## Nobuyuki Matubayasi

## List of Publications by Citations

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| #   | Paper                                                                                                                                                                                                             | IF                 | Citations |
|-----|-------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------|--------------------|-----------|
| 233 | Structural study of supercritical water. I. Nuclear magnetic resonance spectroscopy. <i>Journal of Chemical Physics</i> , <b>1997</b> , 107, 9133-9140                                                            | 3.9                | 157       |
| 232 | Theory of solutions in the energetic representation. I. Formulation. <i>Journal of Chemical Physics</i> , <b>2000</b> , 113, 6070-6081                                                                            | 3.9                | 133       |
| 231 | Theory of solutions in the energy representation. II. Functional for the chemical potential. <i>Journal of Chemical Physics</i> , <b>2002</b> , 117, 3605-3616                                                    | 3.9                | 128       |
| 230 | Theory of solutions in the energy representation. III. Treatment of the molecular flexibility. <i>Journal of Chemical Physics</i> , <b>2003</b> , 119, 9686-9702                                                  | 3.9                | 108       |
| 229 | Thermodynamics of the Hydration Shell. 1. Excess Energy of a Hydrophobic Solute. <i>The Journal of Physical Chemistry</i> , <b>1994</b> , 98, 10640-10649                                                         |                    | 107       |
| 228 | NMR Study of Water Structure in Super- and Subcritical Conditions. <i>Physical Review Letters</i> , <b>1997</b> , 78, 2573-2576                                                                                   | 7.4                | 103       |
| 227 | Thermodynamics of the Hydration Shell. 2. Excess Volume and Compressibility of a Hydrophobic Solute. <i>The Journal of Physical Chemistry</i> , <b>1996</b> , 100, 2681-2688                                      |                    | 98        |
| 226 | Hydrotropy: monomer-micelle equilibrium and minimum hydrotrope concentration. <i>Journal of Physical Chemistry B</i> , <b>2014</b> , 118, 10515-24                                                                | 3.4                | 83        |
| 225 | Solvent effect on pathways and mechanisms for D-fructose conversion to 5-hydroxymethyl-2-furaldehyde: in situ 13C NMR study. <i>Journal of Physical Chemistry A</i> , <b>2013</b> , 117, 2102                     | - <del>1</del> 3   | 83        |
| 224 | Protein hydration and unfoldinginsights from experimental partial specific volumes and unfolded protein models. <i>Folding &amp; Design</i> , <b>1998</b> , 3, 105-18                                             |                    | 81        |
| 223 | A quantum chemical approach to the free energy calculations in condensed systems: the QM/MM method combined with the theory of energy representation. <i>Journal of Chemical Physics</i> , <b>2004</b> , 121, 398 | 3 <del>3</del> :89 | 75        |
| 222 | Preferential solvation: dividing surface vs excess numbers. <i>Journal of Physical Chemistry B</i> , <b>2014</b> , 118, 3922-30                                                                                   | 3.4                | 74        |
| 221 | The Hofmeister series and protein-salt interactions. <i>Journal of Chemical Physics</i> , <b>2006</b> , 124, 234905                                                                                               | 3.9                | 71        |
| 220 | Kinetic and equilibrium study on formic acid decomposition in relation to the water-gas-shift reaction. <i>Journal of Physical Chemistry A</i> , <b>2006</b> , 110, 11082-90                                      | 2.8                | 70        |
| 219 | Reversible molecular dynamics for rigid bodies and hybrid Monte Carlo. <i>Journal of Chemical Physics</i> , <b>1999</b> , 110, 3291-3301                                                                          | 3.9                | 70        |
| 218 | Structural study of supercritical water. II. Computer simulations. <i>Journal of Chemical Physics</i> , <b>1999</b> , 110, 8000-8011                                                                              | 3.9                | 70        |
| 217 | Monte Carlo Study of the Effect of Pressure on Hydrophobic Association. <i>Journal of Physical Chemistry B</i> , <b>1997</b> , 101, 2054-2060                                                                     | 3.4                | 69        |

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| 216 | A new high-temperature multinuclear-magnetic-resonance probe and the self-diffusion of light and heavy water in sub- and supercritical conditions. <i>Journal of Chemical Physics</i> , <b>2005</b> , 123, 164506        | 3.9                 | 68  |  |
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| 215 | Structural study of supercritical water. III. Rotational dynamics. <i>Journal of Chemical Physics</i> , <b>2001</b> , 114, 4107-4115                                                                                     | 3.9                 | 67  |  |
| 214 | In situ kinetic study on hydrothermal transformation of D-glucose into 5-hydroxymethylfurfural through D-fructose with 13C NMR. <i>Journal of Physical Chemistry A</i> , <b>2011</b> , 115, 14013-21                     | 2.8                 | 65  |  |
| 213 | End-point calculation of solvation free energy of amino-acid analogs by molecular theories of solution. <i>Chemical Physics Letters</i> , <b>2010</b> , 496, 351-355                                                     | 2.5                 | 64  |  |
| 212 | Free-energy analysis of solubilization in micelle. <i>Journal of Chemical Physics</i> , <b>2006</b> , 124, 154908                                                                                                        | 3.9                 | 61  |  |
| 211 | Matching-Mismatching of Water Geometry and Hydrophobic Hydration. <i>Journal of the American Chemical Society</i> , <b>1994</b> , 116, 1450-1456                                                                         | 16.4                | 59  |  |
| 210 | Free-energy analysis of the molecular binding into lipid membrane with the method of energy representation. <i>Journal of Chemical Physics</i> , <b>2008</b> , 128, 195107                                               | 3.9                 | 57  |  |
| 209 | Free-energy and structural analysis of ion solvation and contact ion-pair formation of Li(+) with BF4(-) and PF6(-) in water and carbonate solvents. <i>Journal of Physical Chemistry B</i> , <b>2012</b> , 116, 6476-87 | 3.4                 | 54  |  |
| 208 | Dynamics of Hydrophobic Hydration of Benzene. <i>The Journal of Physical Chemistry</i> , <b>1996</b> , 100, 1345-134                                                                                                     | 19                  | 54  |  |
| 207 | Self-diffusion coefficients for water and organic solvents at high temperatures along the coexistence curve. <i>Journal of Chemical Physics</i> , <b>2008</b> , 129, 214501                                              | 3.9                 | 54  |  |
| 206 | Gelation of carrageenan: Effects of sugars and polyols. <i>Food Hydrocolloids</i> , <b>2016</b> , 54, 284-292                                                                                                            | 10.6                | 52  |  |
| 205 | Ermod: fast and versatile computation software for solvation free energy with approximate theory of solutions. <i>Journal of Computational Chemistry</i> , <b>2014</b> , 35, 1592-608                                    | 3.5                 | 47  |  |
| 204 | On the local and nonlocal components of solvation thermodynamics and their relation to solvation shell models. <i>Journal of Chemical Physics</i> , <b>1998</b> , 109, 4864-4872                                         | 3.9                 | 45  |  |
| 203 | NMR Spectroscopic Evidence for an Intermediate of Formic Acid in the Water©asBhift Reaction.<br>Journal of Physical Chemistry A, <b>2004</b> , 108, 7479-7482                                                            | 2.8                 | 45  |  |
| 202 | Mechanisms and Kinetics of Acetaldehyde Reaction in Supercritical Water: Noncatalytic Disproportionation, Condensation, and Decarbonylation. <i>Journal of Physical Chemistry A</i> , <b>2004</b> , 108, 116             | 535 <sup>8</sup> 11 | 643 |  |
| 201 | Gelation: the role of sugars and polyols on gelatin and agarose. <i>Journal of Physical Chemistry B</i> , <b>2014</b> , 118, 13210-6                                                                                     | 3.4                 | 44  |  |
| 200 | Controlling the equilibrium of formic acid with hydrogen and carbon dioxide using ionic liquid. <i>Journal of Physical Chemistry A</i> , <b>2010</b> , 114, 3510-5                                                       | 2.8                 | 44  |  |
| 199 | Preferential hydration of proteins: A Kirkwood-Buff approach. <i>Chemical Physics Letters</i> , <b>2006</b> , 420, 518-                                                                                                  | 523                 | 44  |  |

| 198 | The origin of cooperative solubilisation by hydrotropes. <i>Physical Chemistry Chemical Physics</i> , <b>2016</b> , 18, 25621-25628                                                                                                             | 3.6  | 42 |
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| 197 | Communication: Free-energy analysis of hydration effect on protein with explicit solvent: equilibrium fluctuation of cytochrome c. <i>Journal of Chemical Physics</i> , <b>2011</b> , 134, 041105                                               | 3.9  | 42 |
| 196 | Hydrothermal reactions of formaldehyde and formic acid: free-energy analysis of equilibrium.<br>Journal of Chemical Physics, <b>2005</b> , 122, 074509                                                                                          | 3.9  | 42 |
| 195 | Hydrothermal carbon-carbon bond formation and disproportionations of C1 aldehydes: formaldehyde and formic acid. <i>Journal of Physical Chemistry A</i> , <b>2005</b> , 109, 6610-9                                                             | 2.8  | 41 |
| 194 | Structure and dynamics of water: from ambient to supercritical. <i>Journal of Molecular Liquids</i> , <b>2001</b> , 90, 75-83                                                                                                                   | 6    | 41 |
| 193 | Evaluation of protein-protein docking model structures using all-atom molecular dynamics simulations combined with the solution theory in the energy representation. <i>Journal of Chemical Physics</i> , <b>2012</b> , 137, 215105             | 3.9  | 39 |
| 192 | A novel quantum mechanical/molecular mechanical approach to the free energy calculation for isomerization of glycine in aqueous solution. <i>Journal of Chemical Physics</i> , <b>2005</b> , 123, 124504                                        | 3.9  | 38 |
| 191 | Energetic origin of proton affinity to the air/water interface. <i>Journal of Physical Chemistry B</i> , <b>2011</b> , 115, 4745-51                                                                                                             | 3.4  | 37 |
| 190 | Self-Consistent Determination of Atomic Charges of Ionic Liquid through a Combination of Molecular Dynamics Simulation and Density Functional Theory. <i>Journal of Chemical Theory and Computation</i> , <b>2016</b> , 12, 804-11              | 6.4  | 36 |
| 189 | Slowdown of H/D exchange reaction rate and water dynamics in ionic liquids: deactivation of solitary water solvated by small anions in 1-butyl-3-methyl-imidazolium chloride. <i>Journal of Physical Chemistry A</i> , <b>2007</b> , 111, 541-3 | 2.8  | 36 |
| 188 | Pair-correlation entropy of hydrophobic hydration: decomposition into translational and orientational contributions and analysis of solute-size effects. <i>Journal of Chemical Physics</i> , <b>2006</b> , 124, 024512                         | 3.9  | 36 |
| 187 | Super- and subcritical hydration of nonpolar solutes. I. Thermodynamics of hydration. <i>Journal of Chemical Physics</i> , <b>2000</b> , 112, 8089-8109                                                                                         | 3.9  | 36 |
| 186 | Gastrophysics: Statistical thermodynamics of biomolecular denaturation and gelation from the Kirkwood-Buff theory towards the understanding of tofu. <i>Food Hydrocolloids</i> , <b>2017</b> , 62, 128-139                                      | 10.6 | 35 |
| 185 | Rotational dynamics of water and benzene controlled by anion field in ionic liquids: 1-butyl-3-methylimidazolium chloride and hexafluorophosphate. <i>Journal of Chemical Physics</i> , <b>2007</b> , 127, 104506                               | 3.9  | 34 |
| 184 | Hot Water Induces an Acid-Catalyzed Reaction in Its Undissociated Form. <i>Bulletin of the Chemical Society of Japan</i> , <b>2004</b> , 77, 691-697                                                                                            | 5.1  | 34 |
| 183 | Self-diffusion of supercritical water in extremely low-density region. <i>Journal of Chemical Physics</i> , <b>2006</b> , 125, 074307                                                                                                           | 3.9  | 33 |
| 182 | Effect of Concentration, Acid, Temperature, and Metal on Competitive Reaction Pathways for Decarbonylation and Decarboxylation of Formic Acid in Hot Water. <i>Chemistry Letters</i> , <b>2004</b> , 33, 572-573                                | 1.7  | 33 |
| 181 | Which carbon oxide is more soluble? Ab initio study on carbon monoxide and dioxide in aqueous solution. <i>Chemical Physics Letters</i> , <b>2000</b> , 323, 257-262                                                                            | 2.5  | 33 |

| 180 | Unifying hydrotropy under Gibbs phase rule. Physical Chemistry Chemical Physics, 2017, 19, 23597-23605                                                                                                                                                                                | 3.6 | 31 |
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| 179 | Computation of the free energy change associated with one-electron reduction of coenzyme immersed in water: a novel approach within the framework of the quantum mechanical/molecular mechanical method combined with the theory of energy representation. <i>Journal of Chemical</i> | 3.9 | 31 |
| 178 | Interaction-component analysis of the urea effect on amino acid analogs. <i>Physical Chemistry Chemical Physics</i> , <b>2013</b> , 15, 4377-91                                                                                                                                       | 3.6 | 30 |
| 177 | Noncatalytic Cannizzaro-type Reaction of Acetaldehyde in Supercritical Water. <i>Chemistry Letters</i> , <b>2003</b> , 32, 310-311                                                                                                                                                    | 1.7 | 28 |
| 176 | Hydration property of globular proteins: An analysis of solvation free energy by energy representation method. <i>Chemical Physics Letters</i> , <b>2010</b> , 497, 218-222                                                                                                           | 2.5 | 27 |
| 175 | Chemical equilibrium of formaldehyde and methanediol in hot water: Free-energy analysis of the solvent effect. <i>Journal of Molecular Liquids</i> , <b>2007</b> , 134, 58-63                                                                                                         | 6   | 27 |
| 174 | Interaction-component analysis of the hydration and urea effects on cytochrome c. <i>Journal of Chemical Physics</i> , <b>2016</b> , 144, 085102                                                                                                                                      | 3.9 | 26 |
| 173 | Free-energy analysis of water affinity in polymer studied by atomistic molecular simulation combined with the theory of solutions in the energy representation. <i>Journal of Chemical Physics</i> , <b>2012</b> , 137, 234903                                                        | 3.9 | 25 |
| 172 | A unified perspective on preferential solvation and adsorption based on inhomogeneous solvation theory. <i>Physica A: Statistical Mechanics and Its Applications</i> , <b>2018</b> , 492, 1988-1996                                                                                   | 3.3 | 25 |
| 171 | Origin of non-linearity in phase solubility: solubilisation by cyclodextrin beyond stoichiometric complexation. <i>Physical Chemistry Chemical Physics</i> , <b>2016</b> , 18, 15205-17                                                                                               | 3.6 | 24 |
| 170 | NMR Study on the Reorientational Relaxation in Supercritical Alcohols. <i>Journal of Physical Chemistry A</i> , <b>2004</b> , 108, 1319-1324                                                                                                                                          | 2.8 | 23 |
| 169 | Enthalpy and Entropy Decomposition of Free-Energy Changes for Side-Chain Conformations of Aspartic Acid and Asparagine in Acidic, Neutral, and Basic Aqueous Solutions. <i>Journal of Physical Chemistry B</i> , <b>2002</b> , 106, 12336-12343                                       | 3.4 | 23 |
| 168 | Kinetic study on disproportionations of C1 aldehydes in supercritical water: methanol from formaldehyde and formic acid. <i>Journal of Physical Chemistry A</i> , <b>2007</b> , 111, 2697-705                                                                                         | 2.8 | 22 |
| 167 | Solvation shell dynamics studied by molecular dynamics simulation in relation to the translational and rotational dynamics of supercritical water and benzene. <i>Journal of Chemical Physics</i> , <b>2007</b> , 127, 1745                                                           | 569 | 22 |
| 166 | Chloride Ion Hydration and Diffusion in Supercritical Water Using a Polarizable Water Model. <i>Journal of Physical Chemistry B</i> , <b>2002</b> , 106, 3979-3986                                                                                                                    | 3.4 | 22 |
| 165 | Investigation of the dominant hydration structures among the ionic species in aqueous solution: novel quantum mechanics/molecular mechanics simulations combined with the theory of energy representation. <i>Journal of Chemical Physics</i> , <b>2008</b> , 128, 064507             | 3.9 | 21 |
| 164 | Free-energy analysis of protein solvation with all-atom molecular dynamics simulation combined with a theory of solutions. <i>Current Opinion in Structural Biology</i> , <b>2017</b> , 43, 45-54                                                                                     | 8.1 | 20 |
| 163 | Structural characteristics of yeast F1-ATPase before and after 16-degree rotation of the Bubunit: theoretical analysis focused on the water-entropy effect. <i>Journal of Chemical Physics</i> , <b>2012</b> , 137, 035102                                                            | 3.9 | 20 |

| 162 | Limited slowdown of endocrine-disruptor diffusion in confined fluid lipid membranes. <i>Physical Review Letters</i> , <b>2004</b> , 93, 248101                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                           | 7.4  | 20 |
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| 161 | Noncatalytic Disproportionation and Decarbonylation Reactions of Benzaldehyde in Supercritical Water. <i>Chemistry Letters</i> , <b>2004</b> , 33, 622-623                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                               | 1.7  | 20 |
| 160 | Energetics of nonpolar and polar compounds in cationic, anionic, and nonionic micelles studied by all-atom molecular dynamics simulation combined with a theory of solutions. <i>Physical Chemistry Chemical Physics</i> , <b>2016</b> , 18, 13223-31                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                    | 3.6  | 20 |
| 159 | Spatial-Decomposition Analysis of Energetics of Ionic Hydration. <i>Journal of Physical Chemistry B</i> , <b>2016</b> , 120, 1813-21                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                     | 3.4  | 19 |
| 158 | Molecular dynamics study of fast dielectric relaxation of water around a molecular-sized ion. <i>Journal of Chemical Physics</i> , <b>2012</b> , 137, 224502                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                             | 3.9  | 19 |
| 157 | Energy-Representation Theory of Solutions: Its Formulation and Application to Soft, Molecular Aggregates. <i>Bulletin of the Chemical Society of Japan</i> , <b>2019</b> , 92, 1910-1927                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                 | 5.1  | 19 |
| 156 | The Role of Interfacial Water in Protein-Ligand Binding: Insights from the Indirect Solvent Mediated Potential of Mean Force. <i>Journal of Chemical Theory and Computation</i> , <b>2018</b> , 14, 512-526                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                              | 6.4  | 18 |
| 155 | Correlation analysis for heat denaturation of Trp-cage miniprotein with explicit solvent. <i>Protein Science</i> , <b>2016</b> , 25, 56-66                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                               | 6.3  | 18 |
| 154 | The mechanism of salt effects on starch gelatinization from a statistical thermodynamic perspective. <i>Food Hydrocolloids</i> , <b>2019</b> , 87, 593-601                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                               | 10.6 | 18 |
| 153 | Scaled Polynomial Expression for Self-Diffusion Coefficients for Water, Benzene, and Cyclohexane over a Wide Range of Temperatures and Densities. <i>Journal of Chemical &amp; Designation of Chemical &amp; Desig</i> | 2.8  | 18 |
| 152 | Communication: exploring the reorientation of benzene in an ionic liquid via molecular dynamics: effect of temperature and solvent effective charge on the slow dynamics. <i>Journal of Chemical Physics</i> , <b>2011</b> , 134, 191101                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                 | 3.9  | 18 |
| 151 | Nuclear magnetic resonance study on rotational dynamics of water and benzene in a series of ionic liquids: anion and cation effects. <i>Journal of Chemical Physics</i> , <b>2012</b> , 137, 194503                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                      | 3.9  | 18 |
| 150 | Mechanisms and kinetics of noncatalytic ether reaction in supercritical water. 2. Proton-transferred fragmentation of dimethyl ether to formaldehyde in competition with hydrolysis. <i>Journal of Physical Chemistry A</i> , <b>2005</b> , 109, 3558-64                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                 | 2.8  | 18 |
| 149 | Molecular dynamics simulations of yeast F1-ATPase before and after 16½ rotation of the 🗟 ubunit.<br>Journal of Physical Chemistry B, <b>2013</b> , 117, 3298-307                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                         | 3.4  | 17 |
| 148 | A theoretical study of the two binding modes between lysozyme and tri-NAG with an explicit solvent model based on the fragment molecular orbital method. <i>Physical Chemistry Chemical Physics</i> , <b>2013</b> , 15, 3646-54                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                          | 3.6  | 17 |
| 147 | Insights into the origins of configurational stability of axially chiral biaryl amines with an intramolecular N-H-N hydrogen bond. <i>Journal of Organic Chemistry</i> , <b>2010</b> , 75, 5031-6                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                        | 4.2  | 17 |
| 146 | Acid-Catalyzed Hydrothermal Formation of Carbon©arbon Bond in Glycolic Acid from a Series of Formaldehyde Producers. <i>Chemistry Letters</i> , <b>2004</b> , 33, 624-625                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                | 1.7  | 17 |
| 145 | Jump in the Rotational Mobility of Benzene Induced by the Clathrate Hydrate Formation. <i>The Journal of Physical Chemistry</i> , <b>1995</b> , 99, 1377-1379                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                            |      | 17 |

| 144 | Hydrotropy and scattering: pre-ouzo as an extended near-spinodal region. <i>Physical Chemistry Chemical Physics</i> , <b>2017</b> , 19, 26734-26742                                                                                                                    | 3.6            | 16 |  |
|-----|------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------|----------------|----|--|
| 143 | Spatial-decomposition analysis of electrical conductivity in concentrated electrolyte solution. <i>Journal of Chemical Physics</i> , <b>2014</b> , 141, 044126                                                                                                         | 3.9            | 16 |  |
| 142 | Simple and exact approach to the electronic polarization effect on the solvation free energy: formulation for quantum-mechanical/molecular-mechanical system and its applications to aqueous solutions. <i>Journal of Chemical Physics</i> , <b>2012</b> , 136, 214503 | 3.9            | 16 |  |
| 141 | Association and Dissociation of Nonpolar Solutes in Super- and Subcritical Water. <i>Journal of Physical Chemistry B</i> , <b>2000</b> , 104, 10352-10358                                                                                                              | 3.4            | 16 |  |
| 140 | Free-Energy Analysis of Peptide Binding in Lipid Membrane Using All-Atom Molecular Dynamics Simulation Combined with Theory of Solutions. <i>Journal of Physical Chemistry B</i> , <b>2018</b> , 122, 3219-3229                                                        | 3.4            | 15 |  |
| 139 | Effective charges of ionic liquid determined self-consistently through combination of molecular dynamics simulation and density-functional theory. <i>Journal of Computational Chemistry</i> , <b>2017</b> , 38, 2559-                                                 | 2569           | 15 |  |
| 138 | Computation of the reduction free energy of coenzyme in aqueous solution by the QM/MM-ER method. <i>Chemical Physics Letters</i> , <b>2008</b> , 456, 176-180                                                                                                          | 2.5            | 15 |  |
| 137 | Noncatalytic kinetic study on site-selective H/D exchange reaction of phenol in sub- and supercritical water. <i>Journal of Chemical Physics</i> , <b>2004</b> , 121, 960-9                                                                                            | 3.9            | 15 |  |
| 136 | Tumbling and spinning diffusions of acetonitrile in water and organic solvents. <i>Journal of Chemical Physics</i> , <b>2000</b> , 112, 1462-1473                                                                                                                      | 3.9            | 15 |  |
| 135 | Boson peak, elasticity, and glass transition temperature in polymer glasses: Effects of the rigidity of chain bending. <i>Scientific Reports</i> , <b>2019</b> , 9, 19514                                                                                              | 4.9            | 15 |  |
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| 8  | New Method for Precise Measurement of Density of Supercritical Fluid over a Wide Range of Pressure. <i>Review of High Pressure Science and Technology/Koatsuryoku No Kagaku To Gijutsu</i> , <b>2006</b> , 16, 379-380                                                                                             | О   |   |
| 7  | Free-Energy Analysis of Solutions in the Method of Energy Representation. <i>Seibutsu Butsuri</i> , <b>2006</b> , 46, 228-231                                                                                                                                                                                      | Ο   |   |
| 6  | Development of a Massively Parallel QM/MM Approach Combined with a Theory of Solutions. <i>Challenges and Advances in Computational Chemistry and Physics</i> , <b>2015</b> , 153-196                                                                                                                              | 0.7 |   |
| 5  | Energy Representation Approach <b>2017</b> ,                                                                                                                                                                                                                                                                       |     |   |
| 4  | 3P164 All-atom hydration analysis of the Bubunit in F1-ATPase(11. Molecular motor,Poster). <i>Seibutsu Butsuri</i> , <b>2013</b> , 53, S239                                                                                                                                                                        | О   |   |
| 3  | A Molecular Thermodynamics Approach to Capture Non-specific FlavourMacromolecule Interactions <b>2019</b> , 522-527                                                                                                                                                                                                |     |   |
| 2  | All-Atom Analysis of Free Energy of Protein Solvation Through Molecular Simulation and Solution Theory <b>2018</b> , 141-155                                                                                                                                                                                       |     |   |
| 1  | Crystallization of Polyethylene Brushes and Its Effect on Interactions with Water. <i>Macromolecules</i> , <b>2021</b> , 54, 8303-8313                                                                                                                                                                             | 5.5 |   |