## Joaquim MarÃ\salo

List of Publications by Year in descending order

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103 papers 2,851 citations

147801 31 h-index 206112 48 g-index

106 all docs

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106 times ranked 1784 citing authors

#	Article	IF	CITATIONS
1	[U(Tp <sup>Me2</sup> ) <sub>2</sub> (bipy)] <sup>+</sup> : A Cationic Uranium(III) Complex with Single-Molecule-Magnet Behavior. Inorganic Chemistry, 2011, 50, 9915-9917.	4.0	119
2	Oxidation Studies of Dipositive Actinide Ions, An2+(An = Th, U, Np, Pu, Am) in the Gas Phase:Â Synthesis and Characterization of the Isolated Uranyl, Neptunyl, and Plutonyl Ions UO22+(g), NpO22+(g), and PuO22+(g). Journal of Physical Chemistry A, 2005, 109, 2768-2781.	2.5	111
3	A new definition of coordination number and its use in lanthanide and actinide coordination and organometallic chemistry. Polyhedron, 1989, 8, 2431-2437.	2.2	105
4	The"Bare―Uranyl(2+) Ion, UO22+. Angewandte Chemie International Edition in English, 1996, 35, 891-894.	4.4	100
5	Single-ion magnet behaviour in [U(TpMe2)2I]. Dalton Transactions, 2012, 41, 13568.	3.3	97
6	Gas-Phase Oxidation Reactions of Neptunium and Plutonium Ions Investigated via Fourier Transform Ion Cyclotron Resonance Mass Spectrometry. Journal of Physical Chemistry A, 2002, 106, 7190-7194.	2.5	90
7	Gas-Phase Energetics of Actinide Oxides: An Assessment of Neutral and Cationic Monoxides and Dioxides from Thorium to Curium. Journal of Physical Chemistry A, 2009, 113, 12599-12606.	2.5	89
8	Gas-Phase Uranyl, Neptunyl, and Plutonyl: Hydration and Oxidation Studied by Experiment and Theory. Inorganic Chemistry, 2012, 51, 6603-6614.	4.0	86
9	FTICR-MS study of the gas-phase thermochemistry of americium oxides. International Journal of Mass Spectrometry, 2003, 228, 457-465.	1.5	63
10	New developments in gas-phase actinide ion chemistry. Coordination Chemistry Reviews, 2006, 250, 776-783.	18.8	63
11	Two-electron versus one-electron reduction of chalcogens by uranium( <scp>iii</scp> ): synthesis of a terminal U( <scp>v</scp> ) persulfide complex. Chemical Science, 2014, 5, 841-846.	7.4	60
12	Reactivity of a Tetrakis(pyrazolyl)borate Oxorhenium Complex. Inorganic Chemistry, 1995, 34, 2113-2120.	4.0	56
13	Infrared Spectra and Quantum Chemical Calculations of the Uranium Carbide Molecules UC and CUC with Triple Bonds. Journal of the American Chemical Society, 2010, 132, 8484-8488.	13.7	55
14	Gas-Phase Synthesis and FT/ICR Mass Spectrometric Characterization of Sandwich Complexes of Sc+, Y+, and Lanthanide lons with 1,3,5-Tri-tert-butylbenzene. Journal of the American Chemical Society, 1994, 116, 8666-8672.	13.7	54
15	Identification of 7,4′-Dihydroxy-5-methoxyflavylium in "Dragon's Blood― To Be or Not To Be an Anthocyanin. Chemistry - A European Journal, 2007, 13, 1417-1422.	3.3	53
16	Gas-Phase Reactions of the Bare Th <sup>2+</sup> and U <sup>2+</sup> Ions with Small Alkanes, CH <sub>4</sub> , C <sub>2</sub> H <sub>6</sub> , and C <sub>3</sub> H <sub>8</sub> : Experimental and Theoretical Study of Elementary Organoactinide Chemistry. Journal of the American Chemical Society, 2011, 133, 1955-1970.	13.7	49
17	On the Upper Limits of Oxidation States in Chemistry. Angewandte Chemie - International Edition, 2018, 57, 3242-3245.	13.8	46
18	On the Origins of Faster Oxo Exchange for Uranyl(V) versus Plutonyl(V). Journal of the American Chemical Society, 2012, 134, 15488-15496.	13.7	45

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19	Gas-Phase Reactions of Hydrocarbons with An+ and AnO+ (An = Th, Pa, U, Np, Pu, Am, Cm):  The Active Role of 5f Electrons in Organoprotactinium Chemistry. Organometallics, 2007, 26, 3947-3956.	2.3	44
20	Uranium(III) Redox Chemistry Assisted by a Hemilabile Bis(phenolate) Cyclam Ligand: Uranium–Nitrogen Multiple Bond Formation Comprising a <i>trans</i> -{RNâ•U(VI)â•NR} <sup>2+</sup> Complex. Inorganic Chemistry, 2015, 54, 9115-9126.	4.0	41
21	Gas-Phase Actinide Ion Chemistry:Â FT-ICR/MS Study of the Reactions of Thorium and Uranium Metal and Oxide Ions with Arenes. Organometallics, 1997, 16, 4581-4588.	2.3	40
22	Synthesis and characterization of rhenium complexes with the stabilizing ligand tetrakis(pyrazol-1-yl)borate. Inorganic Chemistry, 1993, 32, 5114-5118.	4.0	39
23	Formation of some transition metal oxide cluster anions and reactivity towards methanol in the gas phase. International Journal of Mass Spectrometry, 1999, 185-187, 825-835.	1.5	38
24	Diamine Bis(phenolate) as Supporting Ligands in Organoactinide(IV) Chemistry. Synthesis, Structural Characterization, and Reactivity of Stable Dialkyl Derivatives. Organometallics, 2013, 32, 1409-1422.	2.3	38
25	Gas phase actinide ion chemistry: Activation of alkanes and alkenes by thorium cations. International Journal of Mass Spectrometry and Ion Processes, 1996, 157-158, 265-274.	1.8	36
26	Actinide sulfides in the gas phase: experimental and theoretical studies of the thermochemistry of AnS (An = Ac, Th, Pa, U, Np, Pu, Am and Cm). Physical Chemistry Chemical Physics, 2011, 13, 12940.	2.8	36
27	Infrared spectra and quantum chemical calculations of the uranium-carbon molecules UC, CUC, UCH, and U(CC)2. Journal of Chemical Physics, 2011, 134, 244313.	3.0	36
28	Hydration of gas-phase ytterbium ion complexes studied by experiment and theory. Theoretical Chemistry Accounts, 2011, 129, 575-592.	1.4	36
29	Actinide poly(pyrazol-1-yl)borate complexes: synthesis and structure of hydrotris(3,5-dimethylpyrazol-1-yl)boratotrichlorotetrahydrofuran actinide(IV), M[HB(3,5-Me2Pz)3] Cl3(THF) (M=Th and U). Inorganica Chimica Acta, 1987, 132, 137-143.	2.4	33
30	A Mononuclear Uranium(IV) Singleâ€Molecule Magnet with an Azobenzene Radical Ligand. Chemistry - A European Journal, 2015, 21, 17817-17826.	3.3	32
31	Improving the selective extraction of lanthanides by using functionalised ionic liquids. Separation and Purification Technology, 2020, 237, 116354.	7.9	32
32	Oxidation of Gas-Phase Protactinium Ions, Pa+and Pa2+:Â Formation and Properties of PaO22+(g), Protactinyl. Journal of Physical Chemistry A, 2006, 110, 5751-5759.	2.5	31
33	Zero-field slow magnetic relaxation in a uranium(iii) complex with a radical ligand. Chemical Communications, 2014, 50, 10262-10264.	4.1	30
34	Synthesis and Hydrolysis of Uranyl, Neptunyl, and Plutonyl Gas-Phase Complexes Exhibiting Discrete Actinide–Carbon Bonds. Organometallics, 2016, 35, 1228-1240.	2.3	30
35	Determination of the ionization energy of NpO2 and comparative ionization energies of actinide oxides. Journal of Nuclear Materials, 2005, 344, 24-29.	2.7	29
36	Gas-Phase Reactions of Doubly Charged Lanthanide Cations with Alkanes and Alkenes. Trends in Metal(2+) Reactivity. Journal of Physical Chemistry A, 2008, 112, 12647-12656.	2.5	29

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37	Gas-phase reactions of lanthanide cations with alcohols. International Journal of Mass Spectrometry, 2004, 234, 51-61.	1.5	28
38	Magnetic Properties of the Layered Lanthanide Hydroxide Series YxDy8-x(OH)20Cl4·6H2O: From Single Ion Magnets to 2D and 3D Interaction Effects. Inorganic Chemistry, 2015, 54, 1949-1957.	4.0	28
39	Gas-Phase Reactions of Uranate lons, UO <sub>2</sub> <sup>â^'</sup> , UO <sub>3</sub> <sup>â^'</sup> , UO <sub>4</sub> +, with Methanol: a Convergence of Experiment and Theory. Inorganic Chemistry, 2010, 49, 3836-3850.	4.0	27
40	Crystal structure diversity in the bis[hydrotris(3,5-dimethylpyrazolyl)borate]iodouranium(iii) complex: from neutral to cationic forms. Dalton Transactions, 2013, 42, 8861.	3.3	26
41	Gas Phase Chemistry of Bis(pentamethylcyclopentadienyl)samarium. Organometallics, 1996, 15, 345-349.	2.3	24
42	Reactivity of Lanthanide, Group 2, and Group 3 Metal and Metal Oxide Cations with Pentamethylcyclopentadiene: A Gas-Phase Synthesis of Cyclopentadienyl Cations. Organometallics, 1997, 16, 3845-3850.	2.3	24
43	Actinide-Transition Metal Heteronuclear Ions and Their Oxides: {IrUO}+ as an Analogue to Uranyl. European Journal of Inorganic Chemistry, 2006, 2006, 3346-3349.	2.0	24
44	Laser desorption Fourier transform mass spectrometric analysis of organoactinides: uranium and thorium polypyrazolylborates. Organometallics, 1991, 10, 2794-2797.	2.3	23
45	Synthesis, characterization and reactivity of lantahnide(II) poly(pyrazol-1-yl)borates (Lnî—»Sm, Eu and Yb); fluorescence studies of $[EuL2(THF)2]$ $[Lî—»B(pz)4, HB(pz)3]$ ; X-ray crystal structures of $[Eu\{B(pz)4\}2(THF)2]$		

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55	Oxo-Exchange of Gas-Phase Uranyl, Neptunyl, and Plutonyl with Water and Methanol. Inorganic Chemistry, 2014, 53, 2163-2170.	4.0	19
56	A novel samarium( <scp>ii</scp> ) complex bearing a dianionic bis(phenolate) cyclam ligand: synthesis, structure and electron-transfer reactions. Dalton Transactions, 2016, 45, 3778-3790.	3.3	19
57	Thermal stability and specific heats of coordinating ionic liquids. Thermochimica Acta, 2020, 684, 178482.	2.7	19
58	Some reactions of hydrotris(3,5-dimethylpyrazolyl)-borato trichloroactinides(IV), MCl3(HBL3)·THF (M ≡) Tj 219-224.	ETQq0 0 0 0.8	) rgBT /Overlo 18
59	Gas phase reactivity of rare earth metal cations with trialkylorthoformates: synthesis of neutral rare earth alkoxides. International Journal of Mass Spectrometry, 2000, 195-196, 139-148.	1.5	17
60	The uranium–nitrogen bond in U(iv) complexes supported by the hydrotris(3,5-dimethylpyrazolyl)borate ligand. Dalton Transactions, 2005, , 3353.	3.3	17
61	Molecular Uranates: Laser Synthesis of Uranium Oxide Anions in the Gas Phase. Inorganic Chemistry, 2009, 48, 5055-5057.	4.0	17
62	Thorium and Uranium Carbide Cluster Cations in the Gas Phase: Similarities and Differences between Thorium and Uranium. Inorganic Chemistry, 2013, 52, 10968-10975.	4.0	16
63	CO <sub>2</sub> conversion to phenyl isocyanates by uranium( <scp>vi</scp> ) bis(imido) complexes. Chemical Communications, 2020, 56, 431-434.	4.1	16
64	Bis[hydrotris(pyrazolyl)borato[dichloroactinide(IV) complexes: X-ray crystal structures of ThCl2(HBPz3)2 and UCl2(HBPz3)2. Polyhedron, 1990, 9, 1645-1652.	2.2	15
65	Über Oxidationszahlâ€Obergrenzen in der Chemie. Angewandte Chemie, 2018, 130, 3297-3300.	2.0	15
66	Pentavalent Curium, Berkelium, and Californium in Nitrate Complexes: Extending Actinide Chemistry and Oxidation States. Inorganic Chemistry, 2018, 57, 9453-9467.	4.0	15
67	Alkoxide and aryloxide derivatives of actinide(IV) polypyrazolylborates. Part I. Uranium(IV) and thorium(IV) hydrotris(3,5-dimethylpyrazol-1-yl)borate complexes. Inorganica Chimica Acta, 1987, 134, 309-314.	2.4	14
68	Rare earth metal complexes anchored on a new dianionic bis(phenolate)dimethylamineCyclam ligand. Journal of Organometallic Chemistry, 2013, 728, 57-67.	1.8	14
69	Synthesis and Properties of Uranium Sulfide Cations. An Evaluation of the Stability of Thiouranyl, {Sâ•Uâ•6} <sup>2+</sup> . Inorganic Chemistry, 2013, 52, 14162-14167.	4.0	14
70	Metal-organic frameworks based on uranyl and phosphonate ligands. Acta Crystallographica Section B: Structural Science, Crystal Engineering and Materials, 2014, 70, 28-36.	1.1	14
71	Revealing Disparate Chemistries of Protactinium and Uranium. Synthesis of the Molecular Uranium Tetroxide Anion, UO <sub>4</sub> <sup>â€"</sup> . Inorganic Chemistry, 2017, 56, 3686-3694.	4.0	14
72	Synthesis and characterization of polynuclear lanthanide aryloxides. Journal of Alloys and Compounds, 2001, 323-324, 169-172.	5 <b>.</b> 5	13

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<b>7</b> 3	Europium(II) and ytterbium(II) aryloxide chemistry: synthesis and crystal structure of [Eu(OC6H3But2-2,6)2(THF)3]·0.75C7H8 and [Yb(OC6H3But2-2,6)2(NCMe)4]. Polyhedron, 2003, 22, 1425-14	29 <del>2.2</del>	12
74	Gas-phase reactions of doubly charged actinide cations with alkanes and alkenesâ€"probing the chemical activity of 5f electrons from Th to Cm. Physical Chemistry Chemical Physics, 2011, 13, 18322.	2.8	12
<b>7</b> 5	Alkoxide and aryloxide derivatives of actinide(IV) polypyrazolylborates. Part II. Uranium(IV) bis[hydrotris(pyrazol-1-yl)borate] complexes. Inorganica Chimica Acta, 1987, 134, 315-320.	2.4	11
76	Dissociation of Gas-Phase Bimetallic Clusters as a Probe of Charge Densities: The Effective Charge of Uranyl. Journal of Physical Chemistry A, 2014, 118, 2159-2166.	2.5	11
77	Gas-Phase Reaction Studies of Dipositive Hafnium and Hafnium Oxide lons: Generation of the Peroxide HfO <sub>2</sub> <sup>2+</sup> . Journal of Physical Chemistry A, 2012, 116, 12399-12405.	2.5	10
78	A 2D Layered Lanthanide Hydroxide Showing Slow Relaxation of Magnetization – Dy <sub>8</sub> (OH) <sub>20</sub> Cl <sub>4</sub> ·6H <sub>2</sub> O. European Journal of Inorganic Chemistry, 2013, 2013, 5059-5063.	2.0	10
79	Molecular Spectroscopy and Reactions of Actinides in the Gas Phase and Cryogenic Matrices. , 2010, , 4079-4156.		10
80	U(IV) and Th(IV) hydrotris(3,5-dimethylpyrazolyl)borate complexes with asymmetric metal centres. Inorganica Chimica Acta, 1987, 139, 83-85.	2.4	9
81	Comment on "Controversy on the First Ionization Potential of PuO2 (Nearly) Settled by New Experimental Evidence― Journal of Physical Chemistry A, 2006, 110, 4131-4132.	2.5	8
82	Chemical evidence of the stability of praseodymium( $\nu$ ) in gas-phase oxide nitrate complexes. Chemical Communications, 2019, 55, 14139-14142.	4.1	8
83	Synthesis and characterization of UCl2[HB(3,5-Me2pz)3](3,5-Me2pz) and MCl3[HB(3,5-Me2pz)3](3,5-Me2pzH) (M=U(IV) and Th(IV); pz=pyrazolyl). Inorganica Chimica Acta, 1987, 139, 79-81.	2.4	7
84	Hydrotris(3,5-dimethylpyrazol-1-yl)borate carboxylate complexes of uranium and thorium. X-ray crystal structure of U(O2CCH3)3[HB(3,5-Me2Pz)3]. Polyhedron, 1992, 11, 501-506.	2.2	7
85	Gas-Phase Oxidation Reactions of Ta <sup>2+</sup> : Synthesis and Properties of TaO <sup>2+</sup> and TaO <sub>2</sub> <sup>2+</sup> . Journal of Physical Chemistry A, 2012, 116, 3534-3540.	2.5	7
86	Uranium(III, IV) and thorium(IV) pyrazolylmethane complexes: Synthesis and structures. Inorganica Chimica Acta, 2012, 385, 53-57.	2.4	7
87	Synthesis, structure and bonding of actinide disulphide dications in the gas phase. Physical Chemistry Chemical Physics, 2017, 19, 10685-10694.	2.8	7
88	Experimental and Computational Study of a Tetraazamacrocycle Bis(aryloxide) Uranyl Complex and of the Analogues {Eâ•Uâ•NR} <sup>2+</sup> (E = O and NR). Inorganic Chemistry, 2022, 61, 346-356.	4.0	6
89	Oxidation of Actinyl(V) Complexes by the Addition of Nitrogen Dioxide Is Revealed via the Replacement of Acetate by Nitrite. Inorganic Chemistry, 2015, 54, 8755-8760.	4.0	5
90	Synthesis of 5H-chromeno[3,4-b]pyridines via DABCO-catalyzed [3 + 3] annulation of 3-nitro-2H-chromenes and allenoates. Organic and Biomolecular Chemistry, 2021, 19, 9711-9722.	2.8	5

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91	A magnetic study of a layered lanthanide hydroxide family: Ln8(OH)20Cl4·nH2O (Ln = Tb, Ho, Er). Dalton Transactions, 2018, 47, 16211-16217.	3.3	4
92	Synthesis and structural characterization of polynuclear divalent ytterbium complexes supported by a bis(phenolate) cyclam ligand. Polyhedron, 2016, 119, 277-285.	2.2	3
93	Gas-Phase Ion Chemistry of Rare Earths and Actinides. Fundamental Theories of Physics, 2014, 45, 1-110.	0.3	2
94	Amavadin and Homologues as Mediators of Water Oxidation. Angewandte Chemie, 2016, 128, 1511-1514.	2.0	2
95	A new krypton complex – experimental and computational investigation of the krypton sulphur pentafluoride cation, [KrSF <sub>5</sub> ] <sup>+</sup> , in the gas phase. Physical Chemistry Chemical Physics, 2022, 24, 14631-14639.	2.8	2
96	Infrared Spectra of Rh12C and Rh13C in Solid Neon and Solid Argon. Chemical Physics Letters, 2012, 528, 7-10.	2.6	1
97	Crystal structure of bis[1-{(3,5-dimethyl-1H-pyrazol-1-yl)methyl}-3,5-dimethyl-1H-pyrazol-2-ium] hexachlorouranate(IV): [H2C(3,5-Me2pz)(3,5-Me2pzH)]2[UCl6]. Journal of Structural Chemistry, 2015, 56, 181-185.	1.0	1
98	Evidence for cathodic electrodeposition of radon species. Journal of Radioanalytical and Nuclear Chemistry, 1984, 86, 373-378.	1.5	0
99	Thermochemistry of Transuranium Actinide Oxide Molecules Investigated by FTICR-MS. AIP Conference Proceedings, 2003, , .	0.4	0
100	Synthesis and characterization of samarium, europium and ytterbium aryloxides. Journal of Alloys and Compounds, 2004, 374, 289-292.	5.5	0
101	Thermochemistry of Elementary Actinide Sulfide Molecules: A Gas-Phase Study of Curium Sulfide. Materials Research Society Symposia Proceedings, 2010, 1264, 1.	0.1	0
102	Preparation of dense 13C pellets using spark plasma sintering technique. Materials Research Innovations, 2013, 17, 289-292.	2.3	0
103	Corrigendum to "Thermal stability and specific heats of coordinating ionic liquids―[Thermochim. Acta 684 (2020) 178482]. Thermochimica Acta, 2020, 685, 178537.	2.7	0