

JosÃ© H Zagal

List of Publications by Year in descending order

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101
papers

5,208
citations

81900

39
h-index

88630

70
g-index

102
all docs

102
docs citations

102
times ranked

3793
citing authors

#	ARTICLE	IF	CITATIONS
1	Metallophthalocyanines as catalysts in electrochemical reactions. <i>Coordination Chemistry Reviews</i> , 1992, 119, 89-136.	18.8	516
2	Metallophthalocyanine-based molecular materials as catalysts for electrochemical reactions. <i>Coordination Chemistry Reviews</i> , 2010, 254, 2755-2791.	18.8	502
3	Reactivity Descriptors for the Activity of Molecular MN ₄ Catalysts for the Oxygen Reduction Reaction. <i>Angewandte Chemie - International Edition</i> , 2016, 55, 14510-14521.	13.8	463
4	Linear versus volcano correlations between electrocatalytic activity and redox and electronic properties of metallophthalocyanines. <i>Electrochimica Acta</i> , 1998, 44, 1349-1357.	5.2	147
5	Oxygen reduction reaction using N ₄ -metallomacrocyclic catalysts: fundamentals on rational catalyst design. <i>Journal of Porphyrins and Phthalocyanines</i> , 2012, 16, 761-784.	0.8	132
6	Electro-oxidation of 2-mercaptoethanol on adsorbed monomeric and electropolymerized cobalt tetra-aminophthalocyanine films. Effect of film thickness. <i>Journal of Electroanalytical Chemistry</i> , 2001, 497, 75-83.	3.8	127
7	Carbon Nanotubes, Phthalocyanines and Porphyrins: Attractive Hybrid Materials for Electrocatalysis and Electroanalysis. <i>Journal of Nanoscience and Nanotechnology</i> , 2009, 9, 2201-2214.	0.9	122
8	Tuning the redox properties of metalloporphyrin- and metallophthalocyanine-based molecular electrodes for the highest electrocatalytic activity in the oxidation of thiols. <i>Physical Chemistry Chemical Physics</i> , 2007, 9, 3383.	2.8	120
9	Does CO poison Fe-based catalysts for ORR?. <i>Electrochemistry Communications</i> , 2010, 12, 628-631.	4.7	119
10	Electroreduction of nitrite by hemin, myoglobin and hemoglobin in surfactant films. <i>Journal of Electroanalytical Chemistry</i> , 2001, 497, 106-113.	3.8	98
11	Electrocatalytic activity of cobalt phthalocyanine CoPc adsorbed on a graphite electrode for the oxidation of reduced l-glutathione (GSH) and the reduction of its disulfide (GSSG) at physiological pH. <i>Bioelectrochemistry</i> , 2007, 70, 147-154.	4.6	84
12	Comparative study of electropolymerized cobalt porphyrin and phthalocyanine based films for the electrochemical activation of thiols. <i>Journal of Materials Chemistry</i> , 2002, 12, 225-232.	6.7	81
13	Electrochemistry of cysteine and cystine on metal-phthalocyanines adsorbed on a graphite electrode. <i>Electrochimica Acta</i> , 1985, 30, 449-454.	5.2	80
14	Biomimetic reduction of O ₂ in an acid medium on iron phthalocyanines axially coordinated to pyridine anchored on carbon nanotubes. <i>Journal of Materials Chemistry A</i> , 2017, 5, 12054-12059.	10.3	76
15	Reactivity of immobilized cobalt phthalocyanines for the electroreduction of molecular oxygen in terms of molecular hardness. <i>Journal of Electroanalytical Chemistry</i> , 2000, 489, 96-100.	3.8	72
16	Enhancement of the Catalytic Activity of Fe Phthalocyanine for the Reduction of O ₂ Anchored to Au(111) via Conjugated Self-Assembled Monolayers of Aromatic Thiols As Compared to Cu Phthalocyanine. <i>Journal of Physical Chemistry C</i> , 2012, 116, 15329-15341.	3.1	69
17	Donor-acceptor intermolecular hardness on charge transfer reactions of substituted cobalt phthalocyanines. <i>Journal of Electroanalytical Chemistry</i> , 2001, 497, 55-60.	3.8	64
18	Glassy carbon electrodes modified with single walled carbon nanotubes and cobalt phthalocyanine and nickel tetrasulfonated phthalocyanine: Highly stable new hybrids with enhanced electrocatalytic performances. <i>Electrochemistry Communications</i> , 2007, 9, 1629-1634.	4.7	64

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19	Trends in reactivity of unsubstituted and substituted cobalt-phthalocyanines for the electrocatalysis of glucose oxidation. <i>Journal of Electroanalytical Chemistry</i> , 2006, 589, 212-218.	3.8	62
20	Towards a unified way of comparing the electrocatalytic activity MN4 macrocyclic metal catalysts for O ₂ reduction on the basis of the reversible potential of the reaction. <i>Electrochemistry Communications</i> , 2014, 41, 24-26.	4.7	62
21	Synthesis and electrocatalytic properties of octaalkoxycobalt phthalocyanine for the oxidation of 2-mercaptoethanol. <i>Journal of Porphyrins and Phthalocyanines</i> , 1999, 03, 355-363.	0.8	61
22	A mechanistic study of the electro-oxidation of hydrazine on phthalocyanines of VO, Cr, Mn, Ni, Cu and Zn attached to graphite electrodes. <i>Journal of Electroanalytical Chemistry and Interfacial Electrochemistry</i> , 1986, 210, 95-110.	0.1	60
23	Metal-centered redox chemistry of substituted cobalt phthalocyanines adsorbed on graphite and correlations with MO calculations and Hammett parameters. <i>Electrocatalytic reduction of a disulfide</i> . <i>Polyhedron</i> , 2000, 19, 2255-2260.	2.2	60
24	Volcano correlations between formal potential and Hammett parameters of substituted cobalt phthalocyanines and their activity for hydrazine electro-oxidation. <i>Electrochemistry Communications</i> , 2002, 4, 182-187.	4.7	60
25	Paradoxical effect of the redox potential of adsorbed metallophthalocyanines on their activity for the oxidation of 2-mercaptoethanol. Inner versus outer sphere electrocatalysis. <i>Electrochemistry Communications</i> , 1999, 1, 389-393.	4.7	58
26	Building Pyridinium Molecular Wires as Axial Ligands for Tuning the Electrocatalytic Activity of Iron Phthalocyanines for the Oxygen Reduction Reaction. <i>ACS Catalysis</i> , 2018, 8, 8406-8419.	11.2	57
27	Biomimicking vitamin B12. A Co phthalocyanine pyridine axial ligand coordinated catalyst for the oxygen reduction reaction. <i>Electrochimica Acta</i> , 2018, 265, 547-555.	5.2	56
28	Electrocatalysis of oxidation of 2-mercaptoethanol, l-cysteine and reduced glutathione by adsorbed and electrodeposited cobalt tetra phenoxypyrrole and tetra ethoxythiophene substituted phthalocyanines. <i>Electrochimica Acta</i> , 2006, 51, 5125-5130.	5.2	54
29	Catalytic electrooxidation of 2-mercaptoethanol on a graphite electrode modified with metalâ€”phthalocyanines. <i>Electrochimica Acta</i> , 1989, 34, 243-247.	5.2	52
30	In search of the most active MN4 catalyst for the oxygen reduction reaction. The case of perfluorinated Fe phthalocyanine. <i>Journal of Materials Chemistry A</i> , 2019, 7, 24776-24783.	10.3	52
31	Electrocatalysis of hydrazine electrooxidation by phthalocyanines adsorbed on graphite. <i>Journal of Electroanalytical Chemistry and Interfacial Electrochemistry</i> , 1980, 109, 389-393.	0.1	51
32	Elucidating the mechanism of the oxygen reduction reaction for pyrolyzed Fe-N-C catalysts in basic media. <i>Electrochemistry Communications</i> , 2019, 102, 78-82.	4.7	51
33	Electrocatalytic oxygen reduction reaction on iron phthalocyanine-modified carbide-derived carbon/carbon nanotube composite electrocatalysts. <i>Electrochimica Acta</i> , 2020, 334, 135575.	5.2	50
34	Reversibility of the l-cysteine/l-cystine redox process at physiological pH on graphite electrodes modified with coenzyme B12 and vitamin B12. <i>Electrochimica Acta</i> , 2002, 48, 323-329.	5.2	49
35	Non-linear correlations between formal potential and Hammett parameters of substituted iron phthalocyanines and catalytic activity for the electro-oxidation of hydrazine. <i>Journal of Solid State Electrochemistry</i> , 2003, 7, 626-631.	2.5	47
36	Comparison of the catalytic activity for O ₂ reduction of Fe and Co MN4 adsorbed on graphite electrodes and on carbon nanotubes. <i>Physical Chemistry Chemical Physics</i> , 2017, 19, 20441-20450.	2.8	45

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37	Electrocatalytic oxygen reduction on transition metal macrocyclic complexes for anion exchange membrane fuel cell application. <i>Current Opinion in Electrochemistry</i> , 2018, 9, 207-213.	4.8	44
38	Catalytic Electrooxidation of 2-Mercaptoethanol on Perchlorinated Iron Phthalocyanine Adsorbed on a Graphite Electrode. <i>Electroanalysis</i> , 1998, 10, 571-575.	2.9	43
39	Experimental reactivity descriptors of M-N-C catalysts for the oxygen reduction reaction. <i>Electrochimica Acta</i> , 2020, 332, 135340.	5.2	42
40	Carbon nanotubes and metalloporphyrins and metallophthalocyanines-based materials for electroanalysis. <i>Journal of Porphyrins and Phthalocyanines</i> , 2012, 16, 713-740.	0.8	41
41	ReaktivitÃtsdeskriptoren fÃ¼r die AktivitÃt von molekularen MN4-Katalysatoren zur Sauerstoffreduktion. <i>Angewandte Chemie</i> , 2016, 128, 14726-14738.	2.0	39
42	Oxygen Reduction Reaction at Penta-Coordinated Co Phthalocyanines. <i>Frontiers in Chemistry</i> , 2020, 8, 22.	3.6	37
43	Electrocatalytic oxidation of hydrazine in alkaline media promoted by iron tetrapyrrolineporphyrin adsorbed on graphite surface. <i>Journal of the Brazilian Chemical Society</i> , 2008, 19, 720-726.	0.6	36
44	Transition metal phthalocyanine-modified shungite-based cathode catalysts for alkaline membrane fuel cell. <i>International Journal of Hydrogen Energy</i> , 2021, 46, 4365-4377.	7.1	36
45	Mapping transition metal-MN4 macrocyclic complex catalysts performance for the critical reactivity descriptors. <i>Current Opinion in Electrochemistry</i> , 2021, 27, 100683.	4.8	36
46	Electrocatalysis of 2-mercaptoethanesulfonic acid oxidation on cobalt phthalocyanine modified electrodes. Effect of surface concentration of the catalyst. <i>Electrochimica Acta</i> , 2001, 46, 3397-3404.	5.2	35
47	Electroreduction of oxygen in alkaline solution on iron phthalocyanine modified carbide-derived carbons. <i>Electrochimica Acta</i> , 2019, 299, 999-1010.	5.2	34
48	Mapping transition metal-nitrogen-carbon catalyst performance on the critical descriptor diagram. <i>Current Opinion in Electrochemistry</i> , 2021, 27, 100687.	4.8	34
49	Tuning the redox properties of Co-N4 macrocyclic complexes for the catalytic electrooxidation of glucose. <i>Electrochimica Acta</i> , 2008, 53, 4883-4888.	5.2	33
50	Enhanced catalytic activity of Fe phthalocyanines linked to Au(111) via conjugated self-assembled monolayers of aromatic thiols for O ₂ reduction. <i>Electrochemistry Communications</i> , 2011, 13, 1182-1185.	4.7	32
51	Inverted correlations between rate constants and redox potential of the catalyst for the electrooxidation of 2-aminoethanethiol mediated by surface confined substituted cobalt-phthalocyanines. <i>Journal of Electroanalytical Chemistry</i> , 2005, 580, 50-56.	3.8	30
52	Linear versus volcano correlations for the electrocatalytic oxidation of hydrazine on graphite electrodes modified with MN4 macrocyclic complexes. <i>Electrochimica Acta</i> , 2014, 140, 314-319.	5.2	30
53	Effect of film thickness on the electro-reduction of molecular oxygen on electropolymerized cobalt tetra-aminophthalocyanine films. <i>Journal of Solid State Electrochemistry</i> , 2005, 9, 21-29.	2.5	29
54	Optimization of the electrocatalytic activity of MN4-macrocyclics adsorbed on graphite electrodes for the electrochemical oxidation of L-cysteine by tuning the M (II)/(I) formal potential of the catalyst: an overview. <i>Electrochimica Acta</i> , 2014, 140, 482-488.	5.2	29

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55	Fundamental Aspects on the Catalytic Activity of Metallomacrocyclics for the Electrochemical Reduction of O ₂ . , 2006, , 41-82.		27
56	Insights into the electronic structure of Fe penta-coordinated complexes. Spectroscopic examination and electrochemical analysis for the oxygen reduction and oxygen evolution reactions. Journal of Materials Chemistry A, 2021, 9, 23802-23816.	10.3	27
57	Enhancing the electrocatalytic activity of Fe phthalocyanines for the oxygen reduction reaction by the presence of axial ligands: Pyridine-functionalized single-walled carbon nanotubes. Electrochimica Acta, 2021, 398, 139263.	5.2	27
58	Electroreduction of O ₂ catalyzed by vitamin B12 adsorbed on a graphite electrode. Journal of Electroanalytical Chemistry and Interfacial Electrochemistry, 1987, 237, 145-148.	0.1	26
59	O ₂ reduction on electrodes modified with nitrogen doped carbon nanotubes synthesized with different metal catalysts. Diamond and Related Materials, 2016, 64, 119-129.	3.9	22
60	Probing the Fe ⁺ /Fe(ⁿ -1) ⁺ redox potential of Fe phthalocyanines and Fe porphyrins as a reactivity descriptor in the electrochemical oxidation of cysteamine. Journal of Electroanalytical Chemistry, 2018, 819, 502-510.	3.8	22
61	Penta-coordinated transition metal macrocycles as electrocatalysts for the oxygen reduction reaction. Journal of Solid State Electrochemistry, 2021, 25, 15-31.	2.5	22
62	Theoretical and Experimental Study of Bonding and Optical Properties of Self-Assembly Metallophthalocyanines Complexes on a Gold Surface. A Survey of the Substrateâ€™Surface Interaction.. Journal of Physical Chemistry C, 2011, 115, 23512-23518.	3.1	21
63	Electroreduction of oxygen on mixtures of phthalocyanines co-adsorbed on a graphite electrode. Journal of Electroanalytical Chemistry and Interfacial Electrochemistry, 1984, 181, 295-300.	0.1	20
64	Overoxidized Polypyrrole/Cobalt Tetrasulfonated Phthalocyanine Modified Ultramicro-Carbon-Fiber Electrodes for the Electrooxidation of 2-Mercaptoethanol. Electroanalysis, 2001, 13, 1136-1139.	2.9	20
65	Electrocatalytic activity of modified gold electrodes based on self-assembled monolayers of 4-mercaptopyridine and 4-aminothiophenol on Au(111) surfaces chemically functionalized with substituted and unsubstituted iron phthalocyanines. Electrochimica Acta, 2013, 114, 7-13.	5.2	20
66	Optimizing the reactivity of surface confined cobalt N4-macrocyclics for the electrocatalytic oxidation of L-cysteine by tuning the Co(II)/(I) formal potential of the catalyst. Electrochimica Acta, 2014, 126, 37-41.	5.2	20
67	Influence of cyano substituents on the electron density and catalytic activity towards the oxygen reduction reaction for iron phthalocyanine. The case for Fe(II) 2,3,9,10,16,17,23,24-octa(cyano)phthalocyanine. Electrochemistry Communications, 2020, 118, 106784.	4.7	20
68	Theoretical study of the interaction energy profile of cobalt phthalocyanine and 2-mercaptoethanol. Effect of the graphite on the global reactivity. Computational and Theoretical Chemistry, 2002, 580, 193-200.	1.5	19
69	Building Nanoscale Molecular Wires Exploiting Electrocatalytic Interactions. Electrochimica Acta, 2015, 179, 611-617.	5.2	19
70	Theoretical Modeling of the Oxidation of Hydrazine by Iron(II) Phthalocyanine in the Gas Phase. Influence of the Metal Character. Journal of Physical Chemistry A, 2006, 110, 11870-11875.	2.5	18
71	Volcano correlations for the reactivity of surface-confined cobalt N4-macrocyclics for the electrocatalytic oxidation of 2-mercaptoacetate. Journal of Solid State Electrochemistry, 2008, 12, 473-481.	2.5	18
72	Multiscale Approach to the Study of the Electronic Properties of Two Thiophene Curcuminoid Molecules. Chemistry - A European Journal, 2016, 22, 12808-12818.	3.3	18

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73	Reactivity indexes for the electrocatalytic oxidation of hydrogen peroxide promoted by several ligand-substituted and unsubstituted Co phthalocyanines adsorbed on graphite. <i>Journal of Electroanalytical Chemistry</i> , 2016, 765, 22-29.	3.8	18
74	Theoretical and Experimental Reactivity Predictors for the Electrocatalytic Activity of Copper Phenanthroline Derivatives for the Reduction of Dioxygen. <i>Journal of Physical Chemistry C</i> , 2019, 123, 19468-19478.	3.1	18
75	Reactivity trends of surface-confined Co-tetraphenyl porphyrins and vitamin B12 for the oxidation of 2-aminoethanethiol: Comparison with Co-phthalocyanines and oxidation of other thiols. <i>Journal of Electroanalytical Chemistry</i> , 2010, 639, 88-94.	3.8	17
76	Theoretical study of the electron transfer reaction of hydrazine with cobalt(II) phthalocyanine and substituted cobalt(II) phthalocyanines. <i>Journal of Coordination Chemistry</i> , 2003, 56, 1269-1275.	2.2	16
77	Environment effects on the oxidation of thiols: cobalt phthalocyanine as a test case. <i>Chemical Physics Letters</i> , 2003, 376, 690-697.	2.6	15
78	In Search of the Best Iron N4-Macrocyclic Catalysts Adsorbed on Graphite Electrodes and on Multi-walled Carbon Nanotubes for the Oxidation of L-Cysteine by Adjusting the Fe(II)/(I) Formal Potential of the Complex. <i>Electrocatalysis</i> , 2014, 5, 426-437.	3.0	15
79	Preparation and Characterization of Electrodes Modified with Pyrrole Surfactant, Multiwalled Carbon Nanotubes and Metallophthalocyanines for the Electrochemical Detection of Thiols. <i>Electroanalysis</i> , 2014, 26, 507-512.	2.9	14
80	Molecular conductance versus inductive effects of axial ligands on the electrocatalytic activity of self-assembled iron phthalocyanines: The oxygen reduction reaction. <i>Electrochimica Acta</i> , 2019, 327, 134996.	5.2	14
81	Surface on Surface. Survey of the Monolayer Gold-Graphene Interaction from Au ₁₂ and PAH via Relativistic DFT Calculations. <i>Journal of Physical Chemistry C</i> , 2016, 120, 7358-7364.	3.1	12
82	Activity volcano plots for the oxygen reduction reaction using FeN4 complexes: From reported experimental data to the electrochemical meaning. <i>Current Opinion in Electrochemistry</i> , 2022, 32, 100923.	4.8	12
83	Solvent Effect on Density Functional Reactivity Indexes Applied to Substituted Nickel Phthalocyanines. <i>Journal of Physical Chemistry A</i> , 2004, 108, 6045-6051.	2.5	11
84	Stripping voltammetry microprobe (SPV): Substantial improvements of the protocol. <i>Journal of Electroanalytical Chemistry</i> , 2015, 745, 61-65.	3.8	11
85	Oxygen Electroreduction on Zinc and Dilithium Phthalocyanine Modified Multiwalled Carbon Nanotubes in Alkaline Media. <i>Journal of the Electrochemical Society</i> , 2017, 164, H338-H344.	2.9	11
86	Reactivity descriptors for Cu bis-phenanthroline catalysts for the hydrogen peroxide reduction reaction. <i>Electrochimica Acta</i> , 2020, 357, 136881.	5.2	9
87	Tuning the Formal Potential of Metallomacrocyclics for Maximum Catalytic Activity For the Oxidation of Thiols and Hydrazine. <i>ECS Transactions</i> , 2009, 19, 97-112.	0.5	8
88	Preparation, spectroscopic, and electrochemical characterization of metal(II) complexes with Schiff base ligands derived from chitosan: correlations of redox potentials with Hammett parameters. <i>Journal of Coordination Chemistry</i> , 2014, 67, 4114-4124.	2.2	8
89	Reactivity descriptors for iron porphyrins and iron phthalocyanines as catalysts for the electrooxidation of reduced glutathione. <i>Journal of Solid State Electrochemistry</i> , 2016, 20, 3199-3208.	2.5	8
90	Fundamental Studies on the Electrocatalytic Properties of Metal Macrocyclics and Other Complexes for the Electroreduction of O ₂ . <i>Lecture Notes in Energy</i> , 2013, , 157-212.	0.3	7

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91	Fe ₃ O ₄ Templated Pyrolyzed Fe-N-C Catalysts. Understanding the role of N-Functions and Fe ₃ C on the ORR Activity and Mechanism. ChemElectroChem, 2022, 9, .	3.4	6
92	Strategies to improve the catalytic activity and stability of bioinspired Cu molecular catalysts for the ORR. Current Opinion in Electrochemistry, 2022, 35, 101035.	4.8	6
93	Mapping experimental and theoretical reactivity descriptors of Fe macrocyclic complexes deposited on graphite or on multi walled carbon nanotubes for the oxidation of thiols: Thioglycolic acid oxidation. Electrochimica Acta, 2021, 391, 138905.	5.2	5
94	Optical Determination of Differential Coverage-Potential Relations of Redox-Active Species Immobilized on Electrode Surfaces. Analytical Chemistry, 2005, 77, 6942-6946.	6.5	4
95	Inverted Linear Correlation Between the Catalytic Activity of Iron Phthalocyanines and the Formal Potential of the Catalyst in the Electrooxidation of L-Cysteine. Electrocatalysis, 2012, 3, 153-159.	3.0	4
96	Polyaniline nanostructure electrode: morphological control by a hybrid template. Journal of Solid State Electrochemistry, 2016, 20, 1175-1180.	2.5	3
97	Optimizing the Electrocatalytic Activity of Surface Confined Co Macrocyclics for the Electrooxidation of Thiocyanate at pH=4. Electroanalysis, 2011, 23, 711-718.	2.9	2
98	Tailoring electroactive surfaces by non-template molecular assembly. Towards electrooxidation of L-cysteine. Electrochimica Acta, 2017, 254, 201-213.	5.2	2
99	Effect of pH on the Electrochemical Behavior of Hydrogen Peroxide in the Presence of Pseudomonas aeruginosa. Frontiers in Bioengineering and Biotechnology, 2021, 9, 749057.	4.1	2
100	Modified Electrodes with MN4 Complexes: Conception and Electroanalytical Performances for the Detection of Thiols. , 2016, , 277-321.		0
101	Redox Potentials as Reactivity Descriptors in Electrochemistry. , 2020, , .		0