

Hakan Kayi

List of Publications by Year in descending order

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34
papers

373
citations

758635

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839053

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34
times ranked

432
citing authors

#	ARTICLE	IF	CITATIONS
1	Anticancer investigation of platinum and copper-based complexes containing quinoxaline ligands. <i>Journal of Molecular Structure</i> , 2022, 1250, 131928.	1.8	3
2	Monodisperse-porous cerium oxide microspheres as a new support with appreciable catalytic activity for a composite catalyst in benzyl alcohol oxidation. <i>New Journal of Chemistry</i> , 2021, 45, 2019-2029.	1.4	6
3	Capture of Carbonyl Sulfide by Organic Liquid Mixtures: A Systematic DFT Investigation. <i>Industrial & Engineering Chemistry Research</i> , 2021, 60, 1366-1374.	1.8	6
4	Monodisperse-porous cerium oxide microspheres carrying iridium oxide nanoparticles as a heterogeneous catalyst for water oxidation. <i>Applied Surface Science</i> , 2021, 547, 149219.	3.1	6
5	Artificial Neural Network Prediction of Sulfur Content of Diesel fuel from its Physical Properties. <i>IOP Conference Series: Materials Science and Engineering</i> , 2019, 518, 062008.	0.3	0
6	Effect of chalcogen atoms on the electronic band gaps of donor-acceptor-donor type semiconducting polymers: a systematic DFT investigation. <i>Journal of Molecular Modeling</i> , 2019, 25, 167.	0.8	2
7	Radical cleavage pathway and DNA docking studies of novel chemotherapeutic platinum agent of 5,6-di-2-ithienyl-2,3-dihydropyrazine. <i>Polyhedron</i> , 2019, 170, 25-33.	1.0	4
8	Innovative Carbon Dioxide Capturing Organic Solvent: Reaction Mechanism and Kinetics. <i>Chemical Engineering and Technology</i> , 2017, 40, 737-744.	0.9	8
9	Design of novel tellurium and selenium containing semiconducting polymers using quantum mechanical tools. <i>Computational and Theoretical Chemistry</i> , 2017, 1099, 45-54.	1.1	4
10	Experimental and theoretical investigation of the reaction between CO ₂ and carbon dioxide binding organic liquids. <i>Turkish Journal of Chemistry</i> , 2016, 40, 706-719.	0.5	4
11	Kinetics of CO ₂ capture by carbon dioxide binding organic liquids: Experimental and molecular modelling studies. <i>International Journal of Greenhouse Gas Control</i> , 2016, 49, 379-386.	2.3	18
12	Kinetics of CO ₂ Capture by Carbon Dioxide Binding Organic Liquids. <i>Green Energy and Technology</i> , 2016, , 591-603.	0.4	0
13	A theoretical investigation of 4,7-di(furan-2-yl)benzo[c][1,2,5]selenadiazole-based donor-acceptor type conjugated polymer. <i>Computational and Theoretical Chemistry</i> , 2015, 1054, 38-45.	1.1	7
14	Spectroelectrochemical Investigation of Nuclease Active Pt(II) Complexes Containing Pyrrole Oxime. <i>Electrochimica Acta</i> , 2015, 158, 333-341.	2.6	3
15	Correlations between hardness, electrostatic interactions, and thermodynamic parameters in the decomposition reactions of 3-buten-1-ol, 3-methoxy-1-propene, and ethoxyethene. <i>Structural Chemistry</i> , 2015, 26, 547-554.	1.0	3
16	Conformational behaviors of trans-2,3- and trans-2,5-dihalo-1,4-diselenanes. A complete basis set, hybrid-density functional theory study and natural bond orbital interpretations. <i>Journal of Molecular Modeling</i> , 2014, 20, 2249.	0.8	9
17	A computational study on 4,7-di(furan-2-yl)benzo[c][1,2,5]thiadiazole monomer and its oligomers. <i>Journal of Molecular Modeling</i> , 2014, 20, 2269.	0.8	9
18	Deviations from Born-Oppenheimer Theory in Structural Chemistry: Jahn-Teller, Pseudo Jahn-Teller, and Hidden Pseudo Jahn-Teller Effects in C ₃ H ₃ and C ₃ H ₃ ⁺ . <i>Journal of Physical Chemistry A</i> , 2013, 117, 8671-8679.	1.1	16

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19	Natural bond orbital, nuclear magnetic resonance analysis and hybrid-density functional theory study of π -aromaticity in Al_2F_6 , Al_2Cl_6 , Al_2Br_6 and Al_2I_6 . <i>Journal of Molecular Modeling</i> , 2013, 19, 2549-2557.	0.8	7
20	Pseudo Jahn-Teller origin of bending instability of triatomic molecules. <i>Journal of Molecular Structure</i> , 2012, 1023, 108-114.	1.8	16
21	A theoretical investigation of the relative stability of hydrated glycine and methylcarbamic acid from water clusters to interstellar ices. <i>Physical Chemistry Chemical Physics</i> , 2012, 14, 4942.	1.3	25
22	A computational study on the structures of methylamine-carbon dioxide-water clusters: evidence for the barrier free formation of the methylcarbamic acid zwitterion ($CH_3NH_2+COO^-$) in interstellar water ices. <i>Physical Chemistry Chemical Physics</i> , 2011, 13, 11083.	1.3	21
23	A theoretical investigation of the low energy conformers of the isomers glycine and methylcarbamic acid and their role in the interstellar medium. <i>Physical Chemistry Chemical Physics</i> , 2011, 13, 15774.	1.3	17
24	AM1* parameters for palladium and silver. <i>Journal of Molecular Modeling</i> , 2011, 17, 2585-2600.	0.8	7
25	AM1* parameters for cobalt and nickel. <i>Journal of Molecular Modeling</i> , 2010, 16, 29-47.	0.8	18
26	AM1* parameters for gold. <i>Journal of Molecular Modeling</i> , 2010, 16, 1029-1038.	0.8	7
27	AM1* parameters for manganese and iron. <i>Journal of Molecular Modeling</i> , 2010, 16, 1109-1126.	0.8	12
28	AM1* parameters for bromine and iodine. <i>Journal of Molecular Modeling</i> , 2009, 15, 295-308.	0.8	22
29	AM1* parameters for vanadium and chromium. <i>Journal of Molecular Modeling</i> , 2009, 15, 1253-1269.	0.8	15
30	Ni(II), Cu(II), and Zn(II) complexes of tetradentate schiff base containing two thiadiazoles units: Structural, spectroscopic, magnetic properties, and molecular modeling studies. <i>Heteroatom Chemistry</i> , 2008, 19, 700-712.	0.4	21
31	AM1* parameters for copper and zinc. <i>Journal of Molecular Modeling</i> , 2007, 13, 965-979.	0.8	35
32	Prediction of lower critical solution temperature of N-isopropylacrylamide-acrylic acid copolymer by an artificial neural network model. <i>Journal of Molecular Modeling</i> , 2005, 11, 55-60.	0.8	12
33	Stimuli-responsive properties of aminophenylboronic acid-carrying thermosensitive copolymers. <i>Polymer International</i> , 2003, 52, 649-657.	1.6	30
34	Theoretical investigation of carbon dioxide capture by aqueous boric acid solution: A termolecular reaction mechanism. <i>Journal of Boron</i> , 0, , .	0.0	0