

Min Sik Kim

List of Publications by Year in descending order

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Version: 2024-02-01

22
papers

1,029
citations

516710

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docs citations

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times ranked

1105
citing authors

| # | ARTICLE | IF | CITATIONS |
|----|--|------|-----------|
| 1 | Prediction of Oxidant Exposures and Micropollutant Abatement during Ozonation Using a Machine Learning Method. <i>Environmental Science & Technology</i> , 2021, 55, 709-718. | 10.0 | 21 |
| 2 | Chloride-Mediated Enhancement in Heat-Induced Activation of Peroxymonosulfate: New Reaction Pathways for Oxidizing Radical Production. <i>Environmental Science & Technology</i> , 2021, 55, 5382-5392. | 10.0 | 86 |
| 3 | Effect of Fe ³⁺ as an electron-transfer mediator on WO ₃ -induced activation of peroxymonosulfate under visible light. <i>Chemical Engineering Journal</i> , 2021, 411, 128529. | 12.7 | 19 |
| 4 | Degradation of aqueous organic pollutants using an Fe ₂ O ₃ /WO ₃ composite photocatalyst as a magnetically separable peroxymonosulfate activator. <i>Separation and Purification Technology</i> , 2021, 267, 118610. | 7.9 | 19 |
| 5 | Occurrence of unknown reactive species in UV/H ₂ O ₂ system leading to false interpretation of hydroxyl radical probe reactions. <i>Water Research</i> , 2021, 201, 117338. | 11.3 | 18 |
| 6 | Nonradical activation of peroxymonosulfate by hematite for oxidation of organic compounds: A novel mechanism involving high-valent iron species. <i>Chemical Engineering Journal</i> , 2021, 426, 130743. | 12.7 | 42 |
| 7 | Modeling of ozone decomposition, oxidant exposures, and the abatement of micropollutants during ozonation processes. <i>Water Research</i> , 2020, 169, 115230. | 11.3 | 31 |
| 8 | Reduction of chlorendic acid by zero-valent iron: Kinetics, products, and pathways. <i>Journal of Hazardous Materials</i> , 2020, 384, 121269. | 12.4 | 6 |
| 9 | Cupric ion in combination with hydrogen peroxide and hydroxylamine applied to inactivation of different microorganisms. <i>Journal of Hazardous Materials</i> , 2020, 400, 123305. | 12.4 | 10 |
| 10 | Accelerated oxidation of microcystin-LR by Fe(II)-tetrapolyphosphate/oxygen in the presence of magnesium and calcium ions. <i>Water Research</i> , 2020, 184, 116172. | 11.3 | 2 |
| 11 | Freezing-enhanced non-radical oxidation of organic pollutants by peroxymonosulfate. <i>Chemical Engineering Journal</i> , 2020, 388, 124226. | 12.7 | 17 |
| 12 | Visible light-induced activation of peroxymonosulfate in the presence of ferric ions for the degradation of organic pollutants. <i>Separation and Purification Technology</i> , 2020, 240, 116620. | 7.9 | 27 |
| 13 | Differential Microbicidal Effects of Bimetallic Iron-Copper Nanoparticles on <i>Escherichia coli</i> and MS2 Coliphage. <i>Environmental Science & Technology</i> , 2019, 53, 2679-2687. | 10.0 | 31 |
| 14 | Ozonation of Microcystins: Kinetics and Toxicity Decrease. <i>Environmental Science & Technology</i> , 2019, 53, 6427-6435. | 10.0 | 17 |
| 15 | Comment on "Investigation of the Iron-Peroxo Complex in the Fenton Reaction: Kinetic Indication, Decay Kinetics, and Hydroxyl Radical Yields". <i>Environmental Science & Technology</i> , 2018, 52, 4481-4482. | 10.0 | 1 |
| 16 | Oxidation of Microcystins by Permanganate: pH and Temperature-Dependent Kinetics, Effect of DOM Characteristics, and Oxidation Mechanism Revisited. <i>Environmental Science & Technology</i> , 2018, 52, 7054-7063. | 10.0 | 39 |
| 17 | Oxidation of microcystin-LR by ferrous-tetrapolyphosphate in the presence of oxygen and hydrogen peroxide. <i>Water Research</i> , 2017, 114, 277-285. | 11.3 | 34 |
| 18 | Nanoparticulate zero-valent iron coupled with polyphosphate: the sequential redox treatment of organic compounds and its stability and bacterial toxicity. <i>Environmental Science: Nano</i> , 2017, 4, 396-405. | 4.3 | 10 |

| # | ARTICLE | IF | CITATIONS |
|----|--|------|-----------|
| 19 | Control of the red tide dinoflagellate <i>Cochlodinium polykrikoides</i> by ozone in seawater. <i>Water Research</i> , 2017, 109, 237-244. | 11.3 | 15 |
| 20 | Disintegration of Waste Activated Sludge by Thermally-Activated Persulfates for Enhanced Dewaterability. <i>Environmental Science & Technology</i> , 2016, 50, 7106-7115. | 10.0 | 223 |
| 21 | Oxidative treatment of waste activated sludge by different activated persulfate systems for enhancing sludge dewaterability. <i>Sustainable Environment Research</i> , 2016, 26, 177-183. | 4.2 | 41 |
| 22 | Use of CaO as an activator for producing a price-competitive non-cement structural binder using ground granulated blast furnace slag. <i>Cement and Concrete Research</i> , 2013, 54, 208-214. | 11.0 | 320 |