Nicolas Clavier

List of Publications by Year
in descending order

118	3,689	117453	161609
papers	citations	34 h-index	54 g-index

126	126	126	2123
all docs	docs citations	times ranked	citing authors

#	Article	IF	Citations
1	Impact of impurities on the fabrication and performances of yttrium-doped thoria electrolyte ceramics. Journal of Nuclear Materials, 2022, 560, 153499.	1.3	2
2	A multiscale in situ high temperature high resolution transmission electron microscopy study of ThO₂ sintering. Nanoscale, 2021, 13, 7362-7374.	2.8	6
3	The Role of Water and Hydroxyl Groups in the Structures of Stetindite and Coffinite, MSiO₄ (M = Ce, U). Inorganic Chemistry, 2021, 60, 718-735.	1.9	18
4	Effect of Annealing on Structural and Thermodynamic Properties of ThSiO₄-ErPO₄ Xenotime Solid Solution. Inorganic Chemistry, 2021, 60, 12020-12028.	1.9	2
5	SEraMic: A semi-automatic method for the segmentation of grain boundaries. Journal of the European Ceramic Society, 2021, 41, 5349-5358.	2.8	6
6	Influence of the PuO2 content on the sintering behaviour of UO2-PuO2 freeze-granulated powders under reducing conditions. Journal of the European Ceramic Society, 2021, 41, 6778-6783.	2.8	3
7	Direct sintering of UO2+x oxides prepared under hydrothermal conditions. Journal of the European Ceramic Society, 2021, 41, 6697-6707.	2.8	3
8	Reaction sintering of rhabdophane into monazite-cheralite Nd1-2xThxCaxPO4 (x = 0 â€“ 0.1) ceramics. Journal of the European Ceramic Society, 2020, 40, 911-922.	2.8	11
9	Sintering of a UO2-PuO2 freeze-granulated powder under reducing conditions. Journal of the European Ceramic Society, 2020, 40, 5900-5908.	2.8	6
10	Hydrothermal Conversion of Thorium Oxalate into ThO₂Â·nH₂O Oxide. Inorganic Chemistry, 2020, 59, 14954-14966.	1.9	13
11	Early stages of UO2+x sintering by in situ high-temperature environmental scanning electron microscopy. Journal of the European Ceramic Society, 2020, 40, 5891-5899.	2.8	5
12	Charge compensation mechanisms in Nd-doped UO2 samples for stoichiometric and hypo-stoichiometric conditions: Lack of miscibility gap. Journal of Nuclear Materials, 2020, 539, 152276.	1.3	21
13	Structural changes of Nd- and Ce-doped ammonium diuranate microspheres during the conversion to U1â^'LnO2Â±. Journal of Nuclear Materials, 2020, 542, 152454.	1.3	2
14	Structural and Thermodynamic Investigation of the Perovskite Ba₂NaMoO_{5.5}. Inorganic Chemistry, 2020, 59, 6120-6130.	1.9	1
15	Oxidation as an Early Stage in the Multistep Thermal Decomposition of Uranium(IV) Oxalate into U3O8. Inorganic Chemistry, 2020, 59, 8589-8602.	1.9	14
16	Uranium removal from mining water using Cu substituted hydroxyapatite. Journal of Hazardous Materials, 2020, 392, 122501.	6.5	43
17	Impact of liquid sodium corrosion on microstructure and electrical properties of yttrium-doped thoria prepared by co-precipitation. Corrosion Science, 2020, 171, 108721.	3.0	4
18	Hydrothermal Conversion of Uranium(IV) Oxalate into Oxides: A Comprehensive Study. Inorganic Chemistry, 2020, 59, 3260-3273.	1.9	24

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19	Synthesis, Crystal Structure, and Enthalpies of Formation of Churchite-type REPO₄Â·2H₂O (RE = Gd to Lu) Materials. Crystal Growth and Design, 2019, 19, 4641-4649.	1.4	20
20	First Stage of Sintering of ThO2 Microspheres: a HT-ESEM and HT-HRTEM Study. Microscopy and Microanalysis, 2019, 25, 49-50.	0.2	0
21	From Th-Rhabdophane to Monazite-Cheralite Solid Solutions: Thermal Behavior of Nd_{1â€“2x}Th_xCa_xPO₄Â·nH₂O (x = 0â€“0.15). Crystal Growth and Design, 2019, 19, 2794-2801.	1.4	9
22	Working with the ESEM at high temperature. Materials Characterization, 2019, 151, 15-26.	1.9	29
23	Impact of the cationic homogeneity on Th0.5U0.5O2 densification and chemical durability. Journal of Nuclear Materials, 2019, 514, 368-379.	1.3	5
24	Direct synthesis of pure brannerite UTi2O6. Journal of Nuclear Materials, 2019, 515, 401-406.	1.3	12
25	Determination of the isotopic composition of single subâ€micrometerâ€sized uranium particles by laser ablation coupled with multiâ€collector inductively coupled plasma mass spectrometry. Rapid Communications in Mass Spectrometry, 2019, 33, 419-428.	0.7	9
26	Structural and thermodynamic study of Cs3Na(MoO4)2: Margin to the safe operation of sodium cooled fast reactors. Journal of Solid State Chemistry, 2019, 269, 1-8.	1.4	4
27	Dissolution kinetics of monazite LnPO4 (Ln = La to Gd): A multiparametric study. Applied Geochemistry, 2018, 93, 81-93.	1.4	25
28	Effect of powder morphology on sintering kinetics, microstructure and mechanical properties of monazite ceramics. Journal of the European Ceramic Society, 2018, 38, 227-234.	2.8	25
29	Synthesis of size-controlled UO2 microspheres from the hydrothermal conversion of U(iv) aspartate. CrystEngComm, 2018, 20, 7749-7760.	1.3	21
30	Thermodynamics and Stability of Rhabdophanes, Hydrated Rare Earth Phosphates REPO4 Â· n H2O. Frontiers in Chemistry, 2018, 6, 604.	1.8	27
31	Monazite, rhabdophane, xenotime & churchite: Vibrational spectroscopy of gadolinium phosphate polymorphs. Spectrochimica Acta - Part A: Molecular and Biomolecular Spectroscopy, 2018, 205, 85-94.	2.0	49
32	Solubility product of the thorium phosphate hydrogen-phosphate hydrate (Th 2 (PO 4) 2 (HPO 4)Â·H 2) Tj ETQq0 0 0 rgBT /Overlock 10	1.0	7
33	Thorium aspartate tetrahydrate precursor to ThO 2 : Comparison of hydrothermal and thermal conversions. Journal of Nuclear Materials, 2017, 487, 331-342.	1.3	19
34	In pursuit of the rhabdophane crystal structure: from the hydrated monoclinic LnPO 4 .0.667H 2 O to the hexagonal LnPO 4 (Ln = Nd, Sm, Gd, Eu and Dy). Journal of Solid State Chemistry, 2017, 249, 221-227.	1.4	42
35	Densification behavior and microstructure evolution of yttrium-doped ThO 2 ceramics. Journal of the European Ceramic Society, 2017, 37, 3381-3391.	2.8	14
36	Structural and thermodynamic study of dicesium molybdate Cs2Mo2O7: Implications for fast neutron reactors. Journal of Solid State Chemistry, 2017, 253, 89-102.	1.4	20

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37	Incorporation of thorium in the rhabdophane structure: Synthesis and characterization of Pr 1-2x Ca x Th x PO 4 Â·nH 2 O solid solutions. Journal of Nuclear Materials, 2017, 492, 88-96.	1.3	14
38	Synthesis and Direct Sintering of Nanosized (M^{IV},M^{III})O_{2â€x} Hydrated Oxides as Electrolyte Ceramics. ChemPhysChem, 2017, 18, 2666-2674.	1.0	3
39	Novel approaches for the in situ study of the sintering of nuclear oxide fuel materials and their surrogates. Radiochimica Acta, 2017, 105, 879-892.	0.5	9
40	High-temperature electron microscopy study of ThO 2 microspheres sintering. Journal of the European Ceramic Society, 2017, 37, 727-738.	2.8	25
41	In Situ Study of CeO2 Microspheres Sintering Using HT-ESEM. Microscopy and Microanalysis, 2016, 22, 62-63.	0.2	0
42	Determination of the Solubility of Rhabdophanes LnPO₄Â·0.667H₂O (Ln = La to) Tj ETQq0 0 0 rgBT /Overlock	1.0	47
43	Preparation, characterization and sintering of yttrium-doped ThO2 for oxygen sensors applications. Journal of Alloys and Compounds, 2016, 689, 374-382.	2.8	19
44	The effect of the synthesis route of monazite precursors on the microstructure of sintered pellets. Progress in Nuclear Energy, 2016, 92, 298-305.	1.3	17
45	Energetics of a Uranothorite (Th_{1â€“x}U_xSiO₄) Solid Solution. Chemistry of Materials, 2016, 28, 7117-7124.	3.2	31
46	Incorporation of Thorium in the Zircon Structure Type through the Th_{1â€“x}Er_x(SiO₄)_{1â€“x}(PO₄_x</sub> Thoriteâ€“Xenotime Solid Solution. Inorganic Chemistry, 2016, 55, 11273-11282.	1.9	18
47	Charged defects during alpha-irradiation of actinide oxides as revealed by Raman and luminescence spectroscopy. Nuclear Instruments & Methods in Physics Research B, 2016, 374, 67-70.	0.6	26
48	Vibrational spectroscopy of synthetic analogues of ankoleite, chernikovite and intermediate solid solution. Spectrochimica Acta - Part A: Molecular and Biomolecular Spectroscopy, 2016, 156, 143-150.	2.0	21
49	From in Situ HT-ESEM Observations to Simulation: How Does Polycrystallinity Affects the Sintering of CeO₂ Microspheres?. Journal of Physical Chemistry C, 2016, 120, 386-395.	1.5	27
50	First experimental determination of the solubility constant of coffinite. Geochimica Et Cosmochimica Acta, 2016, 181, 36-53.	1.6	35
51	Thermodynamics of formation of coffinite, USiO ₄. Proceedings of the National Academy of Sciences of the United States of America, 2015, 112, 6551-6555.	3.3	72
52	Catalytic dissolution of ceriaâ€“lanthanide mixed oxides provides environmentally friendly partitioning of lanthanides and platinum. Hydrometallurgy, 2015, 151, 107-115.	1.8	10
53	An original precipitation route toward the preparation and the sintering of highly reactive uranium cerium dioxide powders. Journal of Nuclear Materials, 2015, 462, 173-181.	1.3	25
54	Coffinite, USiO₄, Is Abundant in Nature: So Why Is It So Difficult To Synthesize?. Inorganic Chemistry, 2015, 54, 6687-6696.	1.9	38

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55	From uranium(IV) oxalate to sintered UO 2 : Consequences of the powders' thermal history on the microstructure. Journal of the European Ceramic Society, 2015, 35, 4535-4546.	2.8	25
56	In situ HT-ESEM study of crystallites growth within CeO2 microspheres. Ceramics International, 2015, 41, 14703-14711.	2.3	18
57	Dissolution of Th1â^'U O2: Effects of chemical composition and microstructure. Journal of Nuclear Materials, 2015, 457, 304-316.	1.3	35
58	Effect of hydration and thermal treatment on ceria surface using non-intrusive techniques. Journal of Nuclear Materials, 2014, 444, 359-367.	1.3	8
59	High-temperature behavior of dicesium molybdate Cs2MoO4: Implications for fast neutron reactors. Journal of Solid State Chemistry, 2014, 215, 225-230.	1.4	22
60	Chemical and mineralogical modifications of simplified radioactive waste calcine during heat treatment. Journal of Nuclear Materials, 2014, 448, 8-19.	1.3	8
61	From thorite to coffinite: A spectroscopic study of Th1â^'xUxSiO4 solid solutions. Spectrochimica Acta - Part A: Molecular and Biomolecular Spectroscopy, 2014, 118, 302-307.	2.0	29
62	Dissolution of uranium mixed oxides: The role of oxygen vacancies vs the redox reactions. Progress in Nuclear Energy, 2014, 72, 101-106.	1.3	16
63	Environmental SEM monitoring of Ce_{1â^'x}Ln_xO_{2â^'x/2}mixed-oxide microstructural evolution during dissolution. Journal of Materials Chemistry A, 2014, 2, 5193-5203.	5.2	52
64	Monoclinic Form of the Rhabdophane Compounds: REEPO₄Â·0.667H₂O. Crystal Growth and Design, 2014, 14, 5090-5098.	1.4	61
65	Preparation and characterisation of uranium oxides with spherical shapes and hierarchical structures. CrystEngComm, 2014, 16, 6944-6954.	1.3	30
66	Dilatometric study of U1â^'xAmxO2Â±Î´ and U1â^'xCexO2Â±Î´ reactive sintering. Journal of Nuclear Materials, 2013, 441, 40-46.	1.3	18
67	Purification of uranothorite solid solutions from polyphase systems. Journal of Nuclear Materials, 2013, 441, 73-83.	1.3	17
68	Solubility properties of synthetic and natural meta-torbernite. Journal of Nuclear Materials, 2013, 442, 195-207.	1.3	23
69	From Uranothorites to Coffinite: A Solid Solution Route to the Thermodynamic Properties of USiO₄. Inorganic Chemistry, 2013, 52, 6957-6968.	1.9	33
70	Versatile Monazite: Resolving geological records and solving challenges in materials science: Monazite as a promising long-term radioactive waste matrix: Benefits of high-structural flexibility and chemical durability. American Mineralogist, 2013, 98, 833-847.	0.9	151
71	Combining in situ HT-ESEM observations and dilatometry: An original and fast way to the sintering map of ThO2. Materials Chemistry and Physics, 2013, 137, 742-749.	2.0	32
72	The Flexible Ba₇UM₂S_{12.5}O_{0.5} (M = V, Fe) Compounds: Syntheses, Structures and Spectroscopic, Resistivity, and Electronic Properties. Inorganic Chemistry, 2013, 52, 12057-12063.	1.9	9

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73	Kinetics of Structural and Microstructural Changes at the Solid/Solution Interface during Dissolution of Cerium(IV)â€“Neodymium(III) Oxides. Journal of Physical Chemistry C, 2012, 116, 12027-12037.	1.5	16
74	In Situ <scp>HT</scp>â€<scp>ESEM</scp> Observation of <scp><scp>CeO</scp></scp>₂ Grain Growth During Sintering. Journal of the American Ceramic Society, 2012, 95, 3683-3690.	1.9	24
75	Catalytic dissolution of ceria under mild conditions. Journal of Materials Chemistry, 2012, 22, 14734.	6.7	29
76	Triclinicâ€“Cubic Phase Transition and Negative Expansion in the Actinide IV (Th, U, Np, Pu) Diphosphates. Inorganic Chemistry, 2012, 51, 4314-4322.	1.9	27
77	Synthesis and characterization of Th1â^'xLnxO2â^'x/2 mixed-oxides. Materials Research Bulletin, 2012, 47, 4017-4025.	2.7	51
78	Dissolution of Cerium(IV)â€“Lanthanide(III) Oxides: Comparative Effect of Chemical Composition, Temperature, and Acidity. Inorganic Chemistry, 2012, 51, 3868-3878.	1.9	44
79	Multiparametric study of Th1â^'xLnxO2â^'x/2 mixed oxides dissolution in nitric acid media. Journal of Nuclear Materials, 2012, 429, 237-244.	1.3	22
80	Preparation and characterization of synthetic Th0.5U0.5SiO4 uranothorite. Progress in Nuclear Energy, 2012, 57, 155-160.	1.3	27
81	Dynamic aspects of cerium dioxide sintering: HT-ESEM study of grain growth and pore elimination. Journal of the European Ceramic Society, 2012, 32, 353-362.	2.8	26
82	Calcined resin microsphere pelletization (CRMP): A novel process for sintered metallic oxide pellets. Journal of the European Ceramic Society, 2012, 32, 3199-3209.	2.8	37
83	Multiparametric Dissolution of Thoriumâ€“Cerium Dioxide Solid Solutions. Inorganic Chemistry, 2011, 50, 11702-11714.	1.9	65
84	Stability and Structural Evolution of Ce^{IV}_{1â€“x}Ln^{III}_xO_{2â€“x/2} Solid Solutions: A Coupled Î¼-Raman/XRD Approach. Inorganic Chemistry, 2011, 50, 7150-7161.	1.9	109
85	Influence of Crystallization State and Microstructure on the Chemical Durability of Ceriumâ€“Neodymium Mixed Oxides. Inorganic Chemistry, 2011, 50, 9059-9072.	1.9	60
86	How To Explain the Difficulties in the Coffinite Synthesis from the Study of Uranothorite?. Inorganic Chemistry, 2011, 50, 11117-11126.	1.9	33
87	Negative thermal expansion in Th2O(PO4)2. Materials Research Bulletin, 2011, 46, 1777-1780.	2.7	14
88	Occurence of an Octanuclear Motif of Uranyl Isophthalate with Cationâ€“Cation Interactions through Edge-Sharing Connection Mode. Inorganic Chemistry, 2011, 50, 6243-6249.	1.9	89
89	Tetrameric entity resulting from two distinct dinuclear uranyl-centered motifs bridged through Î¼2-OH and pyridazine-3,6-dicarboxylate. Inorganic Chemistry Communication, 2011, 14, 429-432.	1.8	24
90	Crystal chemistry of the monazite structure. Journal of the European Ceramic Society, 2011, 31, 941-976.	2.8	318

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91	Preparation of morphology controlled Th1â^'xUxO2 sintered pellets from low-temperature precursors. Powder Technology, 2011, 208, 454-460.	2.1	54
92	Kinetics of dissolution of thorium and uranium doped britholite ceramics. Journal of Nuclear Materials, 2010, 404, 33-43.	1.3	37
93	X-Ray Diffraction and Î¼-Raman Investigation of the Monoclinic-Orthorhombic Phase Transition in Th_{1â^'x}U_x(C₂O₄)₂Â·2H₂O Solid Solutions. Inorganic Chemistry, 2010, 49, 1921-1931.	4.9	60
94	Preparation, sintering and leaching of optimized uranium thorium dioxides. Journal of Nuclear Materials, 2009, 385, 400-406.	1.3	55
95	Synthesis and characterization of coffinite. Journal of Nuclear Materials, 2009, 393, 449-458.	1.3	46
96	Synthesis, Raman and Rietveld analysis of thorium diphosphate. Journal of Solid State Chemistry, 2008, 181, 3352-3356.	1.4	25
97	Preparation of Optimized Uranium and Thorium Bearing Brabantite or Monazite/Brabantite Solid Solutions. Journal of the American Ceramic Society, 2008, 91, 3673-3682.	1.9	50
98	Comparative Behavior of Britholites and Monazite/Brabantite Solid Solutions during Leaching Tests: A Combined Experimental and DFT Approach. Inorganic Chemistry, 2008, 47, 10971-10979.	1.9	56
99	Hydrothermal Method of Preparation of Actinide(IV) Phosphate Hydrogenphosphate Hydrates and Study of Their Conversion into Actinide(IV) Phosphate Diphosphate Solid Solutions. Inorganic Chemistry, 2007, 46, 10390-10399.	1.9	15
100	Actinide solubility-controlling phases during the dissolution of phosphate ceramics. Journal of Nuclear Materials, 2007, 362, 451-458.	1.3	80
101	Crystal structures of Th(OH)PO4, U(OH)PO4 and Th2O(PO4)2. Condensation mechanism of MIV(OH)PO4 (M=Th, U) into M2O(PO4)2. Solid State Sciences, 2007, 9, 619-627.	1.5	41
102	Synthesis, Characterization, Sintering, and Leaching of Î²-TUPD/Monazite Radwaste Matrices. Inorganic Chemistry, 2006, 45, 220-229.	1.9	75
103	Behavior of thoriumâ€“uranium (IV) phosphateâ€“diphosphate sintered samples during leaching tests. Part I â€“ Kinetic study. Journal of Nuclear Materials, 2006, 349, 291-303.	1.3	52
104	Behavior of thoriumâ€“uranium (IV) phosphateâ€“diphosphate sintered samples during leaching tests. Part II. Saturation processes. Journal of Nuclear Materials, 2006, 349, 304-316.	1.3	50
105	Improvement of the preparation of sintered pellets of thorium phosphate-diphosphate and associated solid solutions from crystallized precursors. Journal of Nuclear Materials, 2006, 352, 209-216.	1.3	22
106	From thorium phosphate hydrogenphosphate hydrate to Î²-thorium phosphate diphosphate: Structural evolution to a radwaste storage ceramic. Journal of Solid State Chemistry, 2006, 179, 3007-3016.	1.4	17
107	Investigation in thorium phosphate by NMR II-phosphorus dipolar networks. Solid State Nuclear Magnetic Resonance, 2006, 29, 294-304.	1.5	5
108	Investigation in hydrated thorium phosphates by NMR I-relation proton phosphorus. Solid State Nuclear Magnetic Resonance, 2006, 30, 29-44.	1.5	0

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109	Kinetic and Thermodynamic Study of the Chemistry of Neoformed Phases during the Dissolution of Phosphate Based Ceramics. Materials Research Society Symposia Proceedings, 2006, 985, 1.	0.1	0
110	Separation of uranium(VI) from tri- and tetravalent elements in phosphoric acid solutions. Radiochimica Acta, 2006, 94, .	0.5	0
111	Hydrothermal Methods as a New Way of Actinide Phosphate Preparation. Materials Research Society Symposia Proceedings, 2006, 985, 1.	0.1	2
112	Characterization of the thorium phosphate-hydrogenphosphate hydrate (TPHPH) and study of its transformation into the thorium phosphate-diphosphate (Î²-TPD). Materials Research Bulletin, 2005, 40, 2225-2242.	2.7	43
113	Synthesis and characterization of uranium (IV) phosphate-hydrogenphosphate hydrate and cerium (IV) phosphate-hydrogenphosphate hydrate. Journal of Solid State Chemistry, 2005, 178, 1054-1063.	1.4	39
114	Synthesis and characterization of low-temperature precursors of thoriumâ€“uranium (IV) phosphateâ€“diphosphate solid solutions. Journal of Nuclear Materials, 2004, 335, 397-409.	1.3	21
115	Immobilisation of actinides in phosphate matrices. Comptes Rendus Chimie, 2004, 7, 1141-1152.	0.2	107
116	Sintering of Î²-Thoriumâ^'Uranium(IV) Phosphateâ^'Diphosphate Solid Solutions from Low-Temperature Precursors. Chemistry of Materials, 2004, 16, 3357-3366.	3.2	31
117	Preparation and characterization of lanthanumâ€“gadolinium monazites as ceramics for radioactive waste storage. New Journal of Chemistry, 2003, 27, 957-967.	1.4	142
118	Study of Actinides Incorporation in Thorium Phosphate-Diphosphate/Monazite Based Ceramics. Materials Research Society Symposia Proceedings, 2003, 802, 111.	0.1	2