Lara Å tajner

List of Publications by Year in descending order

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840776 677142 23 689 11 22 h-index citations g-index papers 23 23 23 899 docs citations times ranked citing authors all docs

#	Article	IF	CITATIONS
1	Simultaneous Influence of Gradients in Natural Organic Matter and Abiotic Parameters on the Behavior of Silver Nanoparticles in the Transition Zone from Freshwater to Saltwater Environments. Nanomaterials, 2022, 12, 296.	4.1	8
2	Precipitation of Calcium Phosphates and Calcium Carbonates in the Presence of Differently Charged Liposomes. Minerals (Basel, Switzerland), 2022, 12, 208.	2.0	5
3	Exposure of microplastics to organic matter in waters enhances microplastic encapsulation into calcium carbonate. Environmental Chemistry Letters, 2022, 20, 2235-2242.	16.2	11
4	Role of Hydrodynamics, Li+ Addition and Transformation Kinetics on the Formation of Plate-Like {001} Calcite Crystals. Crystals, 2021, 11, 250.	2.2	6
5	Phytotoxicity of Silver Nanoparticles on Tobacco Plants: Evaluation of Coating Effects on Photosynthetic Performance and Chloroplast Ultrastructure. Nanomaterials, 2021, 11, 744.	4.1	19
6	Precipitation at Room Temperature as a Fast and Versatile Method for Calcium Phosphate/TiO2 Nanocomposites Synthesis. Nanomaterials, 2021, 11, 1523.	4.1	8
7	Microbial response to the presence of invasive ctenophore Mnemiopsis leidyi in the coastal waters of the Northeastern Adriatic. Estuarine, Coastal and Shelf Science, 2021, 259, 107459.	2.1	4
8	Effect of pH and Type of Stirring on the Spontaneous Precipitation of CaCO3 at Identical Initial Supersaturation, Ionic Strength and a(Ca2+)/a(CO32â^) Ratio. Crystals, 2021, 11, 1075.	2.2	11
9	The influence of the saline and artificial saliva on gamma induced radical concentration in dental bone graft materials based on calcium sulfate studied by EPR spectroscopy. Radiation Physics and Chemistry, 2020, 177, 109138.	2.8	O
10	Calcium phosphate formation on TiO2 nanomaterials of different dimensionality. Colloids and Surfaces A: Physicochemical and Engineering Aspects, 2020, 593, 124615.	4.7	10
11	Adsorption of Aspartate Derivatives to Calcite Surfaces in Aqueous Environment. Crystal Growth and Design, 2020, 20, 2853-2859.	3.0	10
12	Factors affecting calcium phosphate mineralization within bulk alginate hydrogels. Journal of Polymer Research, 2019, 26, 1.	2.4	6
13	Supramolecular Hydrogels with Properties Tunable by Calcium Ions: A Bio-Inspired Chemical System. ACS Applied Bio Materials, 2019, 2, 5819-5828.	4.6	13
14	The effect of different amino acids on spontaneous precipitation of calcium carbonate polymorphs. Journal of Crystal Growth, 2018, 486, 71-81.	1.5	42
15	Preparation and characterization of calcium oxalate dihydrate seeds suitable for crystal growth kinetic analyses. Journal of Crystal Growth, 2018, 500, 91-97.	1.5	7
16	Comparative Study of Calcium Carbonates and Calcium Phosphates Precipitation in Model Systems Mimicking the Inorganic Environment for Biomineralization. Crystal Growth and Design, 2017, 17, 1103-1117.	3.0	36
17	Complexation between lysozyme and sodium poly(styrenesulfonate): The effect of pH, reactant concentration and titration direction. Colloids and Surfaces A: Physicochemical and Engineering Aspects, 2015, 483, 171-180.	4.7	10
18	Effect of silver nanoparticles on Mediterranean sea urchin embryonal development is species specific and depends on moment of first exposure. Marine Environmental Research, 2015, 111, 50-59.	2.5	55

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#	Article	IF	CITATION
19	Influence of some polysaccharides on the production of calcium carbonate filler particles. Journal of Crystal Growth, 2008, 310, 4554-4560.	1.5	57
20	Experimental design approach to calcium carbonate precipitation in a semicontinuous process. Powder Technology, 2007, 171, 192-199.	4.2	72
21	Incorporation of Inorganic Anions in Calcite. European Journal of Inorganic Chemistry, 2004, 2004, 4579-4585.	2.0	58
22	Effect of Inorganic Anions on the Morphology and Structure of Magnesium Calcite. Chemistry - A European Journal, 2004, 10, 1647-1656.	3.3	86
23	Vaterite growth and dissolution in aqueous solution III. Kinetics of transformation. Journal of Crystal Growth, 1997, 177, 248-257.	1.5	155