

# Hakan Kayi

## List of Publications by Year in descending order

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34  
papers

373  
citations

758635

12  
h-index

839053

18  
g-index

34  
all docs

34  
docs citations

34  
times ranked

432  
citing authors

#	ARTICLE	IF	CITATIONS
1	AM1* parameters for copper and zinc. <i>Journal of Molecular Modeling</i> , 2007, 13, 965-979.	0.8	35
2	Stimuli-responsive properties of aminophenylboronic acid-carrying thermosensitive copolymers. <i>Polymer International</i> , 2003, 52, 649-657.	1.6	30
3	A theoretical investigation of the relative stability of hydrated glycine and methylcarbamic acid from water clusters to interstellar ices. <i>Physical Chemistry Chemical Physics</i> , 2012, 14, 4942.	1.3	25
4	AM1* parameters for bromine and iodine. <i>Journal of Molecular Modeling</i> , 2009, 15, 295-308.	0.8	22
5	Ni(II), Cu(II), and Zn(II) complexes of tetradentate schiff base containing two thiadiazoles units: Structural, spectroscopic, magnetic properties, and molecular modeling studies. <i>Heteroatom Chemistry</i> , 2008, 19, 700-712.	0.4	21
6	A computational study on the structures of methylamine-carbon dioxide-water clusters: evidence for the barrier free formation of the methylcarbamic acid zwitterion (CH <sub>3</sub> NH <sub>2</sub> +COO <sup>-</sup> ) in interstellar water ices. <i>Physical Chemistry Chemical Physics</i> , 2011, 13, 11083.	1.3	21
7	AM1* parameters for cobalt and nickel. <i>Journal of Molecular Modeling</i> , 2010, 16, 29-47.	0.8	18
8	Kinetics of CO <sub>2</sub> capture by carbon dioxide binding organic liquids: Experimental and molecular modelling studies. <i>International Journal of Greenhouse Gas Control</i> , 2016, 49, 379-386.	2.3	18
9	A theoretical investigation of the low energy conformers of the isomers glycine and methylcarbamic acid and their role in the interstellar medium. <i>Physical Chemistry Chemical Physics</i> , 2011, 13, 15774.	1.3	17
10	Pseudo Jahn-Teller origin of bending instability of triatomic molecules. <i>Journal of Molecular Structure</i> , 2012, 1023, 108-114.	1.8	16
11	Deviations from Born-Oppenheimer Theory in Structural Chemistry: Jahn-Teller, Pseudo Jahn-Teller, and Hidden Pseudo Jahn-Teller Effects in C <sub>3</sub> H <sub>3</sub> and C <sub>3</sub> H <sub>3</sub> <sup>+</sup> . <i>Journal of Physical Chemistry A</i> , 2013, 117, 8671-8679.	1.1	16
12	AM1* parameters for vanadium and chromium. <i>Journal of Molecular Modeling</i> , 2009, 15, 1253-1269.	0.8	15
13	Prediction of lower critical solution temperature of N-isopropylacrylamide-acrylic acid copolymer by an artificial neural network model. <i>Journal of Molecular Modeling</i> , 2005, 11, 55-60.	0.8	12
14	AM1* parameters for manganese and iron. <i>Journal of Molecular Modeling</i> , 2010, 16, 1109-1126.	0.8	12
15	Conformational behaviors of trans-2,3- and trans-2,5-dihalo-1,4-diselenanes. A complete basis set, hybrid-density functional theory study and natural bond orbital interpretations. <i>Journal of Molecular Modeling</i> , 2014, 20, 2249.	0.8	9
16	A computational study on 4,7-di(furan-2-yl)benzo[c][1,2,5]thiadiazole monomer and its oligomers. <i>Journal of Molecular Modeling</i> , 2014, 20, 2269.	0.8	9
17	Innovative Carbon Dioxide-Capturing Organic Solvent: Reaction Mechanism and Kinetics. <i>Chemical Engineering and Technology</i> , 2017, 40, 737-744.	0.9	8
18	AM1* parameters for gold. <i>Journal of Molecular Modeling</i> , 2010, 16, 1029-1038.	0.8	7

#	ARTICLE	IF	CITATIONS
19	AM1* parameters for palladium and silver. Journal of Molecular Modeling, 2011, 17, 2585-2600.	0.8	7
20	Natural bond orbital, nuclear magnetic resonance analysis and hybrid-density functional theory study of $\pi$ -aromaticity in Al <sub>2</sub> F <sub>6</sub> , Al <sub>2</sub> Cl <sub>6</sub> , Al <sub>2</sub> Br <sub>6</sub> and Al <sub>2</sub> I <sub>6</sub> . Journal of Molecular Modeling, 2013, 19, 2549-2557.	0.8	7
21	A theoretical investigation of 4,7-di(furan-2-yl)benzo[c][1,2,5]selenadiazole-based donor-acceptor type conjugated polymer. Computational and Theoretical Chemistry, 2015, 1054, 38-45.	1.1	7
22	Monodisperse-porous cerium oxide microspheres as a new support with appreciable catalytic activity for a composite catalyst in benzyl alcohol oxidation. New Journal of Chemistry, 2021, 45, 2019-2029.	1.4	6
23	Capture of Carbonyl Sulfide by Organic Liquid Mixtures: A Systematic DFT Investigation. Industrial & Engineering Chemistry Research, 2021, 60, 1366-1374.	1.8	6
24	Monodisperse-porous cerium oxide microspheres carrying iridium oxide nanoparticles as a heterogeneous catalyst for water oxidation. Applied Surface Science, 2021, 547, 149219.	3.1	6
25	Experimental and theoretical investigation of the reaction between CO <sub>2</sub> and carbon dioxide binding organic liquids. Turkish Journal of Chemistry, 2016, 40, 706-719.	0.5	4
26	Design of novel tellurium and selenium containing semiconducting polymers using quantum mechanical tools. Computational and Theoretical Chemistry, 2017, 1099, 45-54.	1.1	4
27	Radical cleavage pathway and DNA docking studies of novel chemotherapeutic platinum agent of 5,6-di-2-ithienyl-2,3-dihydropyrazine. Polyhedron, 2019, 170, 25-33.	1.0	4
28	Spectroelectrochemical Investigation of Nuclease Active Pt(II) Complexes Containing Pyrrole Oxime. Electrochimica Acta, 2015, 158, 333-341.	2.6	3
29	Correlations between hardness, electrostatic interactions, and thermodynamic parameters in the decomposition reactions of 3-buten-1-ol, 3-methoxy-1-propene, and ethoxyethene. Structural Chemistry, 2015, 26, 547-554.	1.0	3
30	Anticancer investigation of platinum and copper-based complexes containing quinoxaline ligands. Journal of Molecular Structure, 2022, 1250, 131928.	1.8	3
31	Effect of chalcogen atoms on the electronic band gaps of donor-acceptor-donor type semiconducting polymers: a systematic DFT investigation. Journal of Molecular Modeling, 2019, 25, 167.	0.8	2
32	Kinetics of CO <sub>2</sub> Capture by Carbon Dioxide Binding Organic Liquids. Green Energy and Technology, 2016, , 591-603.	0.4	0
33	Artificial Neural Network Prediction of Sulfur Content of Diesel fuel from its Physical Properties. IOP Conference Series: Materials Science and Engineering, 2019, 518, 062008.	0.3	0
34	Theoretical investigation of carbon dioxide capture by aqueous boric acid solution: A termolecular reaction mechanism. Journal of Boron, 0, , .	0.0	0