

# JosÃ© H Zagal

## List of Publications by Year in descending order

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101  
papers

5,208  
citations

81900

39  
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88630

70  
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102  
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102  
docs citations

102  
times ranked

3793  
citing authors

#	ARTICLE	IF	CITATIONS
1	Activity volcano plots for the oxygen reduction reaction using FeN <sub>4</sub> complexes: From reported experimental data to the electrochemical meaning. <i>Current Opinion in Electrochemistry</i> , 2022, 32, 100923.	4.8	12
2	Fe <sub>3</sub> O <sub>4</sub> Templated Pyrolyzed Fe~N~C Catalysts. Understanding the role of N~Functions and Fe <sub>3</sub> C on the ORR Activity and Mechanism. <i>ChemElectroChem</i> , 2022, 9, .	3.4	6
3	Strategies to improve the catalytic activity and stability of bioinspired Cu molecular catalysts for the ORR. <i>Current Opinion in Electrochemistry</i> , 2022, 35, 101035.	4.8	6
4	Penta-coordinated transition metal macrocycles as electrocatalysts for the oxygen reduction reaction. <i>Journal of Solid State Electrochemistry</i> , 2021, 25, 15-31.	2.5	22
5	Transition metal phthalocyanine-modified shungite-based cathode catalysts for alkaline membrane fuel cell. <i>International Journal of Hydrogen Energy</i> , 2021, 46, 4365-4377.	7.1	36
6	Insights into the electronic structure of Fe penta-coordinated complexes. Spectroscopic examination and electrochemical analysis for the oxygen reduction and oxygen evolution reactions. <i>Journal of Materials Chemistry A</i> , 2021, 9, 23802-23816.	10.3	27
7	Mapping transition metal-MN <sub>4</sub> macrocyclic complex catalysts performance for the critical reactivity descriptors. <i>Current Opinion in Electrochemistry</i> , 2021, 27, 100683.	4.8	36
8	Mapping transition metal~nitrogen~carbon catalyst~performance on the critical descriptor~diagram. <i>Current Opinion in Electrochemistry</i> , 2021, 27, 100687.	4.8	34
9	Mapping experimental and theoretical reactivity descriptors of Fe macrocyclic complexes deposited on graphite or on multi walled carbon nanotubes for the oxidation of thiols: Thioglycolic acid oxidation. <i>Electrochimica Acta</i> , 2021, 391, 138905.	5.2	5
10	Enhancing the electrocatalytic activity of Fe phthalocyanines for the oxygen reduction reaction by the presence of axial ligands: Pyridine-functionalized single-walled carbon nanotubes. <i>Electrochimica Acta</i> , 2021, 398, 139263.	5.2	27
11	Effect of pH on the Electrochemical Behavior of Hydrogen Peroxide in the Presence of <i>Pseudomonas aeruginosa</i> . <i>Frontiers in Bioengineering and Biotechnology</i> , 2021, 9, 749057.	4.1	2
12	Electrocatalytic oxygen reduction reaction on iron phthalocyanine-modified carbide-derived carbon/carbon nanotube composite electrocatalysts. <i>Electrochimica Acta</i> , 2020, 334, 135575.	5.2	50
13	Experimental reactivity descriptors of M-N-C catalysts for the oxygen reduction reaction. <i>Electrochimica Acta</i> , 2020, 332, 135340.	5.2	42
14	Influence of cyano substituents on the electron density and catalytic activity towards the oxygen reduction reaction for iron phthalocyanine. The case for Fe(II) 2,3,9,10,16,17,23,24-octa(cyano)phthalocyanine. <i>Electrochemistry Communications</i> , 2020, 118, 106784.	4.7	20
15	Reactivity descriptors for Cu bis-phenanthroline catalysts for the hydrogen peroxide reduction reaction. <i>Electrochimica Acta</i> , 2020, 357, 136881.	5.2	9
16	Redox Potentials as Reactivity Descriptors in Electrochemistry. , 2020, , .		0
17	Oxygen Reduction Reaction at Penta-Coordinated Co Phthalocyanines. <i>Frontiers in Chemistry</i> , 2020, 8, 22.	3.6	37
18	Theoretical and Experimental Reactivity Predictors for the Electrocatalytic Activity of Copper Phenanthroline Derivatives for the Reduction of Dioxygen. <i>Journal of Physical Chemistry C</i> , 2019, 123, 19468-19478.	3.1	18

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19	Molecular conductance versus inductive effects of axial ligands on the electrocatalytic activity of self-assembled iron phthalocyanines: The oxygen reduction reaction. <i>Electrochimica Acta</i> , 2019, 327, 134996.	5.2	14
20	Electroreduction of oxygen in alkaline solution on iron phthalocyanine modified carbide-derived carbons. <i>Electrochimica Acta</i> , 2019, 299, 999-1010.	5.2	34
21	Elucidating the mechanism of the oxygen reduction reaction for pyrolyzed Fe-N-C catalysts in basic media. <i>Electrochemistry Communications</i> , 2019, 102, 78-82.	4.7	51
22	In search of the most active MN <sub>4</sub> catalyst for the oxygen reduction reaction. The case of perfluorinated Fe phthalocyanine. <i>Journal of Materials Chemistry A</i> , 2019, 7, 24776-24783.	10.3	52
23	Electrocatalytic oxygen reduction on transition metal macrocyclic complexes for anion exchange membrane fuel cell application. <i>Current Opinion in Electrochemistry</i> , 2018, 9, 207-213.	4.8	44
24	Biomimicking vitamin B12. A Co phthalocyanine pyridine axial ligand coordinated catalyst for the oxygen reduction reaction. <i>Electrochimica Acta</i> , 2018, 265, 547-555.	5.2	56
25	Probing the Fe <sup>2+</sup> /Fe <sup>(n-1)+</sup> redox potential of Fe phthalocyanines and Fe porphyrins as a reactivity descriptor in the electrochemical oxidation of cysteamine. <i>Journal of Electroanalytical Chemistry</i> , 2018, 819, 502-510.	3.8	22
26	Building Pyridinium Molecular Wires as Axial Ligands for Tuning the Electrocatalytic Activity of Iron Phthalocyanines for the Oxygen Reduction Reaction. <i>ACS Catalysis</i> , 2018, 8, 8406-8419.	11.2	57
27	Biomimetic reduction of O <sub>2</sub> in an acid medium on iron phthalocyanines axially coordinated to pyridine anchored on carbon nanotubes. <i>Journal of Materials Chemistry A</i> , 2017, 5, 12054-12059.	10.3	76
28	Oxygen Electroreduction on Zinc and Dilithium Phthalocyanine Modified Multiwalled Carbon Nanotubes in Alkaline Media. <i>Journal of the Electrochemical Society</i> , 2017, 164, H338-H344.	2.9	11
29	Tailoring electroactive surfaces by non-template molecular assembly. Towards electrooxidation of L-cysteine. <i>Electrochimica Acta</i> , 2017, 254, 201-213.	5.2	2
30	Comparison of the catalytic activity for O <sub>2</sub> reduction of Fe and Co MN <sub>4</sub> adsorbed on graphite electrodes and on carbon nanotubes. <i>Physical Chemistry Chemical Physics</i> , 2017, 19, 20441-20450.	2.8	45
31	Modified Electrodes with MN <sub>4</sub> Complexes: Conception and Electroanalytical Performances for the Detection of Thiols. , 2016, , 277-321.		0
32	Reaktivitätsdeskriptoren für die Aktivität von molekularen MN <sub>4</sub> -Katalysatoren zur Sauerstoffreduktion. <i>Angewandte Chemie</i> , 2016, 128, 14726-14738.	2.0	39
33	Reactivity Descriptors for the Activity of Molecular MN <sub>4</sub> Catalysts for the Oxygen Reduction Reaction. <i>Angewandte Chemie - International Edition</i> , 2016, 55, 14510-14521.	13.8	463
34	Reactivity descriptors for iron porphyrins and iron phthalocyanines as catalysts for the electrooxidation of reduced glutathione. <i>Journal of Solid State Electrochemistry</i> , 2016, 20, 3199-3208.	2.5	8
35	Multiscale Approach to the Study of the Electronic Properties of Two Thiophene Curcuminoid Molecules. <i>Chemistry - A European Journal</i> , 2016, 22, 12808-12818.	3.3	18
36	Surface on Surface. Survey of the Monolayer Gold-Graphene Interaction from Au <sub>12</sub> and PAH via Relativistic DFT Calculations. <i>Journal of Physical Chemistry C</i> , 2016, 120, 7358-7364.	3.1	12

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37	O <sub>2</sub> reduction on electrodes modified with nitrogen doped carbon nanotubes synthesized with different metal catalysts. <i>Diamond and Related Materials</i> , 2016, 64, 119-129.	3.9	22
38	Reactivity indexes for the electrocatalytic oxidation of hydrogen peroxide promoted by several ligand-substituted and unsubstituted Co phthalocyanines adsorbed on graphite. <i>Journal of Electroanalytical Chemistry</i> , 2016, 765, 22-29.	3.8	18
39	Polyaniline nanostructure electrode: morphological control by a hybrid template. <i>Journal of Solid State Electrochemistry</i> , 2016, 20, 1175-1180.	2.5	3
40	Building Nanoscale Molecular Wires Exploiting Electrocatalytic Interactions. <i>Electrochimica Acta</i> , 2015, 179, 611-617.	5.2	19
41	Stripping voltammetry microprobe (SPV): Substantial improvements of the protocol. <i>Journal of Electroanalytical Chemistry</i> , 2015, 745, 61-65.	3.8	11
42	Preparation, spectroscopic, and electrochemical characterization of metal(II) complexes with Schiff base ligands derived from chitosan: correlations of redox potentials with Hammett parameters. <i>Journal of Coordination Chemistry</i> , 2014, 67, 4114-4124.	2.2	8
43	Linear versus volcano correlations for the electrocatalytic oxidation of hydrazine on graphite electrodes modified with MN <sub>4</sub> macrocyclic complexes. <i>Electrochimica Acta</i> , 2014, 140, 314-319.	5.2	30
44	Preparation and Characterization of Electrodes Modified with Pyrrole Surfactant, Multiwalled Carbon Nanotubes and Metallophthalocyanines for the Electrochemical Detection of Thiols. <i>Electroanalysis</i> , 2014, 26, 507-512.	2.9	14
45	Optimizing the reactivity of surface confined cobalt N <sub>4</sub> -macrocyclics for the electrocatalytic oxidation of L-cysteine by tuning the Co(II)/(I) formal potential of the catalyst. <i>Electrochimica Acta</i> , 2014, 126, 37-41.	5.2	20
46	Optimization of the electrocatalytic activity of MN <sub>4</sub> -macrocyclics adsorbed on graphite electrodes for the electrochemical oxidation of L-cysteine by tuning the M (II)/(I) formal potential of the catalyst: an overview. <i>Electrochimica Acta</i> , 2014, 140, 482-488.	5.2	29
47	Towards a unified way of comparing the electrocatalytic activity MN <sub>4</sub> macrocyclic metal catalysts for O <sub>2</sub> reduction on the basis of the reversible potential of the reaction. <i>Electrochemistry Communications</i> , 2014, 41, 24-26.	4.7	62
48	In Search of the Best Iron N <sub>4</sub> -Macrocyclic Catalysts Adsorbed on Graphite Electrodes and on Multi-walled Carbon Nanotubes for the Oxidation of L-Cysteine by Adjusting the Fe(II)/(I) Formal Potential of the Complex. <i>Electrocatalysis</i> , 2014, 5, 426-437.	3.0	15
49	Fundamental Studies on the Electrocatalytic Properties of Metal Macrocyclics and Other Complexes for the Electroreduction of O <sub>2</sub> . <i>Lecture Notes in Energy</i> , 2013, , 157-212.	0.3	7
50	Electrocatalytic activity of modified gold electrodes based on self-assembled monolayers of 4-mercaptopyridine and 4-aminothiophenol on Au(111) surfaces chemically functionalized with substituted and unsubstituted iron phthalocyanines. <i>Electrochimica Acta</i> , 2013, 114, 7-13.	5.2	20
51	Carbon nanotubes and metalloporphyrins and metallophthalocyanines-based materials for electroanalysis. <i>Journal of Porphyrins and Phthalocyanines</i> , 2012, 16, 713-740.	0.8	41
52	Oxygen reduction reaction using N <sub>4</sub> -metallo macrocyclic catalysts: fundamentals on rational catalyst design. <i>Journal of Porphyrins and Phthalocyanines</i> , 2012, 16, 761-784.	0.8	132
53	Enhancement of the Catalytic Activity of Fe Phthalocyanine for the Reduction of O <sub>2</sub> Anchored to Au(111) via Conjugated Self-Assembled Monolayers of Aromatic Thiols As Compared to Cu Phthalocyanine. <i>Journal of Physical Chemistry C</i> , 2012, 116, 15329-15341.	3.1	69
54	Inverted Linear Correlation Between the Catalytic Activity of Iron Phthalocyanines and the Formal Potential of the Catalyst in the Electrooxidation of L-Cysteine. <i>Electrocatalysis</i> , 2012, 3, 153-159.	3.0	4

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55	Theoretical and Experimental Study of Bonding and Optical Properties of Self-Assembly Metallophthalocyanines Complexes on a Gold Surface. A Survey of the Substrateâ€™Surface Interaction.. Journal of Physical Chemistry C, 2011, 115, 23512-23518.	3.1	21
56	Enhanced catalytic activity of Fe phthalocyanines linked to Au(111) via conjugated self-assembled monolayers of aromatic thiols for O <sub>2</sub> reduction. Electrochemistry Communications, 2011, 13, 1182-1185.	4.7	32
57	Optimizing the Electrocatalytic Activity of Surface Confined Co Macrocyclics for the Electrooxidation of Thiocyanate at pHâ€™...4. Electroanalysis, 2011, 23, 711-718.	2.9	2
58	Metallophthalocyanine-based molecular materials as catalysts for electrochemical reactions. Coordination Chemistry Reviews, 2010, 254, 2755-2791.	18.8	502
59	Reactivity trends of surface-confined Co-tetraphenyl porphyrins and vitamin B12 for the oxidation of 2-aminoethanethiol: Comparison with Co-phthalocyanines and oxidation of other thiols. Journal of Electroanalytical Chemistry, 2010, 639, 88-94.	3.8	17
60	Does CO poison Fe-based catalysts for ORR?. Electrochemistry Communications, 2010, 12, 628-631.	4.7	119
61	Carbon Nanotubes, Phthalocyanines and Porphyrins: Attractive Hybrid Materials for Electrocatalysis and Electroanalysis. Journal of Nanoscience and Nanotechnology, 2009, 9, 2201-2214.	0.9	122
62	Tuning the Formal Potential of Metallomacrocyclics for Maximum Catalytic Activity For the Oxidation of Thiols and Hydrazine. ECS Transactions, 2009, 19, 97-112.	0.5	8
63	Volcano correlations for the reactivity of surface-confined cobalt N4-macrocyclics for the electrocatalytic oxidation of 2-mercaptoacetate. Journal of Solid State Electrochemistry, 2008, 12, 473-481.	2.5	18
64	Tuning the redox properties of Co-N4 macrocyclic complexes for the catalytic electrooxidation of glucose. Electrochimica Acta, 2008, 53, 4883-4888.	5.2	33
65	Electrocatalytic oxidation of hydrazine in alkaline media promoted by iron tetrapyrrolineporphyrin adsorbed on graphite surface. Journal of the Brazilian Chemical Society, 2008, 19, 720-726.	0.6	36
66	Tuning the redox properties of metalloporphyrin- and metallophthalocyanine-based molecular electrodes for the highest electrocatalytic activity in the oxidation of thiols. Physical Chemistry Chemical Physics, 2007, 9, 3383.	2.8	120
67	Electrocatalytic activity of cobalt phthalocyanine CoPc adsorbed on a graphite electrode for the oxidation of reduced l-glutathione (GSH) and the reduction of its disulfide (GSSG) at physiological pH. Bioelectrochemistry, 2007, 70, 147-154.	4.6	84
68	Glassy carbon electrodes modified with single walled carbon nanotubes and cobalt phthalocyanine and nickel tetrasulfonated phthalocyanine: Highly stable new hybrids with enhanced electrocatalytic performances. Electrochemistry Communications, 2007, 9, 1629-1634.	4.7	64
69	Fundamental Aspects on the Catalytic Activity of Metallomacrocyclics for the Electrochemical Reduction of O <sub>2</sub> . , 2006, , 41-82.		27
70	Theoretical Modeling of the Oxidation of Hydrazine by Iron(II) Phthalocyanine in the Gas Phase. Influence of the Metal Character. Journal of Physical Chemistry A, 2006, 110, 11870-11875.	2.5	18
71	Electrocatalysis of oxidation of 2-mercaptoethanol, l-cysteine and reduced glutathione by adsorbed and electrodeposited cobalt tetra phenoxyprole and tetra ethoxythiophene substituted phthalocyanines. Electrochimica Acta, 2006, 51, 5125-5130.	5.2	54
72	Trends in reactivity of unsubstituted and substituted cobalt-phthalocyanines for the electrocatalysis of glucose oxidation. Journal of Electroanalytical Chemistry, 2006, 589, 212-218.	3.8	62

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73	Inverted correlations between rate constants and redox potential of the catalyst for the electrooxidation of 2-aminoethanethiol mediated by surface confined substituted cobalt-phthalocyanines. <i>Journal of Electroanalytical Chemistry</i> , 2005, 580, 50-56.	3.8	30
74	Effect of film thickness on the electro-reduction of molecular oxygen on electropolymerized cobalt tetra-aminophthalocyanine films. <i>Journal of Solid State Electrochemistry</i> , 2005, 9, 21-29.	2.5	29
75	Optical Determination of Differential Coverage <sup>2+</sup> Potential Relations of Redox-Active Species Immobilized on Electrode Surfaces. <i>Analytical Chemistry</i> , 2005, 77, 6942-6946.	6.5	4
76	Solvent Effect on Density Functional Reactivity Indexes Applied to Substituted Nickel Phthalocyanines. <i>Journal of Physical Chemistry A</i> , 2004, 108, 6045-6051.	2.5	11
77	Non-linear correlations between formal potential and Hammett parameters of substituted iron phthalocyanines and catalytic activity for the electro-oxidation of hydrazine. <i>Journal of Solid State Electrochemistry</i> , 2003, 7, 626-631.	2.5	47
78	Environment effects on the oxidation of thiols: cobalt phthalocyanine as a test case. <i>Chemical Physics Letters</i> , 2003, 376, 690-697.	2.6	15
79	Theoretical study of the electron transfer reaction of hydrazine with cobalt(II) phthalocyanine and substituted cobalt(II) phthalocyanines. <i>Journal of Coordination Chemistry</i> , 2003, 56, 1269-1275.	2.2	16
80	Comparative study of electropolymerized cobalt porphyrin and phthalocyanine based films for the electrochemical activation of thiols. <i>Journal of Materials Chemistry</i> , 2002, 12, 225-232.	6.7	81
81	Theoretical study of the interaction energy profile of cobalt phthalocyanine and 2-mercaptoethanol. Effect of the graphite on the global reactivity. <i>Computational and Theoretical Chemistry</i> , 2002, 580, 193-200.	1.5	19
82	Volcano correlations between formal potential and Hammett parameters of substituted cobalt phthalocyanines and their activity for hydrazine electro-oxidation. <i>Electrochemistry Communications</i> , 2002, 4, 182-187.	4.7	60
83	Reversibility of the l-cysteine/l-cystine redox process at physiological pH on graphite electrodes modified with coenzyme B12 and vitamin B12. <i>Electrochimica Acta</i> , 2002, 48, 323-329.	5.2	49
84	Overoxidized Polypyrrole/Cobalt Tetrasulfonated Phthalocyanine Modified Ultramicro-Carbon-Fiber Electrodes for the Electrooxidation of 2-Mercaptoethanol. <i>Electroanalysis</i> , 2001, 13, 1136-1139.	2.9	20
85	Electrocatalysis of 2-mercaptoethanesulfonic acid oxidation on cobalt phthalocyanine modified electrodes. Effect of surface concentration of the catalyst. <i>Electrochimica Acta</i> , 2001, 46, 3397-3404.	5.2	35
86	Donor <sup>+</sup> acceptor intermolecular hardness on charge transfer reactions of substituted cobalt phthalocyanines. <i>Journal of Electroanalytical Chemistry</i> , 2001, 497, 55-60.	3.8	64
87	Electro-oxidation of 2-mercaptoethanol on adsorbed monomeric and electropolymerized cobalt tetra-aminophthalocyanine films. Effect of film thickness. <i>Journal of Electroanalytical Chemistry</i> , 2001, 497, 75-83.	3.8	127
88	Electroreduction of nitrite by hemin, myoglobin and hemoglobin in surfactant films. <i>Journal of Electroanalytical Chemistry</i> , 2001, 497, 106-113.	3.8	98
89	Reactivity of immobilized cobalt phthalocyanines for the electroreduction of molecular oxygen in terms of molecular hardness. <i>Journal of Electroanalytical Chemistry</i> , 2000, 489, 96-100.	3.8	72
90	Metal-centered redox chemistry of substituted cobalt phthalocyanines adsorbed on graphite and correlations with MO calculations and Hammett parameters. Electrocatalytic reduction of a disulfide. <i>Polyhedron</i> , 2000, 19, 2255-2260.	2.2	60

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91	Paradoxical effect of the redox potential of adsorbed metallophthalocyanines on their activity for the oxidation of 2-mercaptoethanol. Inner versus outer sphere electrocatalysis. <i>Electrochemistry Communications</i> , 1999, 1, 389-393.	4.7	58
92	Synthesis and electrocatalytic properties of octaalkoxycobalt phthalocyanine for the oxidation of 2-mercaptoethanol. <i>Journal of Porphyrins and Phthalocyanines</i> , 1999, 03, 355-363.	0.8	61
93	Linear versus volcano correlations between electrocatalytic activity and redox and electronic properties of metallophthalocyanines. <i>Electrochimica Acta</i> , 1998, 44, 1349-1357.	5.2	147
94	Catalytic Electrooxidation of 2-Mercaptoethanol on Perchlorinated Iron Phthalocyanine Adsorbed on a Graphite Electrode. <i>Electroanalysis</i> , 1998, 10, 571-575.	2.9	43
95	Metallophthalocyanines as catalysts in electrochemical reactions. <i>Coordination Chemistry Reviews</i> , 1992, 119, 89-136.	18.8	516
96	Catalytic electrooxidation of 2-mercaptoethanol on a graphite electrode modified with metal phthalocyanines. <i>Electrochimica Acta</i> , 1989, 34, 243-247.	5.2	52
97	Electroreduction of O <sub>2</sub> catalyzed by vitamin B12 adsorbed on a graphite electrode. <i>Journal of Electroanalytical Chemistry and Interfacial Electrochemistry</i> , 1987, 237, 145-148.	0.1	26
98	A mechanistic study of the electro-oxidation of hydrazine on phthalocyanines of VO, Cr, Mn, Ni, Cu and Zn attached to graphite electrodes. <i>Journal of Electroanalytical Chemistry and Interfacial Electrochemistry</i> , 1986, 210, 95-110.	0.1	60
99	Electrochemistry of cysteine and cystine on metal-phthalocyanines adsorbed on a graphite electrode. <i>Electrochimica Acta</i> , 1985, 30, 449-454.	5.2	80
100	Electroreduction of oxygen on mixtures of phthalocyanines co-adsorbed on a graphite electrode. <i>Journal of Electroanalytical Chemistry and Interfacial Electrochemistry</i> , 1984, 181, 295-300.	0.1	20
101	Electrocatalysis of hydrazine electrooxidation by phthalocyanines adsorbed on graphite. <i>Journal of Electroanalytical Chemistry and Interfacial Electrochemistry</i> , 1980, 109, 389-393.	0.1	51