List of Publications by Year in descending order

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ΟΙ ΑΡΑ ΡΑΓΟΙ Ο

#	Article	IF	CITATIONS
1	Critical micelle concentration of surfactants in aqueous buffered and unbuffered systems. Analytica Chimica Acta, 2005, 548, 95-100.	5.4	317
2	Solute-solvent and solvent-solvent interactions in binary solvent mixtures. Part 7. Comparison of the enhancement of the water structure in alcohol-water mixtures measured by solvatochromic indicators. Journal of Physical Organic Chemistry, 1998, 11, 185-192.	1.9	199
3	Solute–solvent and solvent–solvent interactions in binary solvent mixtures. Part 1. A comparison of several preferential solvation models for describing ET(30) polarity of bipolar hydrogen bond acceptor-cosolvent mixtures. Journal of the Chemical Society Perkin Transactions II, 1995, , 1607-1615.	0.9	195
4	Determination of dissociation constants of flavonoids by capillary electrophoresis. Electrophoresis, 2005, 26, 1886-1895.	2.4	194
5	Static Dielectric Constants of Acetonitrile/Water Mixtures at Different Temperatures and Debyeâ	1.9	139
6	Setup and validation of shake-flask procedures for the determination of partition coefficients (logD) from low drug amounts. European Journal of Pharmaceutical Sciences, 2015, 76, 181-191.	4.0	139
7	Determination of mono- and disulphonated azo dyes by liquid chromatography–atmospheric pressure ionization mass spectrometry. Journal of Chromatography A, 1997, 777, 177-192.	3.7	104
8	Linear free energy relationship analysis of microemulsion electrokinetic chromatographic determination of lipophilicity. Journal of Chromatography A, 1996, 752, 243-249.	3.7	102
9	Solute–solvent and solvent–solvent interactions in binary solvent mixtures. Part 6. A quantitative measurement of the enhancement of the water structure in 2-methylpropan-2-ol–water and propan-2-ol–water mixtures by solvatochromic indicators. Journal of the Chemical Society Perkin Transactions II. 1997 1341-1348.	0.9	97
10	Hydrogen bonding. Part 40. Factors that influence the distribution of solutes between water and sodium dodecylsulfate micelles. Journal of the Chemical Society Perkin Transactions II, 1995, , 887.	0.9	93
11	Chromatographic Estimation of Drug Disposition Properties by Means of Immobilized Artificial Membranes (IAM) and C18 Columns. Journal of Medicinal Chemistry, 2006, 49, 4861-4870.	6.4	92
12	Factors that influence tadpole narcosis. An LFER analysis. Journal of the Chemical Society Perkin Transactions II, 1995, , 1843.	0.9	86
13	Solute–solvent and solvent–solvent interactions in binary solvent mixtures. Part 3. The ET(30) polarity of binary mixtures of hydroxylic solvents. Journal of the Chemical Society Perkin Transactions II, 1996, , 1497-1503.	0.9	85
14	Solute–solvent interactions in micellar electrokinetic chromatography. Journal of Chromatography A, 2002, 942, 237-248.	3.7	85
15	Determination of sulphonated azo dyes in water and wastewater. TrAC - Trends in Analytical Chemistry, 1997, 16, 405-419.	11.4	83
16	Equilibrium solubility measurement of ionizable drugs – consensus recommendations for improving data quality. ADMET and DMPK, 2016, 4, 117.	2.1	78
17	Acidâ^'Base Equilibria in Nonpolar Media. Absolute pKa Scale of Bases in Tetrahydrofuran. Journal of Organic Chemistry, 2006, 71, 9062-9067.	3.2	76
18	δConversion Parameter between pH Scales (and) in Acetonitrile/Water Mixtures at Various Compositions and Temperatures. Analytical Chemistry, 2007, 79, 3180-3187.	6.5	74

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19	Molecular interactions between warfarin and human (HSA) or bovine (BSA) serum albumin evaluated by isothermal titration calorimetry (ITC), fluorescence spectrometry (FS) and frontal analysis capillary electrophoresis (FA/CE). Journal of Pharmaceutical and Biomedical Analysis, 2018, 150, 452-459.	2.8	72
20	Solute–solvent interactions in micellar electrokinetic chromatography. Journal of Chromatography A, 1999, 845, 217-226.	3.7	63
21	A comparison between different approaches to estimate the aqueous pKa values of several non-steroidal anti-inflammatory drugs. Analytica Chimica Acta, 1997, 338, 127-134.	5.4	62
22	Effect of temperature on pH measurements and acid–base equilibria in methanol–water mixtures. Journal of Chromatography A, 2003, 1002, 41-53.	3.7	59
23	Molecular interactions between some non-steroidal anti-inflammatory drugs (NSAID׳s) and bovine (BSA) or human (HSA) serum albumin estimated by means of isothermal titration calorimetry (ITC) and frontal analysis capillary electrophoresis (FA/CE). Talanta, 2014, 130, 241-250.	5.5	59
24	Critical evaluation of buffering solutions for p <i>K</i> _a determination by capillary electrophoresis. Electrophoresis, 2008, 29, 2841-2851.	2.4	54
25	Potentiometric determination of aqueous dissociation constants of flavonols sparingly soluble in water. Talanta, 2008, 74, 1008-1013.	5.5	54
26	Solubility–pH profiles of some acidic, basic and amphoteric drugs. European Journal of Pharmaceutical Sciences, 2013, 48, 291-300.	4.0	54
27	Variation of acidity constants and pH values of some organic acids in water—2-propanol mixtures with solvent composition. Effect of preferential solvation. Analytica Chimica Acta, 1995, 302, 109-119.	5.4	52
28	Hydrogen bonding. Part 41.1 Factors that influence the distribution of solutes between water and hexadecylpyridinium chloride micelles. Journal of the Chemical Society Perkin Transactions II, 1997, , 19-24.	0.9	51
29	Analysis of prominent flavonoid aglycones by high-performance liquid chromatography using a monolithic type column. Journal of Chromatography A, 2006, 1131, 51-57.	3.7	51
30	Selectivity of single, mixed, and modified pseudostationary phases in electrokinetic chromatography. Electrophoresis, 2006, 27, 1900-1914.	2.4	51
31	Solute–solvent and solvent–solvent interactions in binary solvent mixtures. Part 5. Preferential solvation of solvatochromic indicators in mixtures of propan-2-ol with hexane, benzene, ethanol and methanol. Journal of the Chemical Society Perkin Transactions II, 1997, , 243-248.	0.9	50
32	Effect of temperature on the chromatographic retention of ionizable compounds. Journal of Chromatography A, 2004, 1042, 23-36.	3.7	47
33	Correlation of the toxicity of organic compounds to tadpoles using the Abraham model. Science of the Total Environment, 2006, 371, 99-109.	8.0	47
34	Solute-solvent interactions in micellar electrokinetic chromatography: IV. Characterization of electroosmotic flow and micellar markers. Electrophoresis, 2002, 23, 56.	2.4	45
35	A potentially simpler approach to measure aqueous pKa of insoluble basic drugs containing amino groups. Journal of Pharmaceutical Sciences, 2003, 92, 1473-1481.	3.3	44
36	Acidity and Hydrophobicity of Several New Potential Antitubercular Drugs: Isoniazid and Benzimidazole Derivatives. Journal of Chemical & Engineering Data, 2012, 57, 330-338.	1.9	43

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37	Chromatographic Determination of Aqueous Dissociation Constants of Some Water-Insoluble Nonsteroidal Antiinflammatory Drugs. Journal of Pharmaceutical Sciences, 2002, 91, 991-999.	3.3	40
38	Autoprotolysis in aqueous organic solvent mixtures. Analytical Chemistry, 1993, 65, 2294-2299.	6.5	39
39	Acidity constants in methanol/water mixtures of polycarboxylic acids used in drug salt preparations. European Journal of Pharmaceutical Sciences, 2006, 28, 118-127.	4.0	39
40	Fast high-throughput method for the determination of acidity constants by capillary electrophoresis. Journal of Chromatography A, 2009, 1216, 3646-3651.	3.7	39
41	Fast high-throughput method for the determination of acidity constants by capillary electrophoresis. II. Acidic internal standards. Journal of Chromatography A, 2010, 1217, 8340-8345.	3.7	37
42	Characterization of the Solvation Properties of Sodiumn-Dodecyl Sulfate Micelles in Buffered and Unbuffered Aqueous Phases by Solvatochromic Indicators. Langmuir, 2003, 19, 55-62.	3.5	36
43	Critical validation of a new simpler approach to estimate aqueous pKa of drugs sparingly soluble in water. Analytica Chimica Acta, 2005, 550, 210-221.	5.4	34
44	A fast high throughput method for the determination of acidity constants by capillary electrophoresis. 3. Basic internal standards. Journal of Chromatography A, 2011, 1218, 3928-3934.	3.7	34
45	Solute–solvent interactions in micellar electrokinetic chromatography. Journal of Chromatography A, 2001, 907, 257-265.	3.7	33
46	Determination of the hydrophobicity of organic compounds measured as logPo/w through a new chromatographic method. Journal of Chromatography A, 2010, 1217, 3026-3037.	3.7	33
47	lonic equilibria in aqueous organic solvent mixtures the dissociation constants of acids and salts in tetrahydrofuran/water mixtures. Analytica Chimica Acta, 1997, 340, 133-141.	5.4	31
48	Dissociation constants of several non-steroidal anti-inflammatory drugs in isopropyl alcohol/water mixtures. Analytica Chimica Acta, 1997, 350, 249-255.	5.4	31
49	Effect of temperature on the chromatographic retention of ionizable compounds. Journal of Chromatography A, 2005, 1077, 159-169.	3.7	30
50	Characterization of immobilized artificial membrane (IAM) and XTerra columns by means of chromatographic models. Journal of Chromatography A, 2005, 1081, 163-173.	3.7	29
51	Lipophilicity assessment of basic drugs (logPo/w determination) by a chromatographic method. Journal of Chromatography A, 2011, 1218, 6356-6368.	3.7	29
52	Determination of acidity constants by the capillary electrophoresis internal standard method. IV. Polyprotic compounds. Journal of Chromatography A, 2013, 1279, 108-116.	3.7	29
53	Determination of flavonoid aglycones in several food samples by mixed micellar electrokinetic chromatography. Journal of Separation Science, 2007, 30, 2493-2500.	2.5	28
54	Background electrolytes in 50% methanol/water for the determination of acidity constants of basic drugs by capillary zone electrophoresis. Journal of Chromatography A, 2006, 1123, 113-120.	3.7	27

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55	Autoprotolysis in aqueous organic solvent mixtures. Water/alcohol binary systems. Analytica Chimica Acta, 1996, 335, 291-302.	5.4	26
56	Characterization of the Solvation Properties of Surfactants by Solvatochromic Indicators. Langmuir, 2003, 19, 6685-6692.	3.5	26
57	Simultaneous effect of pH, temperature and mobile phase composition in the chromatographic retention of ionizable compounds. Journal of Chromatography A, 2011, 1218, 4995-5009.	3.7	26
58	Evaluation of the suitability of chromatographic systems to predict human skin permeation of neutral compounds. European Journal of Pharmaceutical Sciences, 2013, 50, 557-568.	4.0	26
59	Poly(2-Ethyl-2-Oxazoline) as an Alternative to Poly(Vinylpyrrolidone) in Solid Dispersions for Solubility and Dissolution Rate Enhancement of Drugs. Journal of Pharmaceutical Sciences, 2018, 107, 2428-2438.	3.3	24
60	Modeling Retention and Selectivity as a Function of pH and Column Temperature in Liquid Chromatography. Analytical Chemistry, 2006, 78, 5858-5867.	6.5	23
61	Prediction of retention in reversed-phase liquid chromatography by means of the polarity parameter model. Journal of Chromatography A, 2009, 1216, 5214-5227.	3.7	22
62	Extension of the liquid chromatography/quantitative structure–property relationship method to assess the lipophilicity of neutral, acidic, basic and amphotheric drugs. Journal of Chromatography A, 2012, 1240, 113-122.	3.7	22
63	The Ca2+–EDTA chelation as standard reaction to validate Isothermal Titration Calorimeter measurements (ITC). Talanta, 2016, 154, 354-359.	5.5	22
64	A Novel, Extremely Bioavailable Cocrystal of Pterostilbene. Crystal Growth and Design, 2021, 21, 2315-2323.	3.0	22
65	Micellar electrokinetic chromatography estimation of size and composition of procyanidins after thiolysis with cysteine. Electrophoresis, 2003, 24, 1404-1410.	2.4	21
66	Physicochemical Properties of a New Multicomponent Cosolvent System for the p <i>K</i> _a Determination of Poorly Soluble Pharmaceutical Compounds. Helvetica Chimica Acta, 2007, 90, 1538-1553.	1.6	21
67	Modeling Nonspecific Toxicity of Organic Compounds to the Fathead Minnow Fish by Means of Chromatographic Systems. Analytical Chemistry, 2012, 84, 3446-3452.	6.5	21
68	Critical comparison of shake-flask, potentiometric and chromatographic methods for lipophilicity evaluation (log P o/w) of neutral, acidic, basic, amphoteric, and zwitterionic drugs. European Journal of Pharmaceutical Sciences, 2018, 122, 331-340.	4.0	21
69	Ionic equilibria in neutral amphiprotic solvents: Structural effects on dissociation constants of several substituted phenols and mercaptopyrimidines in isopropyl alcohol. Talanta, 1989, 36, 1227-1231.	5.5	20
70	lonic equilibria in neutral amphiprotic solvents: relationships between electrolyte pK values and solvent polarity and composition for several binary isopropyl alcohol mixtures. Analytical Chemistry, 1990, 62, 102-107.	6.5	20
71	Soluteâ^'Solvent Interactions in Micellar Electrokinetic Chromatography. 6. Optimization of the Selectivity of Lithium Dodecyl Sulfateâ^'Lithium Perfluorooctanesulfonate Mixed Micellar Buffers. Analytical Chemistry, 2002, 74, 4447-4455.	6.5	20
72	Lipophilicity of amphoteric and zwitterionic compounds: A comparative study of determination methods. Talanta, 2017, 162, 293-299.	5.5	20

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73	Chiral separation of benzoporphyrin derivative mono- and diacids by laser induced fluorescence-capillary electrophoresis. Electrophoresis, 2002, 23, 93.	2.4	18
74	Solute-solvent interactions in micellar electrokinetic chromatography: V. Factors that produce peak splitting. Electrophoresis, 2002, 23, 2408-2416.	2.4	18
75	Determination of the chromatographic hydrophobicity index for ionisable solutes. Journal of Chromatography A, 2007, 1173, 110-119.	3.7	18
76	Acidity of Several Anilinium Derivatives in Pure Tetrahydrofuran. Journal of Solution Chemistry, 2008, 37, 689-700.	1.2	18
77	Effect of vinylpyrrolidone polymers on the solubility and supersaturation of drugs; a study using the Cheqsol method. European Journal of Pharmaceutical Sciences, 2018, 117, 227-235.	4.0	18
78	Evaluation of the Interactions between Human Serum Albumin (HSA) and Non-Steroidal Anti-Inflammatory (NSAIDs) Drugs by Multiwavelength Molecular Fluorescence, Structural and Computational Analysis. Pharmaceuticals, 2021, 14, 214.	3.8	18
79	Mixed micellar electrokinetic capillary chromatography separation of depolymerized grape procyanidins. Electrophoresis, 2003, 24, 707-713.	2.4	17
80	A Fast Method for p <i>K</i> _a Determination by Capillary Electrophoresis. Chemistry and Biodiversity, 2009, 6, 1822-1827.	2.1	17
81	Evaluation of log Po/w values of drugs from some molecular structure calculation softwares. ADMET and DMPK, 2014, 2, .	2.1	17
82	Henry's Law constants or air to water partition coefficients for 1,3,5-triazines by an LFER method. Journal of Environmental Monitoring, 2007, 9, 234-239.	2.1	16
83	Estimation of Biological Properties by Means of Chromatographic Systems: Evaluation of the Factors That Contribute to the Variance of Biologicalâ´'Chromatographic Correlations. Analytical Chemistry, 2010, 82, 10236-10245.	6.5	16
84	The contribution of the hydrogen bond acidity on the lipophilicity of drugs estimated from chromatographic measurements. European Journal of Pharmaceutical Sciences, 2013, 48, 484-493.	4.0	16
85	Synthesis and Characterization of a New Norfloxacin/Resorcinol Cocrystal with Enhanced Solubility and Dissolution Profile. Pharmaceutics, 2022, 14, 49.	4.5	16
86	Enthalpies and constants of dissociation of several neutral and cationic acids in aqueous and methanol/water solutions at various temperatures. Journal of Pharmaceutical and Biomedical Analysis, 2009, 49, 923-930.	2.8	15
87	Temperature variation effects on the determination of acidity constants through the internal standard–capillary electrophoresis method. Electrophoresis, 2013, 34, 1203-1211.	2.4	15
88	Phenothiazines solution complexity – Determination of pKa and solubility-pH profiles exhibiting sub-micellar aggregation at 25 and 37°C. European Journal of Pharmaceutical Sciences, 2016, 93, 163-176.	4.0	15
89	Effect of temperature on the chromatographic retention of ionizable compounds. III. Modeling retention of pharmaceuticals as a function of eluent pH and column temperature in RPLC. Journal of Separation Science, 2008, 31, 969-980.	2.5	14
90	Autoprotolysis in aqueous organic solvent mixtures. Water/dipolar protophilic solvent binary systems. Analytica Chimica Acta, 1997, 349, 367-376.	5.4	13

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91	Hydrophobic and cation exchange mechanisms in the retention of basic compounds in a polymeric column. Journal of Chromatography A, 2004, 1028, 139-148.	3.7	13
92	Acidity of several polyprotic acids, amiodarone and quetiapine hemifumarate in pure methanol. Talanta, 2007, 73, 115-120.	5.5	13
93	Performance of chromatographic systems to model soil–water sorption. Journal of Chromatography A, 2012, 1252, 136-145.	3.7	13
94	Acidâ	1.9	12
95	Molecular Details of INH-C ₁₀ Binding to <i>wt</i> KatG and Its S315T Mutant. Molecular Pharmaceutics, 2015, 12, 898-909.	4.6	12
96	Prediction of the n-octanol/water partition coefficients in the SAMPL6 blind challenge from MST continuum solvation calculations. Journal of Computer-Aided Molecular Design, 2020, 34, 443-451.	2.9	11
97	Interpretation of Hydroxylic Solvent Effects Based on Correlations with Solvent Parameters. Reaction of Et3N with Etl. Collection of Czechoslovak Chemical Communications, 1994, 59, 898-904.	1.0	10
98	Ionic equilibria in aqueous organic solvent mixtures The equilibria of HF in an ethanol + water mixture used for cleaning up semiconductors. Journal of Electroanalytical Chemistry, 1997, 433, 77-83.	3.8	10
99	Interaction of Antioxidant Biobased Epicatechin Conjugates with Biomembrane Models. Journal of Agricultural and Food Chemistry, 2007, 55, 2901-2905.	5.2	9
100	Equations for the Correlation and Prediction of Partition Coefficients of Neutral Molecules and Ionic Species in the Water–Isopropanol Solvent System. Journal of Solution Chemistry, 2021, 50, 458-472.	1.2	9
101	Ionizable Drug Self-Associations and the Solubility Dependence on pH: Detection of Aggregates in Saturated Solutions Using Mass Spectrometry (ESI-Q-TOF-MS/MS). Molecular Pharmaceutics, 2021, 18, 2311-2321.	4.6	9
102	Autoprotolysis in aqueous organic solvent mixtures. Water-amide and water-amine binary systems. Analytica Chimica Acta, 1995, 302, 355-363.	5.4	8
103	Chromatographic hydrophobicity index: pH profile for polyprotic compounds. Journal of Chromatography A, 2009, 1216, 7798-7805.	3.7	8
104	Solute–solvent interactions in micellar electrokinetic chromatography: VII. Characterization of sodium cholate–sodium deoxycholate mixed-micellar systems. Journal of Chromatography A, 2010, 1217, 1701-1708.	3.7	8
105	Characterization of the acidity of residual silanol groups in immobilized artificial membranes. Journal of Chromatography A, 2008, 1182, 233-236.	3.7	7
106	Erratum to "Solute–solvent interactions in micellar electrokinetic chromatography. III. Characterization of the selectivity of micellar electrokinetic chromatography systems―[J. Chromatogr. A 942 (2002) 237–248]. Journal of Chromatography A, 2009, 1216, 6877-6879.	3.7	7
107	Dissolution rate of ciprofloxacin and its cocrystal with resorcinol. ADMET and DMPK, 2018, 6, 61.	2.1	7
108	Molecular characteristics of several drugs evaluated from solvent/water partition measurements: Solvation parameters and intramolecular hydrogen bond indicator. European Journal of Pharmaceutical Sciences, 2022, 168, 106066.	4.0	7

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109	Standardization of potentiometric cells in propan-2-ol-water. Analytica Chimica Acta, 1993, 280, 75-83.	5.4	6
110	Kinetic and Thermodynamic Solubility Values of Some Bioactive Compounds. Chemistry and Biodiversity, 2009, 6, 1789-1795.	2.1	6
111	Isothermal titration calorimetry of Ni(II) binding to histidine and to N-2-aminoethylglycine. Talanta, 2011, 84, 347-354.	5.5	6
112	Chromatographic Hydrophobicity Index (CHI). Advances in Chromatography, 2012, 50, 377-414.	1.0	6
113	Evaluation of the interactions between human serum albumin (HSA) and warfarin or diflunisal by using molecular fluorescence using two approaches. ADMET and DMPK, 2018, 6, 47.	2.1	5
114	Capillary electrophoresis for drug analysis and physicochemical characterization. Handbook of Analytical Separations, 2020, , 633-666.	0.8	5
115	Binding thermodynamics of paromomycin, neomycin, neomycinâ€dinucleotide and â€diPNA conjugates to bacterial and human rRNA. Journal of Molecular Recognition, 2016, 29, 142-150.	2.1	3
116	Potentiometric CheqSol and standardized shake-flask solubility methods are complimentary tools in physicochemical profiling. European Journal of Pharmaceutical Sciences, 2020, 148, 105305.	4.0	2
117	Solubility-pH profiles of a free base and its salt: sibutramine as a case study. ADMET and DMPK, 2017, 5, 253-256.	2.1	2
118	Ionic equilibria in aqueous organic solvent mixtures. Speciation of hydrofluoric acid in several ethanol/water solutions. Journal of Electroanalytical Chemistry, 2019, 848, 113318.	3.8	0