## Syed Muhammad Saqib Nadeem

List of Publications by Year in descending order

Source: https://exaly.com/author-pdf/1652200/publications.pdf

Version: 2024-02-01

2258059 2272923 10 17 3 4 citations h-index g-index papers 10 10 10 5 docs citations times ranked citing authors all docs

#	Article	IF	CITATIONS
1	The kinetics of electron transfer reaction of methylene green and titanium trichloride in different solvents. Russian Journal of Physical Chemistry A, 2016, 90, 1143-1150.	0.6	5
2	lonic-Interaction of Potassium Iodide in Edible Oils + DMF System by Viscosity Method. International Journal of Chemistry, 2012, 4, .	0.3	4
3	Photo-kinetics of photoinduced transformation reaction of methylene green with titanium trichloride in different solvents. Russian Journal of Physical Chemistry A, 2017, 91, 1592-1599.	0.6	3
4	The Photokinetics of Electron Transfer Reaction of Methylene Blue with Titanium Trichloride in Aqueous–Alcoholic Solvents. Journal of the Chinese Chemical Society, 2017, 64, 1147-1155.	1.4	2
5	The kinetics of photoâ€induced chemical transformation reaction of methylene blue and titanium trichloride in different solvents. Journal of the Chinese Chemical Society, 2018, 65, 1317-1325.	1.4	2
6	Viscometric Study of Ionic Interactions of MgSO4 in Water and Water–Ethanol Mixtures at Different Temperatures. Russian Journal of Physical Chemistry A, 2022, 96, 849-859.	0.6	1
7	The photokinetics of phenothiazine dyes with titanium trichloride in different solvents. International Journal of Chemical Kinetics, 2019, 51, 379-389.	1.6	O
8	The study of ionic interactions of monovalent electrolytes in aqueous polyvinyl alcohol and polyacrylamide by conductance method. Ionics, 2020, 26, 2927-2940.	2.4	0
9	The effect of potassium iodide on the viscosity of vegetable oil + N, N-dimethylformamide solvent at different temperatures. Chemical Papers, 0, , 1.	2.2	O
10	A volumetric study of ionic interactions of ammonium sulfate in water and aqueous DMF at different temperatures. Chemical Papers, $0$ , $1$ .	2.2	0