## Céline Poncet-Legrand

List of Publications by Year in descending order

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#	Article	IF	CITATIONS
1	Structural characteristics of Saccharomyces cerevisiae mannoproteins: Impact of their polysaccharide part. Carbohydrate Polymers, 2022, 277, 118758.	10.2	8
2	Impact of the variety on the adsorption of anthocyanins and tannins on grape flesh cell walls. Journal of the Science of Food and Agriculture, 2022, 102, 3379-3392.	3.5	3
3	Impact of grape variety, berry maturity and size on the extractability of skin polyphenols during model wineâ€ŀike maceration experiments. Journal of the Science of Food and Agriculture, 2021, 101, 3257-3269.	3.5	9
4	Wine Thermosensitive Proteins Adsorb First and Better on Bentonite during Fining: Practical Implications and Proposition of Alternative Heat Tests. Journal of Agricultural and Food Chemistry, 2020, 68, 13450-13458.	5.2	8
5	Multimethod Approach for Extensive Characterization of Gallnut Tannin Extracts. Journal of Agricultural and Food Chemistry, 2020, 68, 13426-13438.	5.2	13
6	Grape seed and apple tannins: Emulsifying and antioxidant properties. Food Chemistry, 2015, 178, 38-44.	8.2	44
7	Protein/Polysaccharide Interactions and Their Impact on Haze Formation in White Wines. Journal of Agricultural and Food Chemistry, 2015, 63, 10042-10053.	5.2	23
8	Probing the micellar solubilisation and inter-micellar exchange of polyphenols using the DPPH free radical. Food Chemistry, 2014, 149, 114-120.	8.2	7
9	Condensed Tannin Changes Induced by Autoxidation: Effect of the Initial Degree of Polymerization and Concentration. Journal of Agricultural and Food Chemistry, 2014, 62, 7833-7842.	5.2	16
10	White Wine Proteins: How Does the pH Affect Their Conformation at Room Temperature?. Langmuir, 2013, 29, 10475-10482.	3.5	24
11	Stability of White Wine Proteins: Combined Effect of pH, Ionic Strength, and Temperature on Their Aggregation. Journal of Agricultural and Food Chemistry, 2012, 60, 1308-1319.	5.2	61
12	Characterization of oxidized tannins: comparison of depolymerization methods, asymmetric flow field-flow fractionation and small-angle X-ray scattering. Analytical and Bioanalytical Chemistry, 2011, 401, 1559-1569.	3.7	47
13	Tannin Oxidation: Intra- versus Intermolecular Reactions. Biomacromolecules, 2010, 11, 2376-2386.	5.4	108
14	Flavanols, Flavonols and Dihydroflavonols. , 2009, , 463-507.		13
15	Aggregation of a Proline-Rich Protein Induced by Epigallocatechin Gallate and Condensed Tannins: Effect of Protein Glycosylation. Journal of Agricultural and Food Chemistry, 2008, 56, 6724-6732.	5.2	81
16	Colloidal Dispersions of Tannins in Waterâ^'Ethanol Solutions. Langmuir, 2007, 23, 9949-9959.	3.5	63
17	Interactions between Flavan-3-ols and Poly( <scp>l</scp> -proline) Studied by Isothermal Titration Calorimetry: Effect of the Tannin Structure. Journal of Agricultural and Food Chemistry, 2007, 55, 9235-9240.	5.2	143
18	Interactions between a Non Glycosylated Human Proline-Rich Protein and Flavan-3-ols Are Affected by Protein Concentration and Polyphenol/Protein Ratio. Journal of Agricultural and Food Chemistry, 2007. 55. 4895-4901.	5.2	120

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19	Hybrid Dissymmetrical Colloidal Particles. Chemistry of Materials, 2005, 17, 3338-3344.	6.7	149
20	From Raspberry-like to Dumbbell-like Hybrid Colloids through Surface-assisted Nucleation and Growth of Polystyrene Nodules onto Macromonomer-modified Silica Nanoparticles. Materials Research Society Symposia Proceedings, 2004, 847, 292.	0.1	1
21	Structure determination and colour properties of a new directly linked flavanol–anthocyanin dimer. Tetrahedron Letters, 2004, 45, 8725-8729.	1.4	72
22	Properties of a novel magnetized alginate for magnetic resonance imaging. Biotechnology and Bioengineering, 2003, 83, 282-292.	3.3	53
23	Flavan-3-ol Aggregation in Model Ethanolic Solutions:  Incidence of Polyphenol Structure, Concentration, Ethanol Content, and Ionic Strength. Langmuir, 2003, 19, 10563-10572.	3.5	86
24	Syntheses of Raspberrylike Silica/Polystyrene Materials. Chemistry of Materials, 2002, 14, 2354-2359.	6.7	208
25	Solution Properties of Hydrophobically-Modified Copolymers of N-Isopropylacrylamide and N-L-Valine Acrylamide. A Study by Fluorescence Spectroscopy and Microcalorimetry. Polymer Journal, 2001, 33, 277-283.	2.7	8