

Juha Ahola

List of Publications by Year in descending order

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Version: 2024-02-01

23
papers

663
citations

567281

15
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713466

21
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all docs

23
docs citations

23
times ranked

927
citing authors

#	ARTICLE	IF	CITATIONS
1	Reaction mechanism and kinetics of NO _x reduction by propene on CoO _x /alumina catalysts in lean conditions. <i>Applied Catalysis B: Environmental</i> , 2000, 26, 173-192.	20.2	82
2	Kinetics of Xylose Dehydration into Furfural in Formic Acid. <i>Industrial & Engineering Chemistry Research</i> , 2012, 51, 6297-6303.	3.7	73
3	Kinetics of glucose decomposition in formic acid. <i>Chemical Engineering Research and Design</i> , 2011, 89, 2706-2713.	5.6	59
4	Reaction mechanism and kinetics of NO _x reduction by methane on In/ZSM-5 under lean conditions. <i>Applied Catalysis B: Environmental</i> , 2006, 64, 13-24.	20.2	47
5	Phosphonated nanocelluloses from sequential oxidative-reductive treatment Physicochemical characteristics and thermal properties. <i>Carbohydrate Polymers</i> , 2015, 133, 524-532.	10.2	46
6	Solubility and fractionation of Indulin AT kraft lignin in ethanol-water media. <i>Separation and Purification Technology</i> , 2019, 209, 826-832.	7.9	44
7	Comparison of Lignin Fractions Isolated from Wheat Straw Using Alkaline and Acidic Deep Eutectic Solvents. <i>Journal of Agricultural and Food Chemistry</i> , 2020, 68, 15074-15084.	5.2	36
8	Distinct Effect of Formic and Sulfuric Acids on Cellulose Hydrolysis at High Temperature. <i>Industrial & Engineering Chemistry Research</i> , 2012, 51, 3295-3300.	3.7	33
9	Formic acid aided hot water extraction of hemicellulose from European silver birch (<i>Betula pendula</i>) Tj ETQq1 1 0.784314 rgBT /Over 9.6 32	9.6	32
10	Hydrolysis of organosolv wheat pulp in formic acid at high temperature for glucose production. <i>Bioresource Technology</i> , 2012, 116, 29-35.	9.6	31
11	Investigation of CO oxidation and NO reduction on three-way monolith catalysts with transient response techniques. <i>Applied Catalysis B: Environmental</i> , 1997, 12, 287-308.	20.2	30
12	Comparison of Formic and Sulfuric Acids as a Glucose Decomposition Catalyst. <i>Industrial & Engineering Chemistry Research</i> , 2010, 49, 8444-8449.	3.7	28
13	Acid-catalysed xylose dehydration into furfural in the presence of kraft lignin. <i>Bioresource Technology</i> , 2015, 177, 94-101.	9.6	28
14	Production of levulinic acid from glucose in sulfolane/water mixtures. <i>Chemical Engineering Research and Design</i> , 2019, 148, 291-297.	5.6	25
15	Kinetics of Formic Acid-catalyzed Cellulose Hydrolysis. <i>BioResources</i> , 2014, 9, .	1.0	18
16	Determination of Phenolic Hydroxyl Groups in Technical Lignins by Ionization Difference Ultraviolet Spectrophotometry (ã†µ-IDUS method). <i>Periodica Polytechnica: Chemical Engineering</i> , 0, , .	1.1	16
17	Reaction Mechanism and Microkinetic Model for the Binary Catalyst Combination of In/ZSM-5 and Pt/Al ₂ O ₃ for NO _x Reduction by Methane under Lean Conditions. <i>Industrial & Engineering Chemistry Research</i> , 2007, 46, 2715-2725.	3.7	10
18	Effect of Process Variables on the Solvolysis Depolymerization of Pine Kraft Lignin. <i>Waste and Biomass Valorization</i> , 2020, 11, 3195-3206.	3.4	9

#	ARTICLE	IF	CITATIONS
19	Aqueous phase reforming of birch and pine hemicellulose hydrolysates. <i>Bioresource Technology</i> , 2022, 348, 126809.	9.6	7
20	Integration of in Situ FTIR Studies and Catalyst Activity Measurements in Reaction Kinetic Analysis. <i>Industrial & Engineering Chemistry Research</i> , 2003, 42, 2756-2766.	3.7	5
21	A New Homotopy-Based Strategy for the Robust Determination of All the Feasible Solutions for CSTR Systems. <i>Periodica Polytechnica: Chemical Engineering</i> , 2016, 60, 8-23.	1.1	2
22	Liquid-Liquid Equilibria for the <i>n</i> -Pentyl acetate, <i>n</i> -Hexyl acetate, <i>n</i> -Pentanol, or <i>n</i> -Hexanol + Furfural + Water Systems at 298 and 323 K. <i>Journal of Chemical & Engineering Data</i> , 2021, 66, 210-221.	1.9	2
23	A Simulation Case Study for Bio-based Hydrogen Production from Hardwood Hemicellulose. <i>Computer Aided Chemical Engineering</i> , 2020, 48, 1735-1740.	0.5	0