## **Hee-Seung Lee**

List of Publications by Year in descending order

Source: https://exaly.com/author-pdf/10158325/publications.pdf Version: 2024-02-01



#	Article	IF	CITATIONS
1	A study of the complexes of Hg(II) with polypyridyl ligands by Fluorescence, absorbance Spectroscopy, and DFT calculations. The effect of ligand preorganization and relativistic effects on complex stability. Inorganica Chimica Acta, 2022, 530, 120670.	2.4	3
2	Evidence for Participation of 4f and 5d Orbitals in Lanthanide Metal–Ligand Bonding and That Y(III) Has Less of This Complex-Stabilizing Ability. A Thermodynamic, Spectroscopic, and DFT Study of Their Complexation by the Nitrogen Donor Ligand TPEN. Inorganic Chemistry, 2022, 61, 4627-4638.	4.0	11
3	Strategies for a fluorescent sensor with receptor and fluorophore designed for the recognition of heavy metal ions. Inorganica Chimica Acta, 2020, 499, 119181.	2.4	14
4	Fluorescence and Metal-Binding Properties of the Highly Preorganized Tetradentate Ligand 2,2′-Bi-1,10-phenanthroline and Its Remarkable Affinity for Cadmium(II). Inorganic Chemistry, 2020, 59, 13117-13127.	4.0	13
5	Exciplex formation as an approach to selective Copper(II) fluorescent sensors. Inorganica Chimica Acta, 2020, 506, 119544.	2.4	6
6	Mono-N-benzyl cyclen: A highly selective, multi-functional "turn-on―fluorescence sensor for Pb2+, Hg2+ and Zn2+. Polyhedron, 2020, 179, 114366.	2.2	6
7	Indole-based fluorescence sensors for both cations and anions. Inorganica Chimica Acta, 2018, 482, 478-490.	2.4	8
8	Exciplex Formation and Aggregation Induced Emission in Diâ€( <i>N</i> â€benzyl)cyclen and Its Complexes – Selective Fluorescence with Lead(II), and as the Cadmium(II) Complex, with the Chloride Ion. European Journal of Inorganic Chemistry, 2018, 2018, 3736-3747.	2.0	9
9	Effects of anion coordination on the fluorescence of a photo-induced electron transfer (PET) sensor complexed with metal ions. Polyhedron, 2017, 130, 47-57.	2.2	17
10	Coarse-grained simulations of hemolytic peptide δ-lysin interacting with a POPC bilayer. Biochimica Et Biophysica Acta - Biomembranes, 2016, 1858, 3182-3194.	2.6	5
11	Spectroscopic, structural, and thermodynamic aspects of the stereochemically active lone pair on lead(II): Structure of the lead(II) dota complex. Polyhedron, 2015, 91, 120-127.	2.2	31
12	Peptides with the Same Composition, Hydrophobicity, and Hydrophobic Moment Bind to Phospholipid Bilayers with Different Affinities. Journal of Physical Chemistry B, 2014, 118, 12462-12470.	2.6	24
13	The Effect of ï€ Contacts between Metal Ions and Fluorophores on the Fluorescence of PET Sensors: Implications for Sensor Design for Cations and Anions. Inorganic Chemistry, 2014, 53, 9014-9026.	4.0	38
14	Molecular Dynamics Simulations of Hemolytic Peptide δLysin Interacting with a POPC Lipid Bilayer. Bulletin of the Korean Chemical Society, 2014, 35, 783-792.	1.9	4
15	Mechanism of chelation enhanced fluorescence in complexes of cadmium(ii), and a possible new type of anion sensor. Chemical Communications, 2013, 49, 9749.	4.1	45
16	Role of Fluorophore–Metal Interaction in Photoinduced Electron Transfer (PET) Sensors: Time-Dependent Density Functional Theory (TDDFT) Study. Journal of Physical Chemistry A, 2013, 117, 13345-13355.	2.5	59
17	Selectivity of the Highly Preorganized Tetradentate Ligand 2,9-Di(pyrid-2-yl)-1,10-phenanthroline for Metal Ions in Aqueous Solution, Including Lanthanide(III) Ions and the Uranyl(VI) Cation. Inorganic Chemistry, 2013, 52, 15-27.	4.0	33
18	Mechanism of "Turn-on―Fluorescent Sensors for Mercury(II) in Solution and Its Implications for Ligand Design. Inorganic Chemistry, 2012, 51, 10904-10915.	4.0	113

HEE-SEUNG LEE

#	Article	IF	CITATIONS
19	First principles studies of the electronic properties and catalytic activity of single-walled carbon nanotube doped with Pt clusters and chains. Chemical Physics, 2012, 393, 96-106.	1.9	13
20	Molecular Dynamics Studies of Transportan 10 (Tp10) Interacting with a POPC Lipid Bilayer. Journal of Physical Chemistry B, 2011, 115, 1188-1198.	2.6	52
21	Unusual Metal Ion Selectivities of the Highly Preorganized Tetradentrate Ligand 1,10-Phenanthroline-2,9-dicarboxamide: A Thermodynamic and Fluorescence Study. Inorganic Chemistry, 2011, 50, 8348-8355.	4.0	46