

# Qinxin Luo

## List of Publications by Year in descending order

Source: <https://exaly.com/author-pdf/1559445/publications.pdf>

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papers

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1478505

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#	ARTICLE		IF	CITATIONS
1	The hexagonal perovskite Ba <sub>0.5</sub> Sr <sub>0.5</sub> Co <sub>0.8</sub> Fe <sub>0.2</sub> O <sub>3-<math>\delta</math></sub> as an efficient electrocatalyst for the oxygen evolution reaction. <i>Inorganic Chemistry Frontiers</i> , 2020, 7, 4488-4497.	6.0	16	
2	A new oxygen-free cobalt-based compound SmCoAsF with multiple magnetic transitions. <i>CrystEngComm</i> , 2020, 22, 4268-4274.	2.6	6	
3	Hexagonal Perovskite Ba <sub>0.9</sub> Sr <sub>0.1</sub> Co <sub>0.8</sub> Fe <sub>0.1</sub> Ir <sub>0.1</sub> O <sub>3-<math>\delta</math></sub> as an Efficient Electrocatalyst towards the Oxygen Evolution Reaction. <i>ACS Applied Energy Materials</i> , 2020, 3, 7149-7158.	5.1	32	
4	Preparation of the Orthorhombic Li <sub>x</sub> (C <sub>2</sub> H <sub>8</sub> N <sub>2</sub> ) <sub>y</sub> Fe <sub>2</sub> Se <sub>2</sub> Superconductor by Amine Exchange Method. <i>ChemistrySelect</i> , 2019, 4, 8201-8206.	1.5	2	
5	Influence of Ions and Temperature on Aqueous Biphasic Systems Containing Ionic Liquid and Ammonium Sulfate. <i>Journal of Chemical &amp; Engineering Data</i> , 2019, 64, 3139-3147.	1.9	8	
6	Facile Scalable Synthesis of Carbon-Coated Ge@C and GeX@C (X=S, Se) Anodes for High Performance Lithium-Ion Batteries. <i>ChemistrySelect</i> , 2019, 4, 6587-6592.	1.5	10	
7	Phase diagram of ionic liquid aqueous two-phase systems with N-butylpyridinium tetrafluoroborate, ammonium Citrate/Sodium acetate, and water from 308.15 K to 328.15 K. <i>Thermochimica Acta</i> , 2016, 632, 72-78.	2.7	6	
8	Temperature-dependent phase behavior of ionic liquid solutions containing N-butylpyridinium nitrate, water, and either sodium or ammonium citrate. <i>Fluid Phase Equilibria</i> , 2015, 403, 118-128.	2.5	4	
9	Extraction and mechanistic investigation of trace dibutyl phthalate using an ionic liquid aqueous two-phase system. <i>New Journal of Chemistry</i> , 2015, 39, 6223-6230.	2.8	9	