

Ml Blazquez

List of Publications by Year in descending order

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Version: 2024-02-01

40
papers

3,192
citations

218677

26
h-index

302126

39
g-index

40
all docs

40
docs citations

40
times ranked

3122
citing authors

#	ARTICLE	IF	CITATIONS
1	Sorption and desorption of Cd, Cu and Pb using biomass from an eutrophized habitat in monometallic and bimetallic systems. <i>Journal of Environmental Management</i> , 2011, 92, 2666-2674.	7.8	38
2	Studies on sorption, desorption, regeneration and reuse of sugar-beet pectin gels for heavy metal removal. <i>Journal of Hazardous Materials</i> , 2010, 178, 243-248.	12.4	71
3	Mobilization of phosphorus from iron ore by the bacterium <i>Burkholderia caribensis</i> FeGLO3. <i>Minerals Engineering</i> , 2009, 22, 1-9.	4.3	74
4	Optimization of the continuous biosorption of copper with sugar-beet pectin gels. <i>Journal of Environmental Management</i> , 2009, 90, 1737-1743.	7.8	30
5	Biosorption of cadmium, lead and copper with calcium alginate xerogels and immobilized <i>Fucus vesiculosus</i> . <i>Journal of Hazardous Materials</i> , 2009, 163, 555-562.	12.4	163
6	Gold(III) biosorption and bioreduction with the brown alga <i>Fucus vesiculosus</i> . <i>Journal of Hazardous Materials</i> , 2009, 166, 612-618.	12.4	304
7	Sugar-beet pulp pectin gels as biosorbent for heavy metals: Preparation and determination of biosorption and desorption characteristics. <i>Chemical Engineering Journal</i> , 2009, 150, 289-301.	12.7	171
8	Characterization of the biosorption of cadmium, lead and copper with the brown alga <i>Fucus vesiculosus</i> . <i>Journal of Hazardous Materials</i> , 2008, 158, 316-323.	12.4	143
9	Study of cadmium, zinc and lead biosorption by orange wastes using the subsequent addition method. <i>Bioresource Technology</i> , 2008, 99, 8101-8106.	9.6	73
10	Leaching of chalcopyrite with ferric ion. Part IV: The role of redox potential in the presence of mesophilic and thermophilic bacteria. <i>Hydrometallurgy</i> , 2008, 93, 106-115.	4.3	88
11	Leaching of chalcopyrite with ferric ion. Part I: General aspects. <i>Hydrometallurgy</i> , 2008, 93, 81-87.	4.3	288
12	Leaching of chalcopyrite with ferric ion. Part II: Effect of redox potential. <i>Hydrometallurgy</i> , 2008, 93, 88-96.	4.3	187
13	Bioleaching of a chalcopyrite concentrate with moderate thermophilic microorganisms in a continuous reactor system. <i>Hydrometallurgy</i> , 2007, 87, 100-111.	4.3	50
14	Comparative study of biosorption of heavy metals using different types of algae. <i>Bioresource Technology</i> , 2007, 98, 3344-3353.	9.6	497
15	Biosorption of heavy metals by activated sludge and their desorption characteristics. <i>Journal of Environmental Management</i> , 2007, 84, 419-426.	7.8	203
16	Characterization of brushite as a re-crystallization product formed during bacterial solubilization of hydroxyapatite in batch cultures. <i>Soil Biology and Biochemistry</i> , 2006, 38, 2645-2654.	8.8	94
17	Factors affecting the transformation of a pyritic tailing: scaled-up column tests. <i>Journal of Hazardous Materials</i> , 2005, 118, 35-43.	12.4	5
18	Pyrite behaviour in a tailings pond. <i>Hydrometallurgy</i> , 2005, 76, 25-36.	4.3	26

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19	Biosorption of heavy metals by chemically-activated alga <i>Fucus vesiculosus</i> . <i>Journal of Chemical Technology and Biotechnology</i> , 2005, 80, 1403-1407.	3.2	53
20	Simultaneous uptake of metals by activated sludge. <i>Minerals Engineering</i> , 2003, 16, 723-729.	4.3	53
21	Inhibition of acid rock drainage from uranium ore waste using a conventional neutralization and precipitation treatment. <i>Minerals Engineering</i> , 2002, 15, 1141-1150.	4.3	4
22	Reactivity of a molybdenite concentrate against chemical or bacterial attack. <i>Minerals Engineering</i> , 2001, 14, 987-996.	4.3	21
23	Bioremediation of an industrial acid mine water by metal-tolerant sulphate-reducing bacteria. <i>Minerals Engineering</i> , 2001, 14, 997-1008.	4.3	105
24	Silver-catalysed bioleaching of a chalcopyrite concentrate with mixed cultures of moderately thermophilic microorganisms. <i>Hydrometallurgy</i> , 1999, 51, 37-46.	4.3	69
25	Leaching capacity of a new extremely thermophilic microorganism, <i>Sulfolobus rivotincti</i> . <i>Hydrometallurgy</i> , 1999, 52, 349-366.	4.3	33
26	Bioleaching behaviour of chalcopyrite in the presence of silver at 35Å° and 68Å°C. <i>Process Metallurgy</i> , 1999, , 137-147.	0.1	6
27	Influence of bacteria and sulphite ions on the transformation of pyritic tailings: shake flask tests. <i>Process Metallurgy</i> , 1999, , 537-545.	0.1	0
28	SEM and AES studies of chalcopyrite bioleaching in the presence of catalytic ions. <i>Minerals Engineering</i> , 1997, 10, 825-835.	4.3	13
29	Influence of various factors in the bioleaching of a bulk concentrate with mesophilic microorganisms in the presence of Ag(I). <i>Hydrometallurgy</i> , 1997, 45, 271-287.	4.3	9
30	Chemical and microbiological transformations in a pyritic tailing pond. <i>Minerals Engineering</i> , 1996, 9, 1127-1142.	4.3	10
31	Electrochemistry of chalcopyrite. <i>Hydrometallurgy</i> , 1996, 43, 331-344.	4.3	63
32	A study of the bioleaching of a Spanish uranium ore. Part I: A review of the bacterial leaching in the treatment of uranium ores. <i>Hydrometallurgy</i> , 1995, 38, 39-57.	4.3	65
33	Influence of metallic ions in the bioleaching of chalcopyrite by <i>Sulfolobus BC</i> : Experiments using pneumatically stirred reactors and massive samples. <i>Minerals Engineering</i> , 1995, 8, 949-965.	4.3	19
34	SEM and AES studies of a lead sulphide bioleaching in presence of catalytic ions. <i>Minerals Engineering</i> , 1995, 8, 1503-1512.	4.3	9
35	The catalytic effect of some cations on the biological leaching of a Spanish complex sulphide. <i>Hydrometallurgy</i> , 1993, 34, 151-169.	4.3	26
36	The use of catalytic ions in bioleaching. <i>Hydrometallurgy</i> , 1992, 29, 145-160.	4.3	46

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37	The influence of various ions in the bioleaching of metal sulphides. <i>Hydrometallurgy</i> , 1990, 23, 221-235.	4.3	39
38	Influence of the composition and maximum cycling temperature on the microstructure of Cu-Al-Mn shape memory alloys. <i>Metallography</i> , 1989, 23, 119-133.	0.4	10
39	Studies of zinc sulphide, treated with different solutions of catalyst ions. <i>Vacuum</i> , 1989, 39, 663-664.	3.5	3
40	The influence of composition and grain size on the martensitic transformation temperatures of Cu-Al-Mn shape memory alloys. <i>Scripta Metallurgica</i> , 1987, 21, 1711-1716.	1.2	31