Michihiro Nishikawa

List of Publications by Year in descending order

Source: https://exaly.com/author-pdf/160407/publications.pdf

Version: 2024-02-01

		759233	794594
19	780	12	19
papers	citations	h-index	g-index
19	19	19	1037
all docs	docs citations	times ranked	citing authors
an does	does citations	tilles rankeu	citing authors

#	Article	IF	CITATIONS
1	Discrimination of Methanol from Ethanol in Gasoline Using a Membrane-type Surface Stress Sensor Coated with Copper(I) Complex. Bulletin of the Chemical Society of Japan, 2021, 94, 648-654.	3.2	24
2	Photophysical Properties of Simple Palladium(0) Complexes Bearing Triphenylphosphine Derivatives. Inorganic Chemistry, 2021, 60, 9516-9528.	4.0	7
3	Water-soluble copper(<scp>i</scp>) complexes bearing 2,2′-bicinchoninic acid dipotassium salt with red-light absorption and repeatable colour change upon freezing operation. New Journal of Chemistry, 2019, 43, 277-283.	2.8	3
4	Self-assembly as a key player for materials nanoarchitectonics. Science and Technology of Advanced Materials, 2019, 20, 51-95.	6.1	322
5	Materials Nanoarchitectonics for Mechanical Tools in Chemical and Biological Sensing. Chemistry - an Asian Journal, 2018, 13, 3366-3377.	3.3	40
6	Blue and orange oxygen responsive emissions in the solid state based on copper(<scp>i</scp>) complexes bearing dodecafluorinated diphosphine and 1,10-phenanthroline derivative ligands. Inorganic Chemistry Frontiers, 2017, 4, 639-649.	6.0	11
7	Restoration of Color Caused by Redox Reactions of Copper(I) Complexes Bearing 2,2′-Bipyridine and (⟨i⟩R⟨ i⟩)-(+)-2,2′-Bis(di-⟨i⟩p⟨ i⟩-tolylphosphino)-1,1′-binaphthyl Ligands. Bulletin of the Chemical Society of Japan, 2017, 90, 798-800.	3.2	2
8	Redox Properties of Copper(I) Complex Bearing 4,7-Diphenyl-2,9-dimethyl-1,10-phenanthroline and 1,4-Bis(diphenylphosphino)butane Ligands and Effects of Light in the Presence of Chloroform. Bulletin of the Chemical Society of Japan, 2017, 90, 286-288.	3.2	17
9	Photophysical properties of three coordinated copper(<scp>i</scp>) complexes bearing 1,10-phenanthroline and a monodentate phosphine ligand. Dalton Transactions, 2017, 46, 14804-14811.	3.3	21
10	Emission properties and Cu(<scp>i</scp>)–Cu(<scp>i</scp>) interaction in 2-coordinate dicopper(<scp>i</scp>)-bis(N-heterocyclic)carbene complexes. Dalton Transactions, 2016, 45, 12127-12136.	3.3	32
11	Long-lived and oxygen-responsive photoluminescence in the solid state of copper(<scp>i</scp>) complexes bearing fluorinated diphosphine and bipyridine ligands. Dalton Transactions, 2015, 44, 9170-9181.	3.3	38
12	Highly emissive copper(<scp>i</scp>) complexes bearing diimine and bis(diphenylphosphinomethyl)-2,2-dimethyl-1,3-dioxolane. Dalton Transactions, 2015, 44, 411-418.	3.3	46
13	Photophysical Properties of a Copper(I) Complex Bearing 4,4′-Dimethoxy-2,2′-bipyridine and 1,3-Bis(diphenylphosphino)propane. Bulletin of the Chemical Society of Japan, 2014, 87, 912-914.	3.2	11
14	Steric Interference on the Redox-conjugated Pyrimidine Ring Rotation of Mono- and Dinuclear Copper Complexes with (4-Methyl-2-pyrimidinyl)imine Ligands. Chemistry Letters, 2014, 43, 1037-1039.	1.3	3
15	Stimuli-responsive pyrimidine ring rotation in copper complexes for switching their physical properties. Physical Chemistry Chemical Physics, 2013, 15, 10549.	2.8	46
16	Structural Modification on Copper(I)-pyridylpyrimidine Complexes for Modulation of Rotational Dynamics, Redox Properties, and Phototriggered Isomerization. Inorganic Chemistry, 2013, 52, 8962-8970.	4.0	6
17	Solvated-Ion-Pairing-Sensitive Molecular Bistability Based on Copper(I)-Coordinated Pyrimidine Ring Rotation. Inorganic Chemistry, 2013, 52, 369-380.	4.0	21
18	Reversible Copper(II)/(I) Electrochemical Potential Switching Driven by Visible Light-Induced Coordinated Ring Rotation. Journal of the American Chemical Society, 2012, 134, 10543-10553.	13.7	51

#	Article	IF	CITATIONS
19	Dual Emission Caused by Ring Inversion Isomerization of a 4-Methyl-2-pyridyl-pyrimidine Copper(I) Complex. Journal of the American Chemical Society, 2010, 132, 9579-9581.	13.7	79