by peat. Desalination and Water Treatment, 2015, 54, 2592-2600. | 1.0 | 42 |
| 28 | Biosorption and Desorption of Lead(II) by <i>Hydrilla verticillata</i> . Bioremediation Journal, 2014, 18, 192-203. | 1.0 | 30 |
| 29 | Sorption characteristics of peat of Brunei Darussalam IV: equilibrium, thermodynamics and kinetics of adsorption of methylene blue and malachite green dyes from aqueous solution. Environmental Earth Sciences, 2014, 72, 2263-2277. | 1.3 | 49 |
| 30 | Adsorption Behavior of Methyl Violet 2B Using Duckweed: Equilibrium and Kinetics Studies. Arabian Journal for Science and Engineering, 2014, 39, 6757-6765. | 1.1 | 32 |
| 31 | Biosorption of Cr(III) and Cr(VI) species from aqueous solution by Cabomba caroliniana: kinetic and equilibrium study. Environmental Earth Sciences, 2013, 70, 661-671. | 1.3 | 25 |
| 32 | Interaction of Cr(VI) species with thermally treated brick clay. Environmental Science and Pollution Research, 2011, 18, 75-81. | 2.7 | 12 |
| 33 | Investigation of kinetics of Cr(VI)–fired brick clay interaction. Journal of Hazardous Materials, 2011, 188, 193-197. | 6.5 | 18 |
| 34 | Removal of blue colouration from industrial effluents by burnt brick particles. Journal of the National Science Foundation of Sri Lanka, 2010, 28, 287. | 0.1 | 1 |
| 35 | Solvent extraction followed by ultraviolet detection for investigation of tetramethylthiuram disulfide at soil-water interface. International Journal of Environmental Science and Technology, 2008, 5, 547-554. | 1.8 | 9 |
| 36 | Removal of Sulfate, Phosphate and Colored Substances in Wastewater Effluents using Feldspar. Water Resources Management, 2000, 14, 417-434. | 1.9 | 24 |

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|----|---|-----|-----------|
| 37 | <i>Artocarpus odoratissimus</i> skin as a potential low-cost biosorbent for the removal of methylene blue and methyl violet 2B. Desalination and Water Treatment, 0, , 1-12. | 1.0 | 17 |
| 38 | Characterization of peat samples collected from Brunei Darussalam and their evaluation as potential adsorbents for Cu(II) removal from aqueous solution. Desalination and Water Treatment, 0, , 1-15. | 1.0 | 4 |
| 39 | Enhancement of adsorption characteristics of Methyl violet 2B dye through NaOH treatment of Cucumis melo var. cantalupensis (rock melon) skin. , 0, 180, 336-348. | | 5 |
| 40 | Converting Hylocereus undatus (white dragon fruit) peel waste into a useful potential adsorbent for the removal of toxic Congo red dye. , 0, 185, 307-317. | | 8 |
| 41 | Biosorption of heavy metal ions on peel of Artocarpus nobilis fruit: 2. Improvement of biosorption capacities of Ni(II) through different modifications. , 0, 185, 226-236. | | 3 |
| 42 | Synthesis, characterization, and textile dye adsorption studies of a kaolin-based polymer layer silicate composite. International Journal of Environmental Science and Technology, 0, , 1. | 1.8 | 0 |