

Sarah L Stoll

List of Publications by Year in descending order

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28
papers

846
citations

623734

14
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552781

26
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all docs

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docs citations

29
times ranked

1153
citing authors

#	ARTICLE	IF	CITATIONS
1	Luminescence of Ln(III) Dithiocarbamate Complexes (Ln = La, Pr, Sm, Eu, Gd, Tb, Dy). <i>Inorganic Chemistry</i> , 2008, 47, 1512-1523.	4.0	156
2	Dithiocarbamate Precursors for Rare-Earth Sulfides. <i>Chemistry of Materials</i> , 2005, 17, 3114-3121.	6.7	119
3	Magnetic Properties of Lanthanide Chalcogenide Semiconducting Nanoparticles. <i>Journal of the American Chemical Society</i> , 2006, 128, 11173-11179.	13.7	74
4	Controlled growth of HfO ₂ thin films by atomic layer deposition from cyclopentadienyl-type precursor and water. <i>Journal of Materials Chemistry</i> , 2005, 15, 2271.	6.7	64
5	Size-Dependent Magnetism of EuS Nanoparticles. <i>Chemistry of Materials</i> , 2008, 20, 3368-3376.	6.7	60
6	Monolayer and Multilayer Films of [Mn ₁₂ O ₁₂ (O ₂ CMe) ₁₆]. <i>Nano Letters</i> , 2004, 4, 399-402.	9.1	53
7	Luminescence and Nonlinear Optical Properties in Copper(I) Halide Extended Networks. <i>Inorganic Chemistry</i> , 2016, 55, 11408-11417.	4.0	40
8	Europium chalcogenide magnetic semiconductor nanostructures. <i>Coordination Chemistry Reviews</i> , 2015, 289-290, 279-288.	18.8	36
9	Gadolinium Doped Europium Sulfide. <i>Journal of the American Chemical Society</i> , 2010, 132, 13960-13962.	13.7	34
10	Magnetic Nanobeads as Potential Contrast Agents for Magnetic Resonance Imaging. <i>ACS Nano</i> , 2013, 7, 9040-9048.	14.6	26
11	Thermolysis of lanthanide dithiocarbamate complexes. <i>Journal of Solid State Chemistry</i> , 2010, 183, 52-56.	2.9	25
12	Surface attached manganese ^{II} oxo clusters as potential contrast agents. <i>Chemical Communications</i> , 2009, , 788.	4.1	24
13	Dissolution-Precipitation Synthesis and Characterization of Zinc Whitlockite with Variable Metal Content. <i>ACS Biomaterials Science and Engineering</i> , 2021, 7, 3586-3593.	5.2	22
14	Mononuclear to Polynuclear U ^{IV} Structural Units: Effects of Reaction Conditions on U ^{IV} Furoate Phase Formation. <i>Chemistry - A European Journal</i> , 2020, 26, 5872-5886.	3.3	15
15	Europium Chalcogenide Nanowires by Vapor Phase Conversions. <i>Chemistry of Materials</i> , 2014, 26, 3144-3150.	6.7	14
16	Valence and Magnetic Investigations of Alkali Metal-Doped Europium Sulfide. <i>Chemistry of Materials</i> , 2012, 24, 4390-4396.	6.7	12
17	Single-Source Precursors for Lanthanide Diselenide Nanosheets. <i>Chemistry of Materials</i> , 2019, 31, 7779-7789.	6.7	12
18	Solid-State and Nanoparticle Synthesis of Eu _x Se _{1-x} Solid Solutions. <i>Chemistry of Materials</i> , 2018, 30, 2954-2964.	6.7	10

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19	Dye-coated europium monosulfide. <i>Journal of Solid State Chemistry</i> , 2011, 184, 1324-1327.	2.9	9
20	Paramagnetic Clusters of $Mn_3(O_2CCH_3)_6(Bpy)_2$ in Polyacrylamide Nanobeads as a New Design Approach to a T_1 - T_2 Multimodal Magnetic Resonance Imaging Contrast Agent. <i>ACS Applied Materials & Interfaces</i> , 2019, 11, 18153-18164.	8.0	9
21	Miniemulsion Synthesis of Metal-Oxo Cluster Containing Copolymer Nanobeads. <i>Langmuir</i> , 2011, 27, 12575-12584.	3.5	8
22	Paramagnetic Mn_8Fe_4Co -Polystyrene Nanobeads as a Potential T_1 - T_2 Multimodal Magnetic Resonance Imaging Contrast Agent with <i>In Vivo</i> Studies. <i>ACS Applied Materials & Interfaces</i> , 2021, 13, 39042-39054.	8.0	6
23	Giant band splittings in EuS and EuSe magnetic semiconductor nanocrystals. <i>Chemical Communications</i> , 2020, 56, 5843-5846.	4.1	5
24	Synthesis of Mixed-Valent Lanthanide Sulfide Nanoparticles. <i>Angewandte Chemie - International Edition</i> , 2021, 60, 23134-23141.	13.8	5
25	The Accessibility of the Cell Wall in Scots Pine (<i>Pinus sylvestris</i> L.) Sapwood to Colloidal Fe_3O_4 Nanoparticles. <i>ACS Omega</i> , 2021, 6, 21719-21729.	3.5	4
26	Synthesis of lanthanide chalcogenide nanoparticles. , 2022, , 219-243.		3
27	Synthesis of Mixed-Valent Lanthanide Sulfide Nanoparticles. <i>Angewandte Chemie</i> , 2021, 133, 23318.	2.0	1
28	Using Redox Titrations to Probe the Role of Trivalent Impurity Ions in the Ferromagnetism of Colloidal EuS Nanocrystals. <i>Chemistry of Materials</i> , 2020, 32, 8633-8640.	6.7	0